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**A COMPREHENSIVE PROGRAM FOR
THE COMPILATION AND ANALYSIS OF
THERMAL RADIATIVE PROPERTIES DATA**

by D. P. DeWitt, M. C. Muinzer, and R. S. Hernicz

Prepared by
PURDUE UNIVERSITY
West Lafayette, Ind.
for

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for

NATIONAL AERONAUTICS AND SPACE ADMINISTRATION

FOREWORD

This report was prepared by the Thermophysical Properties Research Center (TPRC), Purdue University, West Lafayette, Indiana, under NASA Contract No. NSR-15-005-037, "Compilation and Analysis of Thermal Radiative Properties Data". The work was administered under the direction of the Office of Advanced Research and Technology, NASA, Washington, D. C., with Mr. Conrad Mook acting as project monitor.

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ABSTRACT

A frequent obstacle in technical developments is the paucity of knowledge on the properties of materials. While there exists an ever increasing volume of literature on the thermal radiative properties, engineering designers are using only a small fraction of what is already available either because it is in a form not directly useful to them or because its existence is not generally known to them, and such information remains buried in the world's literature.

The program described in this report has the objective to identify, collect, extract and analyze data on the thermal radiative properties for general dissemination. Effort is being concentrated on the technological materials of special interest to aerospace requirements under environmental conditions likely to occur in space application. The results of the program will be disseminated through the TPRC SERIES ON THERMOPHYSICAL PROPERTIES OF MATTER presented in three parts: metallic elements and their alloys, nonmetallic solids, and coatings. Data are presented in tabular and graphical forms with accompanying tables giving specifications for the test specimens. For each material four prime properties are identified - emittance, reflectance, absorptance and transmittance - and further grouped according to geometric conditions (hemispherical, normal and angular) and wavelength conditions (spectral, total, solar, and integrated). When sufficient evidence exists, the data has been analyzed and figures of "recommended" values for specific surface and/or environmental conditions are presented. The first part of the program is nearing completion (metallic elements and alloys) and the second and third parts are now well under way.

INTRODUCTION

The phenomenal growth of science and technology in the last twenty years has brought about a universal appreciation of the fact that present limitations in many technical developments are often a direct result of the paucity of knowledge on the properties of materials. Engineering developments in the years ahead will be closely linked to the research being performed today to contribute to a better understanding of the thermal behavior of materials.

The amount of activity in property determinations, especially thermal radiative properties in the more recent years, attests to the seriousness of the situation with the result that the volume of research literature has increased many fold. Despite the widespread efforts, it can safely be said that the present level of research still falls short of existing needs and anticipated demands. The really disturbing feature is that engineering groups across the nation are using only a small fraction of the information already available, either because it is in a form not directly useful to them or because its existence is not generally known to them and such information remains buried in the world's literature.

There are those who would argue that having full access to the world's literature in a convenient form would not be the total or final solution. Because of the elusive nature of radiative properties - the large influence of surface conditions arising from methods of preparation, thermal history, and environmental conditions - there is little assurance that the data, once located, can be considered reliable or applicable to the problem at hand. This point of view has given rise to a rash of measurement programs necessary for generating data required for specific applications. For the most part, the basic problem - that of gaining insight on how to characterize materials, thereby prescribing the suitability for data to the various applications - has been largely ignored.

Since 1960, the TPRC Retrieval Guide [1]* has provided a key to the world's literature permitting rapid identification of research papers on radiative properties data. While this is the first and most necessary step, it is not the most convenient form for design engineers. Since 1962, using the Retrieval Guide as the starting point, pertinent literature on radiative properties has been

*The numbers in brackets refer to bibliographic citations listed in the section References.

examined and data extracted for the purpose of generating a compendia published as a part of the TPRC Data Book [2]. The program now in progress, and being described in this report, is an extension and enlargement of the modest effort started many years ago.

The current comprehensive program has the objective to concentrate on the technological materials of special interest to aerospace requirements under environmental conditions likely to occur in space applications. The result will be an extensive handbook for design use, containing original research data literature and also "recommended" values for surface and/or environmental conditions that can be well characterized.

This program will bring the world's literature under full and current organizational control. Such a tool, besides giving support to the designers, provides the starting point for further research as the topography of the world's knowledge will make evident the paucities and conflicts in data, as well as provide input for characterization studies using a great bulk of available data. The experience at TPRC using this approach for other properties has been rewarding and significant contributions, particularly in the area of thermal conduction in solids [3], have been made.

SCOPE OF THE PROGRAM

The primary objective of this program is to identify, collect, extract, and analyze data on thermal radiative properties for dissemination to and use by engineering design groups. This task, formidable in both its scope and magnitude, presents difficulties in devising an organizational form suitable and convenient for reference by the many interested users of such data.

The materials of interest include all metals, ceramics (excluding glasses), cermets, and coatings of all types especially those particularly suited for thermal control. The temperature range covers from near absolute zero to the material's melting point as only the solid state is being considered. The wavelength range covers from 500 Å to 1000 μ which encompasses the thermal portion of the spectrum, and special attention is given to solar spectrum conditions.

The thermal radiative properties being presented include the prime properties: emittance, reflectance, absorptance, and transmittance. Additionally, the various sub-properties of these prime ones, denoting geometric and wavelength conditions, are further categorized for efficient retrieval. In as much as the nomenclature for these properties is not universally accepted, it has been necessary to develop a consistent set of terms to unambiguously represent the various sub-properties. For the most part, the nomenclature, fully described in a later section, approximates common usage and lends itself well to the compact and systematic organization required of such a comprehensive work.

The following section of this report deals with the more significant problems in organization of the TPRC SERIES which will be the medium for communication of the information generated by the program.

ORGANIZATION OF THE SERIES

The data on thermal radiative properties will be presented in three volumes of the TPRC SERIES ON THERMOPHYSICAL PROPERTIES OF MATTER:

Volume 7 - Thermal Radiative Properties of Metallic Elements and Alloys

Volume 8 - Thermal Radiative Properties of Nonmetallic Solids

Volume 9 - Thermal Radiative Properties of Coatings

Each volume will have, in addition to a data section, a text portion concerning the theory, estimation and measurement of the thermal radiative properties of the materials covered in that volume. The purpose of this text is to provide tools for the use and understanding of the data section, and should be a unique contribution in that it contains, in some detail, the experience of the TPRC staff in the evaluation and estimation of property values. See Appendix A for the Contents of this text.

Following the text, there will be a section on data presentation and other related information. In this section the classification - properties and materials - systems are defined, along with the symbols and abbreviations used in the figures and tables. Since these systems are extensive, the user of the book will find it

desirable, if not necessary, to become familiar with this section for maximum efficiency of data retrieval.

PROPERTY CLASSIFICATION

The primary radiative properties - emittance, reflectance, absorptance, and transmittance - can all be further specified according to geometric and wavelength conditions. In this SERIES the geometric conditions are denoted by the terms angular, normal, and hemispherical. The wavelength conditions are denoted by the terms spectral, total, integrated and solar.

The definition of the terms and their representation by symbols is presented in Appendix B. This section also discusses the classification scheme for grouping related sub-properties for simplified retrieval by the user. What may at first glance appear to be a cumbersome scheme of unnecessary detail is really quite logical to serve the needs of both the casual and experienced user.

PRESENTATION OF DATA

Elements, Alloys and Compounds

The classification scheme for materials in Volumes 7 and 8 is based upon bulk composition rather than surface chemical composition which varies due to weathering, oxidation, etc. This classification scheme is shown in Table 1 which also illustrates the logic for dividing the materials into two volumes, metallic elements and alloys (Volume 7) and nonmetallic solids (Volume 8).

For each material, sub-property data are separately presented in graphical and tabular form, accompanied by a table presenting details of the test conditions and sample preparation.

The format for the presentation of the thermal radiative properties is designed specifically to supply the reader with the aspects of the properties in a comprehensive yet concise form. Each presentation consists of four sections^{*}; Original Data Plot, Analyzed Data Graph, Specification Tables, and Data Tables, respectively.

^{*}In certain cases, where there exists only a small amount of data, the Original Data Plot and/or the Analyzed Data Graph may be omitted.

TABLE 1. CLASSIFICATION OF MATERIALS

		$X_1, \%$	$X_1 + X_2, \%$	$X_2, \%$	$X_3, \%$	
1. Metallic Elements and Alloys						
Metallic Elements		> 99.5	-----	< 0.2	< 0.2	
Alloys	Binary Alloys		-----	≥ 99.5	≥ 0.2	≤ 0.2
	Multiple Alloys		-----	≥ 99.5	> 0.2	> 0.2
			-----	< 99.5	≥ 0.2	≤ 0.2
			-----	< 99.5	> 0.2	> 0.2
			≤ 99.5	-----	< 0.2	< 0.2
2. Nonmetallic Solids*						
Nonmetallic Elements (or Single Compounds)		≥ 95.0		≤ 2.0		
Mixtures		< 95.0		≤ 2.0		
		≥ 95.0		> 2.0		
		< 95.0		> 2.0		

Nomenclature:

X_1 = Major Constituent
 X_2 = Second Highest Constituent
 X_3 = Third Highest Constituent
 $\%$ = Weight Percent

*The compositions of non-metals cannot be determined as accurately as those of metals. Therefore, those percentages indicated only serve as approximate limits.

The Original Data Plot is a graphical representation which presents most of the tabulated data. In overcrowded plots, some of the data which are repetitive in nature are omitted.

The Analyzed Data Graph presents a new and powerful approach to increasing the effectiveness of literature data. It is an evaluative review identifying and "recommending" reliable and/or typical data for various surface and/or environmental conditions. The study considers the interrelationships between the sub-properties to give a consistent set of data. Where the data are well characterized and/or highly reliable, it is represented by a solid curve; where there exists some speculation, the data are represented by dashed lines or a shaded band. Following presentation of selected figures from Volume 7, a brief section discusses this approach of data analysis.

The Specification Table gives the most important information: the curve number correlating the information on the Specification Table with that of the figures and Data Table, the reference number corresponding to the number given in the listed references, the year of the publication from which the data were extracted, independent variable range, parameter(s), geometry (θ , θ' , ω , ω') and the error (%) reported by the author.

The Composition (weight percent), Specification, and Remarks section of the Specification Table provides the available information about the specimen and test conditions. The presentation is standardized in this order:

1. trade name
2. composition (weight percent)
3. film or foil thickness
4. specimen preparation processes
5. surface condition (roughness, etc.)
6. environment
7. type of original presentation of the data (smooth curve, etc.)
8. reference standard
9. other pertinent information
10. author's designation

Following the Specification Table is the Data Table, a tabular presentation of the property values shown on the Figure and described in the Specification Tables.

Coatings

Considerable effort has been given during the past year to the organization of information on coatings. The term "coatings" is a general one and elusive to define. In the context of the SERIES, a coating is a system consisting of a layer (or layers) of any substance(s) upon a substrate. Of interest are all types of coatings used for many applications - covering protection, finishing, thermal control, etc.

Consideration was given to grouping by application, methods of preparation, and durability. Discussions were held with numerous national experts concerning potential classification systems based on different aspects of coatings. Considering suggestions in light of the purpose for which the SERIES was intended, four major groups of coatings have been delineated which are defined in Table 2.

It should be stressed that this system is still being studied for areas of improvement and that as more data are studied in detail, further analysis of the system structure will be possible.

In processing papers on coatings, it is necessary to include more information in the Specification Tables than is required for the non-coatings. In addition, the following parameters if available are given:

1. thickness of coating
2. substrate
3. condition of substrate
4. application technique
5. environmental effects
6. catalyst (paints)
7. pigment-vehicle ratio (paints)
8. properties of coating (viscosity, porosity, etc.)
9. other pertinent information given by author

Special emphasis will be given to environmental conditions before and during measurements.

TABLE 2. CLASSIFICATION OF COATINGS

Conversion - Diffusion Coatings

A layer of a compound, or mixture of compounds formed by the chemical reaction of the substrate with another material. Classified alphabetically by substrate. Examples:

- a. Anodized Aluminum
- b. Durak B
- c. Oxidized Inconel 702

Contact Coatings

A layer, or layers, of a substance coated on a substrate without a chemical reaction occurring between the coating material and the substrate. Classified alphabetically by coating material itself. Subdivided into the following three types, with examples:

1. Inorganic
 - a. Alclad Aluminum
 - b. Flame-sprayed Al_2O_3
 - c. Evaporated Gold Film
2. Organic
 - a. Teflon
 - b. Vinyl
3. Special Purpose
 - a. Anti-reflection Coatings
 - b. Second Surface Mirrors

Pigmented Coatings (Vitreous enamels and paints)

A mixture of pigment and vehicle applied to a substrate. Classified alphabetically by pigment. Subdivided into the following three types, with examples:

1. Inorganic Pigment
 - a. Paint (Sb_2O_3 + nitrocellulose)
 - b. Paint (TiO_2 + epoxy resin)
 - c. Paint (ZnO + silicone)
2. Organic Pigment
 - a. Paint (Vinyl-phenolic)
3. Miscellaneous Paints
 - a. Dynalac H-U
 - b. Korotherm HT
 - c. Colors

Uncharacterized Coatings

Classified alphabetically by commercial name.

MATERIAL INDEX AND REFERENCE LISTINGS

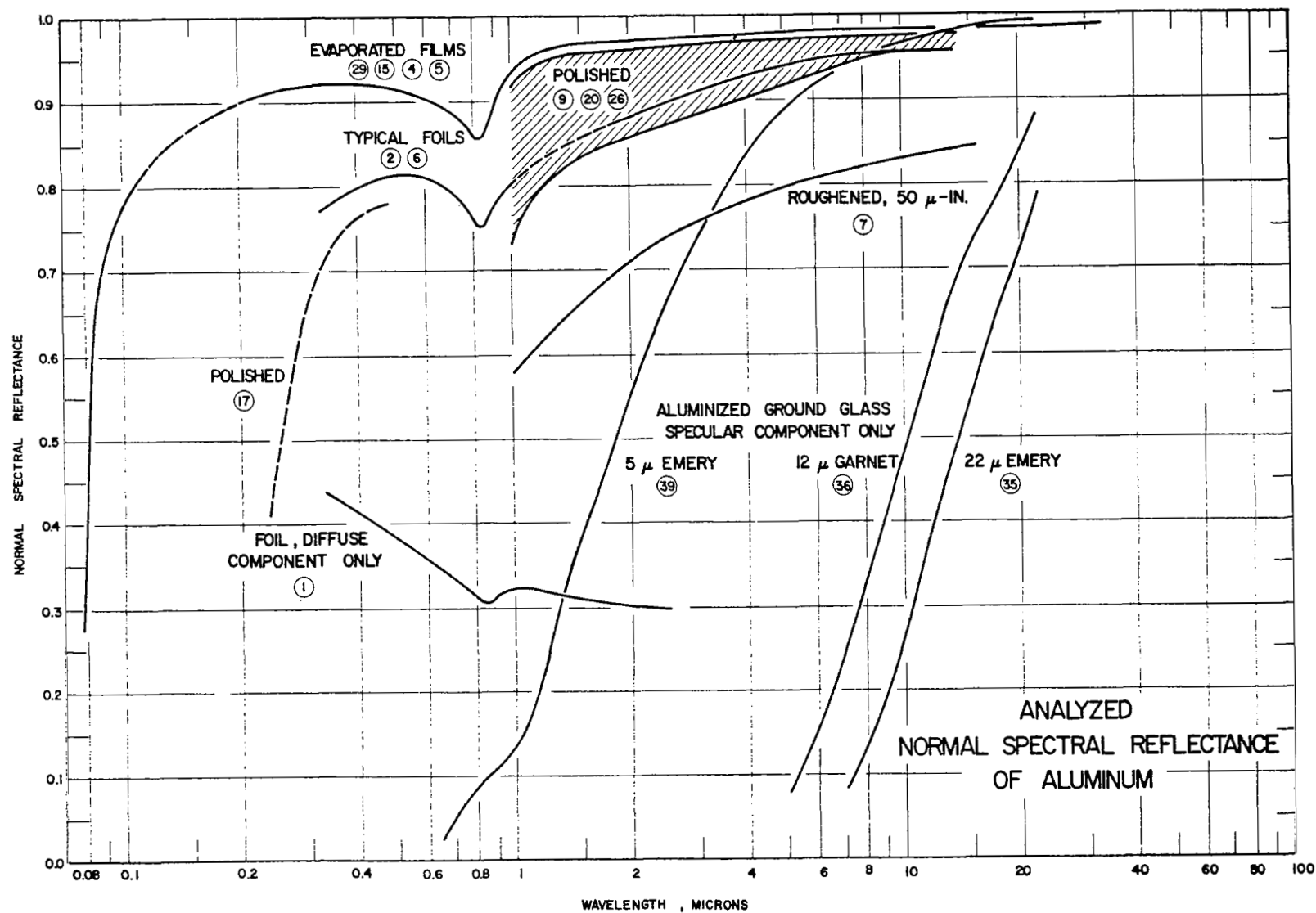
The Material Index is an alphabetical listing of all materials contained in the SERIES volume together with their respective page numbers for each sub-property. Many commercial designations are cross-indexed with their previous designations and synonyms for complete retrieval of the desired data. The Material Index and Grouping of Materials and List of Figures and Tables from Volume 7 is given in Appendix C.

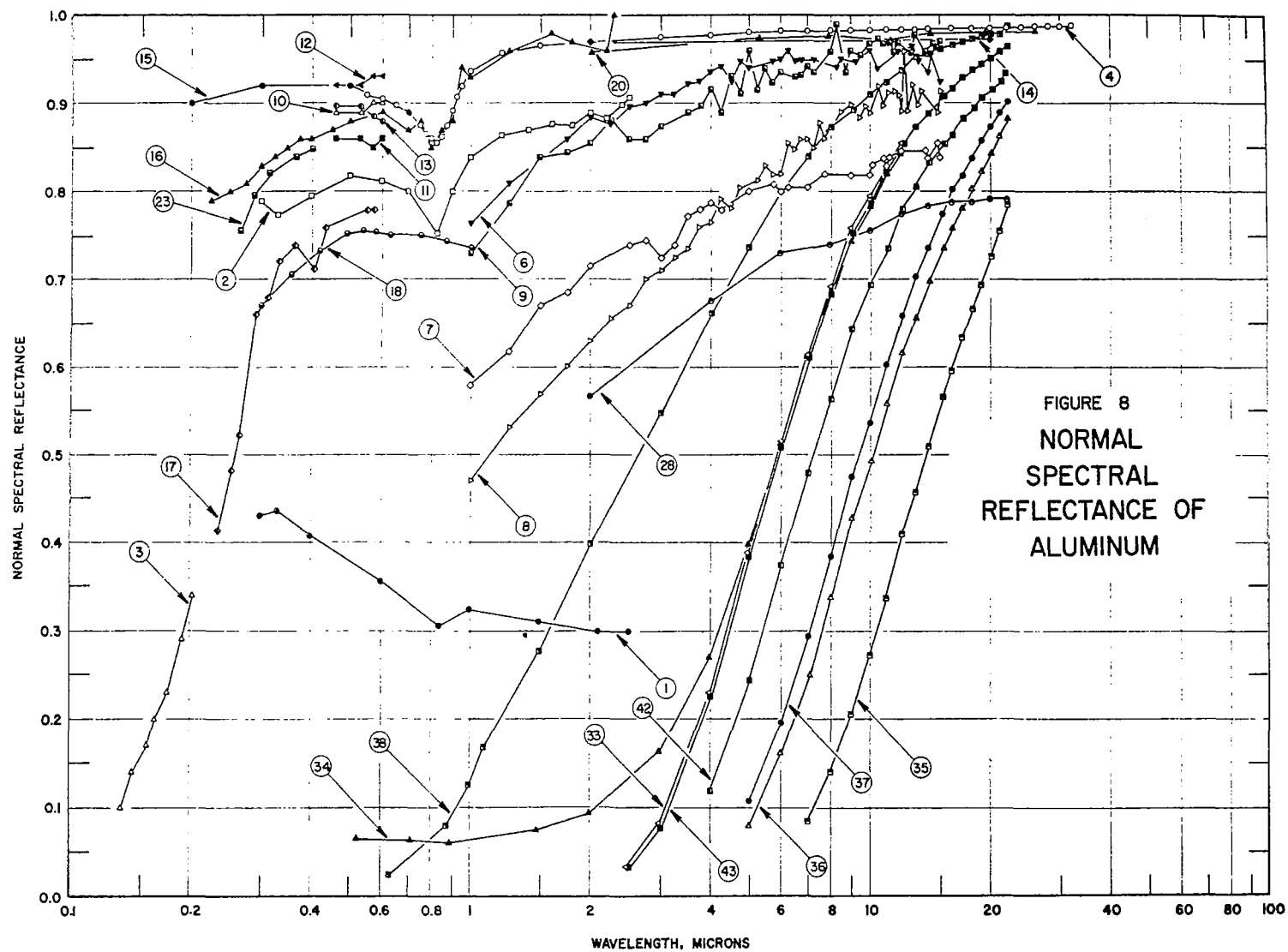
The bibliographic citation for each reference number is furnished in the numerically ordered list of references at the end of the book. Appendix D contains the Reference listing from Volume 7.

SAMPLE FIGURES, SPECIFICATION AND DATA TABLES

The following pages contain several selected sets of Figures, Specification and Data Tables for the purpose of demonstrating the presentation of data in the TPRC SERIES, Volume 7. Included are the following:

Title	Page
Normal Spectral Reflectance of Aluminum	10
Normal Spectral Absorptance of Copper	18
Normal Spectral Transmittance of Gallium	22
Hemispherical Total Emittance of Molybdenum	26
Normal Spectral Emittance of Tungsten	32
Normal Total Emittance of Iron + Chromium + Nickel Alloys	44





SPECIFICATION TABLE NO. 8 NORMAL SPECTRAL REFLECTANCE OF ALUMINUM

Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ θ' ω'	Reported Error, %	Composition (weight percent), Specifications and Remarks
1	123	1960	298	0.30-2.50	$\sim 0^\circ$ 2π		Foil; MgO reference; diffuse reflectance.
2	123	1960	298	0.30-2.50	$\sim 0^\circ$ 2π		Foil; cemented on fiberglass laminate; MgO reference.
3	124	1941	298	0.1347-0.2026	$\sim 0^\circ \sim 0^\circ$		An opaque film on glass deposited by the evaporation process; measured in vacuum (0.001 mm Hg).
4	125	1962	298	0.550-32	5° 5°	± 0.1	99.998 pure; Al film (0.065 to 0.11 μ thick), evaporated at 1×10^{-5} mm Hg, supersmooth fused quartz optical flats as substrate, no watermarks or other blemishes on the substrate surface, no shadows or streaks in the evaporated Al film; freshly prepared; measured in dry nitrogen.
5	125	1962	298	0.550-32	5° 5°	± 0.1	99.998 pure; Al film (0.065 to 0.11 μ thick), evaporated at 1×10^{-5} mm Hg, supersmooth fused quartz optical flats as substrate, no watermarks or other blemishes on the substrate surfaces, no shadows or streaks in the evaporated Al film; aged in air for several weeks; measured in dry nitrogen.
6	126	1953	300	1.00-15.00	5° 2π	± 2.6	Foil (0.001 in. thick); data extracted from smooth curve; converted from β ($2\pi, 5^\circ$).
7	126	1953	300	1.00-15.00	5° 2π	± 2.6	Disc (0.032 in. thick); polished, roughened (roughness approximately 50 microinches); data extracted from smooth curve; converted from β ($2\pi, 5^\circ$).
8	126	1953	300	1.00-15.00	5° 2π	± 4.3	Disc; commercial finish; data extracted from smooth curve; converted from β ($2\pi, 5^\circ$).
9	126	1953	300	1.00-15.00	5° 2π	± 2.7	Disc; polished; data extracted from smooth curve; converted from β ($2\pi, 5^\circ$).
10	127	1955	298	0.46-0.60	10° 2π	± 0.5	99 pure; vacuum deposited on glass; measured immediately after removed from vacuum chamber; calculated by authors from $\rho = 1 - \alpha$ using an incandescent tungsten lamp as source.
11	127	1955	298	0.46-0.60	10° 2π	± 0.5	Above specimen and conditions except exposed to the atmosphere for 8 days.
12	127	1955	298	0.46-0.60	10° 2π		99.99 pure; vacuum deposited on glass; measured immediately after removed from vacuum chamber; calculated by authors from $\rho = 1 - \alpha$ using an incandescent tungsten lamp as source.
13	127	1955	298	0.46-0.60	10° 2π		Above specimen and conditions except exposed to atmosphere for 8 days.
14	128	1962	298	2.00-20.00	$\sim 0^\circ$ 2π		Evaporated Al on mylar substrate (0.20 μ thick); illumination solid angle is cone of 0.034 steradians; converted from β ($2\pi, 0^\circ$).
15	129	1964	298	0.20-0.70	$\sim 0^\circ$ 2π		Evaporated aluminum; data extracted from smooth curve.
16	130	1934	298	0.225-2.3	$\sim 0^\circ \sim 0^\circ$		Deposited on a mirror by evaporation.
17	133	1934	298	0.235-0.578	$\sim 0^\circ \sim 0^\circ$	2	Disc; cold worked, annealed, etch tested, polished, stored in a solution of NaOH + NaF, washed and dried.
18	220	1965	298	0.300-1.000	$\sim 0^\circ$ 2π		Sand blasted.
19	222	1960	298	0.450-0.600	$\sim 7^\circ \sim 7^\circ$	< 0.16	Measured in air.
20	223	1962	298	2.01-25.96	$\sim 0^\circ$ 2π		Polished; converted from β ($2\pi, 0^\circ$).

SPECIFICATION TABLE NO. 8 (continued)

Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ θ' ω'	Reported Error, %	Composition (weight percent), Specifications and Remarks
21	223	1962	298	1.57-25.94	$\sim 0^\circ$ 2π		Above specimen and conditions except after particle impact.
22	223	1962	77	1.91-26.00	$\sim 0^\circ$ 2π		Above specimen and conditions.
23	224	1931	298	0.2653-0.4038	$\sim 5^\circ$ 2π		Acid-etched.
24	216	1949	~ 298	1.01-15.00	0° 2π	5	Foil; data extracted from smooth curve.
25	285	1962	298	1.97-13.05	$\sim 5^\circ \sim 5^\circ$		
26	336	1964	298	2.00-23.99	$\sim 0^\circ \sim 0^\circ$		Polished.
27	336	1964	298	2.00-23.99	$\sim 0^\circ \sim 0^\circ$		Above specimen and conditions except cratered with spherical particles (100 μ dia) of Zircalloy at 1.5 km sec ⁻¹ ; average crater dia 123 μ ; average crater depth 289 μ ; Knoop hardness 22 (100 g load).
28	336	1964	298	2.00-22.00	$\sim 0^\circ \sim 0^\circ$		Different sample, same as above specimen and conditions except cratered with spherical particles (100 μ dia) of tungsten at 7 km sec ⁻¹ ; average crater depth 54 μ ; average crater depth 183 μ ; Knoop hardness 22 (100 g load).
29	341	1967	298	0.079-0.1175	$\sim 0^\circ \sim 0^\circ$		Evaporated film; 99.999 pure; evaporated on microscope slide at 3×10^{-8} mm Hg; measured in vacuum (3×10^{-8} mm Hg) 4 min after evaporation.
30	341	1967	298	0.079-0.1175	$\sim 0^\circ \sim 0^\circ$		Different sample, same as above specimen and conditions except measured 8 min after evaporation.
31	341	1967	298	0.079-0.1175	$\sim 0^\circ \sim 0^\circ$		Different sample, same as above specimen and conditions except measured 12 min after evaporation.
32	341	1967	298	0.079-0.1175	$\sim 0^\circ \sim 0^\circ$		Different sample, same as above specimen and conditions except measured 16 min after evaporation.
33	344	1963	298	2.47-12.08	$\sim 5^\circ \sim 5^\circ$		Aluminized ground glass; aluminum mirror reference; $\omega' = 0.03$ Sr.
34	344	1963	298	0.52-12.07	$\sim 5^\circ \sim 5^\circ$		Aluminized ground steel; aluminum mirror reference; $\omega' = 0.03$ Sr.
35	344	1963	298	7.07-22.10	$\sim 5^\circ \sim 5^\circ$		Aluminized ground glass; glass ground with M302 grinding powder (Al_2O_3 emery) with average particle size of 22 μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.
36	344	1963	298	5.08-22.19	$\sim 5^\circ \sim 5^\circ$		Aluminized ground glass; glass ground with W6 grinding powder ($Fe_3Al_2(SiO_4)_3$ garnet) with average particle size of 12 μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.
37	344	1963	298	5.07-22.12	$\sim 5^\circ \sim 5^\circ$		Aluminized ground glass; glass ground with M 303.5 grinding powder (Al_2O_3 emery) with average particle size of 11 μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.
38	344	1963	298	0.63-22.11	$\sim 5^\circ \sim 5^\circ$		Aluminized ground glass; glass ground with W10 grinding powder ($Fe_3Al_2(SiO_4)_3$ garnet) with average particle size of 5 μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.
39	344	1963	298	0.64-22.14	$\sim 5^\circ \sim 5^\circ$		Aluminized ground glass; glass ground with M 305 grinding powder (Al_2O_3 emery) with average particle size of 5 μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.
40	344	1963	298	3.93-22.23	$\sim 5^\circ \sim 5^\circ$		Aluminized dense flint; flint ground with M303.5 grinding powder (Al_2O_3 emery) with average particle size of 11 μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.

SPECIFICATION TABLE NO. 8 (continued)

Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ θ' ω'	Reported Error, %	Composition (weight percent), Specifications and Remarks
41	344	1963	298	3.88-22.25	$\sim 3^\circ \sim 5^\circ$		Aluminized plate glass; glass ground with M303.5 grinding powder (Al_2O_3 emery) with average particle size of 11μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.
42	344	1963	298	4.06-22.23	$\sim 5^\circ \sim 5^\circ$		Aluminized Pyrex; Pyrex ground with M303.5 grinding powder (Al_2O_3 emery) with average particle size of 11μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.
43	344	1963	298	2.49-22.21	$\sim 5^\circ \sim 5^\circ$		Aluminized fused quartz; quartz ground with M303.5 grinding powder (Al_2O_3 emery) with average particle size of 11μ ; aluminum mirror reference; $\omega' = 0.03$ Sr.

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DATA TABLE NO. 8 (continued)

λ	ρ	λ	ρ	λ	ρ	λ	ρ	λ	ρ	λ	ρ	λ	ρ
<u>CURVE 9 (cont.)</u> <u>T = 300</u>		<u>CURVE 12</u> <u>T = 298</u>		<u>CURVE 16 (cont.)</u> <u>T = 298</u>		<u>CURVE 20</u> <u>T = 298</u>		<u>CURVE 24*</u> <u>T = ~298</u>		<u>CURVE 24 (cont.)*</u> <u>T = ~298</u>		<u>CURVE 27*</u> <u>T = 298</u>	
6.00	0.936	0.46	0.92	1.00	0.93	2.01	0.960	1.01	0.929	13.01	0.976	2.00	0.897
6.50	0.930	0.53	0.92	1.25	0.96	5.27	0.973	1.26	0.928	13.28	0.976	4.00	0.918
6.75	0.932	0.57	0.93	1.6	0.98	7.94	0.979	1.50	0.944	13.50	0.977	5.99	0.933
7.00	0.942	0.60	0.93	1.8	0.97	11.36	0.981	1.79	0.949	13.79	0.976	8.00	0.945
7.25	0.935			2.2	0.96	14.07	0.982	1.98	0.955	14.01	0.976	10.00	0.954
8.00	0.960			2.3	1.00	19.27	0.982	2.25	0.957	14.28	0.977	12.00	0.952
8.25	0.990	<u>CURVE 13</u> <u>T = 298</u>				25.96	0.984	2.50	0.959	14.53	0.973	14.00	0.954
8.75	0.935			<u>CURVE 17</u> <u>T = 298</u>				2.75	0.955	14.75	0.966	16.00	0.958
9.00	0.960	0.46	0.895			<u>CURVE 21*</u> <u>T = 298</u>		2.98	0.956	15.00	0.971	18.00	0.958
9.25	0.960	0.53	0.895	0.235	0.413			3.27	0.964			20.00	0.955
9.50	0.955	0.57	0.885	0.254	0.482	1.57	0.890	3.50	0.955	<u>CURVE 25*</u> <u>T = 298</u>		22.00	0.946
9.75	0.969	0.60	0.880	0.265	0.523	4.34	0.929	3.75	0.963			23.99	0.946
10.00	0.969			0.293	0.662	7.17	0.946	3.96	0.974				
10.50	0.975	<u>CURVE 14</u> <u>T = 298</u>		0.312	0.679	8.53	0.946	4.24	0.940	1.97	0.9670	<u>CURVE 28</u> <u>T = 298</u>	
10.75	0.970			0.334	0.722	11.73	0.953	4.48	0.968	2.97	0.9712		
11.00	0.972			0.366	0.740	14.46	0.954	4.76	0.969	3.95	0.9747		
11.25	0.972	2.00	0.970	0.406	0.712	17.38	0.961	4.98	0.969	4.98	0.9765	2.00	0.566
11.50	0.959	10.00	0.970*	0.435	0.759	19.33	0.961	5.24	0.972	6.00	0.9774	4.00	0.676
11.75	0.972	20.00	0.975	0.546	0.780	22.54	0.957	5.50	0.969	7.03	0.9778	5.99	0.731
12.00	0.960			0.578	0.781	25.94	0.952	5.73	0.965	8.03	0.9783	8.00	0.741
12.25	0.960	<u>CURVE 15</u> <u>T = 298</u>						5.98	0.977	9.00	0.9790	10.00	0.757
12.50	0.963			<u>CURVE 18</u> <u>T = 298</u>		<u>CURVE 22*</u> <u>T = 77</u>		6.26	0.975	10.00	0.9791	12.00	0.777
12.75	0.959							6.49	0.975	11.09	0.9796	14.00	0.786
13.00	0.980	0.20	0.90			1.91	0.908	6.74	0.971	12.04	0.9798	16.00	0.789
13.25	0.969	0.30	0.92	0.300	0.671	3.95	0.939	6.98	0.969	13.05	0.9805	18.00	0.790
13.50	0.964	0.50	0.92	0.357	0.706	5.94	0.937	7.26	0.969			20.00	0.792
13.75	0.960	0.70	0.89	0.419	0.733	7.90	0.945	7.51	0.970	<u>CURVE 26*</u> <u>T = 298</u>		22.00	0.792
14.00	0.965			0.488	0.753	9.94	0.955	7.76	0.976				
14.25	0.951	<u>CURVE 16</u> <u>T = 298</u>		0.538	0.756	11.97	0.952	7.98	0.974			<u>CURVE 29*</u> <u>T = 298</u>	
14.50	0.969			0.579	0.754	13.88	0.953	8.26	0.974	2.00	0.964		
14.75	0.965			0.627	0.751	15.89	0.959	8.53	0.978	4.00	0.970		
15.00	0.972	0.225	0.79	0.752	0.751	17.92	0.962	8.78	0.968	5.99	0.973	0.0790	0.325
		0.250	0.80	0.771	0.747	19.94	0.960	9.00	0.967	8.00	0.977	0.0833	0.624
<u>CURVE 10</u> <u>T = 298</u>		0.275	0.81	1.000	0.738	21.86	0.957	9.29	0.972	10.00	0.980	0.0920	0.767
0.46	0.89	0.300	0.83	<u>CURVE 19*</u> <u>T = 298</u>		23.93	0.949	9.55	0.971	12.00	0.981	0.1175	0.815
0.53	0.89	0.325	0.84			25.00	0.949	9.79	0.974	14.00	0.980		
0.57	0.90	0.350	0.85					10.01	0.973	16.00	0.980	<u>CURVE 30*</u> <u>T = 298</u>	
0.60	0.90	0.375	0.86	0.450	0.9083	<u>CURVE 23</u> <u>T = 298</u>		10.27	0.973	18.00	0.980		
		0.400	0.86	0.500	0.9082			10.53	0.973	20.00	0.979		
		0.450	0.87	0.550	0.9064	0.2653	0.756	10.77	0.973	22.00	0.979	0.0790	0.308
<u>CURVE 11</u> <u>T = 298</u>		0.500	0.88	0.600	0.9026	0.2890	0.796	11.04	0.973	23.99	0.978	0.0833	0.588
0.46	0.86	0.70	0.87	0.600	0.9026	0.3133	0.822	11.30	0.974			0.0920	0.728
0.53	0.86	0.75	0.88			0.3664	0.839	11.52	0.976			0.1175	0.811
0.57	0.85	0.8	0.85			0.4038	0.849	11.78	0.976				
0.60	0.86	0.85	0.87					12.02	0.976				
		0.9	0.88					12.27	0.977				
		0.95	0.94					12.51	0.974				
								12.79	0.976				

* Not shown on plot

* Not shown on plot

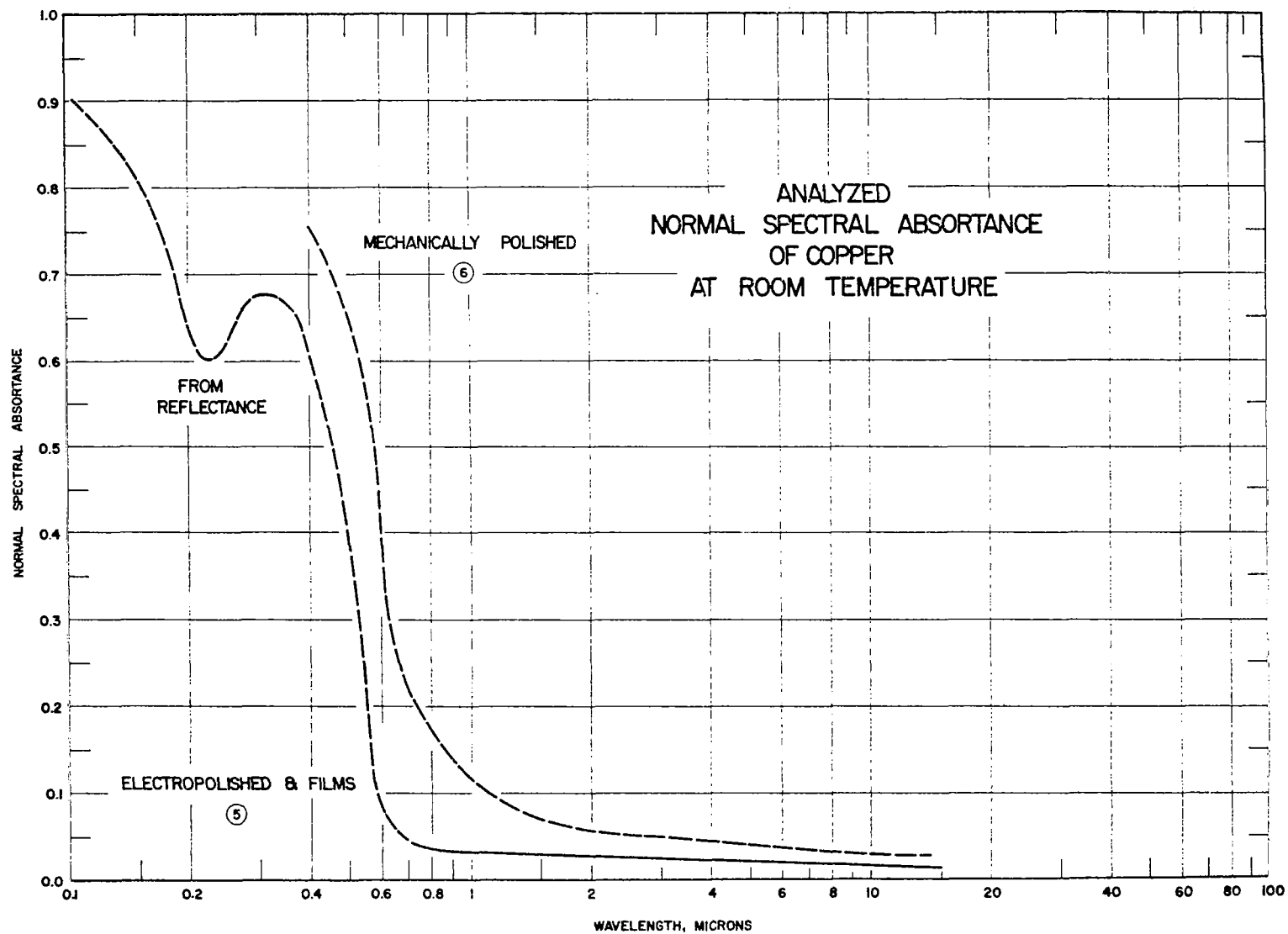
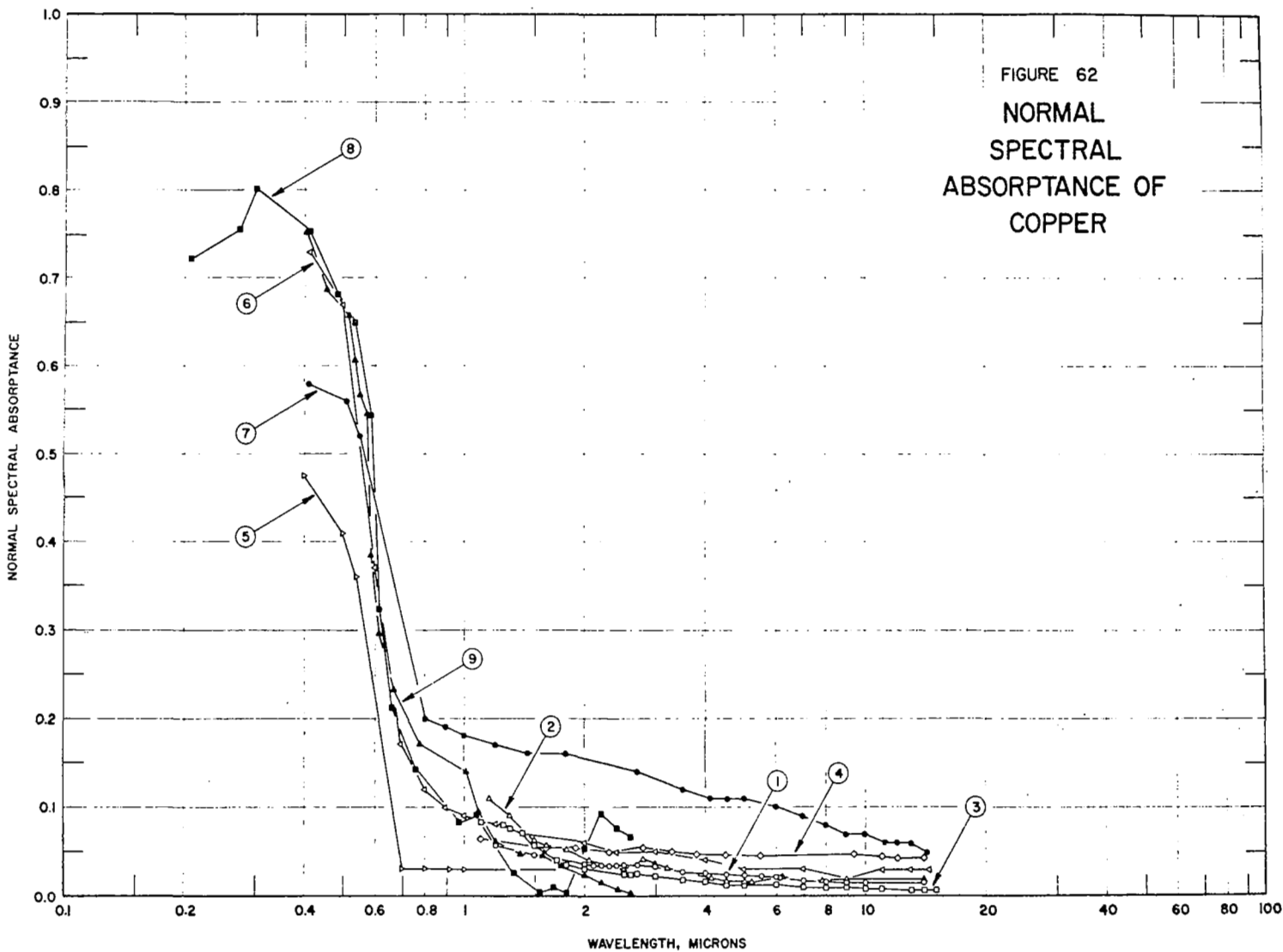


FIGURE 62
NORMAL
SPECTRAL
ABSORPTANCE OF
COPPER



SPECIFICATION TABLE NO. 62 NORMAL SPECTRAL ABSORPTANCE OF COPPER

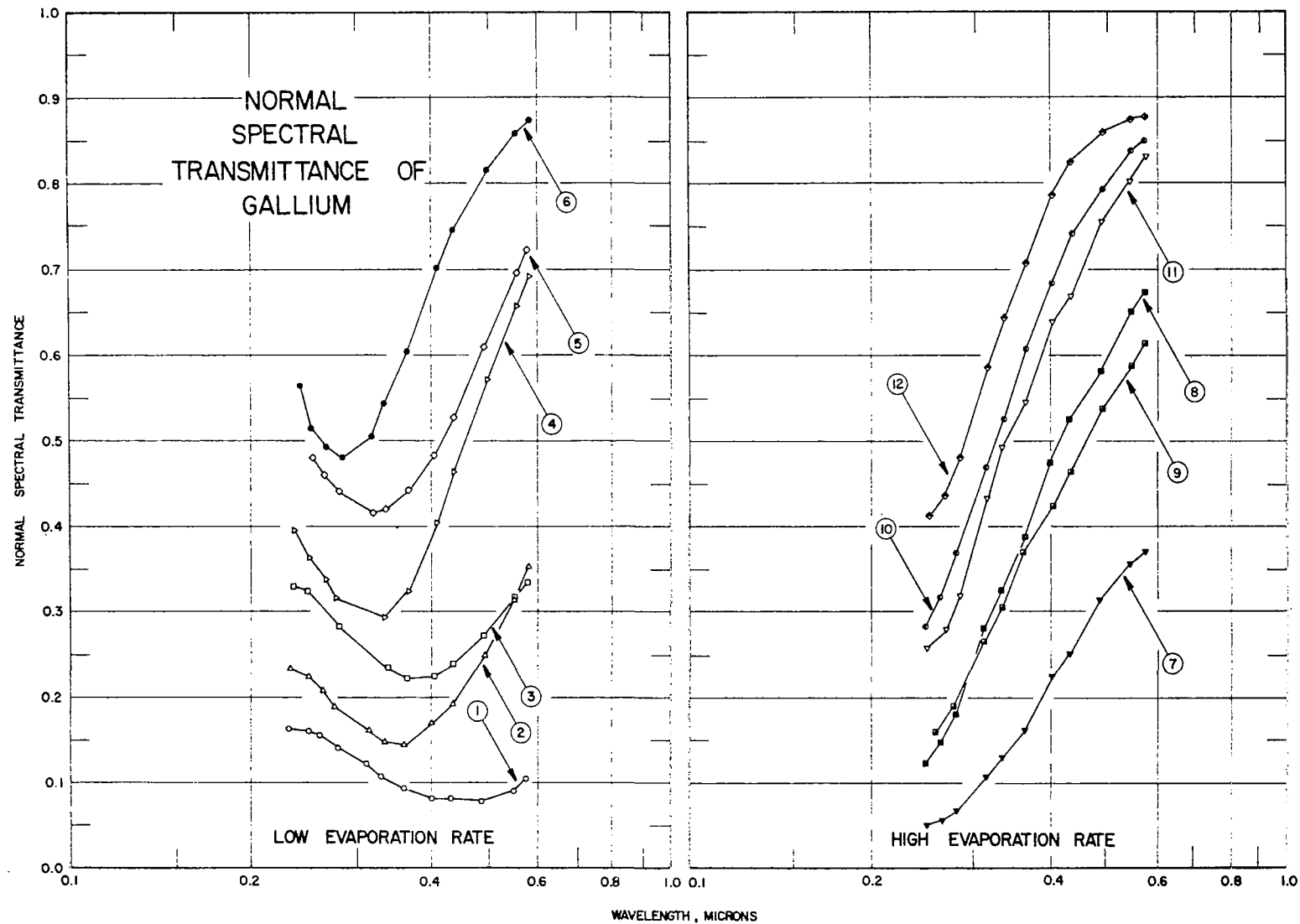
Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ	Reported Error, %	Composition (weight percent), Specifications and Remarks
1	30	1963	294	1.20-14.00	$\sim 0^\circ$		Mechanically polished (surface roughness 0.02μ peak to peak and 5μ lateral); measured in air; data extracted from smooth curve; [Author's designation: Sample 3].
2	30	1963	294	1.15-14.00	$\sim 0^\circ$		Above specimen and conditions except heated at 450 K for 3 hrs; surface oxidation possible.
3	30	1963	294	1.10-15.00	$\sim 0^\circ$		Above specimen and conditions except heated at 922 K for 3 hrs.
4	30	1963	294	1.10-14.00	$\sim 0^\circ$		Above specimen and conditions except heated at 1222 K for 102 hrs.
5	65	1962	294	0.40-2.00	$\sim 0^\circ$		Electropolished; calculated from (1 - ρ).
6	65	1962	294	0.41-14.40	$\sim 0^\circ$		Mechanically polished; calculated from (1 - ρ).
7	65	1962	294	0.41-14.20	$\sim 0^\circ$		Above specimen and conditions except roughened with sand paper; surface roughness 1.25μ .
8	307	1954	~ 298	0.207-2.600	$\sim 0^\circ$		Data extracted from smooth curve.
9	307	1954	~ 298	0.401-2.600	$\sim 0^\circ$		Polished; data extracted from smooth curve.

DATA TABLE NO. 62 NORMAL SPECTRAL ABSORPTANCE OF COPPER

[Wavelength, λ, μ ; Absorptance, α ; Temperature, T, K]

λ	α	λ	α	λ	α	λ	α	λ	α
<u>CURVE 1</u>		<u>CURVE 2 (cont.)</u>		<u>CURVE 4 (cont.)</u>		<u>CURVE 7</u>		<u>CURVE 8 (cont.)</u>	
T = 294						T = 294			
		6.20	0.021	3.30	0.050			2.000	0.053
1.20	0.057	7.00	0.019*	3.80	0.048	0.41	0.58	2.200	0.091
1.50	0.046	7.80	0.018	4.50	0.047	0.51	0.56	2.400	0.075
2.00	0.036	9.00	0.018	5.50	0.046	0.55	0.52	2.600	0.067
2.10	0.034	14.00	0.019	9.40	0.048	0.80	0.20		
2.20	0.034			11.00	0.044	0.90	0.19	<u>CURVE 9</u>	
2.30	0.033	<u>CURVE 3</u>		12.00	0.043	1.00	0.18	T = ~298	
2.40	0.034	T = 294		14.00	0.043	1.20	0.17		
2.50	0.035					1.44	0.16	0.401	0.751
2.70	0.035	1.10	0.082	<u>CURVE 5</u>		1.80	0.16	0.457	0.686
3.00	0.033	1.25	0.080	T = 294		2.70	0.14	0.515	0.658
3.50	0.028	1.30	0.076			3.50	0.12	0.539	0.607
4.00	0.027	1.40	0.070	0.40	0.475	4.05	0.11	0.549	0.567
4.50	0.025	1.50	0.057	0.50	0.410	4.55	0.11	0.579	0.544
5.00	0.023	1.70	0.040	0.54	0.360	5.00	0.11	0.583	0.389
5.50	0.022	1.80	0.035	0.70	0.031	6.00	0.10	0.612	0.296
6.00	0.021	2.00	0.031	0.80	0.031	7.00	0.09	0.664	0.231
6.10	0.021	2.50	0.024	0.92	0.031	8.00	0.08	0.779	0.170
7.00	0.018	2.60	0.024	1.00	0.030	9.00	0.07	1.001	0.104
8.00	0.017	2.70	0.025	2.00	0.031*	10.00	0.07	1.096	0.095
9.00	0.015	3.00	0.022	<u>CURVE 6</u>		11.20	0.06	1.200	0.060
10.00	0.015	3.50	0.019	T = 294		12.00	0.06	1.394	0.048
14.00	0.015	4.00	0.016			13.00	0.06	1.594	0.047
		4.50	0.013	0.41	0.73	14.20	0.05	1.798	0.034
<u>CURVE 2</u>		5.00	0.012	0.50	0.67	<u>CURVE 8</u>		2.000	0.023
T = 294		6.00	0.012	0.60	0.37	T = ~298		2.200	0.014
1.15	0.110	7.00	0.011	0.70	0.17			2.400	0.008
1.30	0.090	8.00	0.010	0.80	0.12			2.600	0.023
1.50	0.062	9.00	0.009	0.90	0.10	0.207	0.724		
1.60	0.057	10.00	0.008	1.00	0.09	0.275	0.755		
1.80	0.051	11.00	0.008	1.20	0.08	0.301	0.802		
2.05	0.040	13.00	0.007	1.40	0.07*	0.415	0.752		
2.30	0.032*	14.00	0.007	1.60	0.06	0.484	0.681		
2.50	0.031	15.00	0.007	2.00	0.05	0.535	0.649		
2.80	0.041			2.40	0.05	0.583	0.542		
3.00	0.035	<u>CURVE 4</u>		3.00	0.05	0.615	0.324		
3.20	0.031	T = 294		4.00	0.04	0.660	0.212		
3.50	0.028*	1.10	0.064	5.00	0.03	0.759	0.141		
4.10	0.020	1.50	0.055*	7.00	0.03	0.971	0.083		
4.50	0.017	1.60	0.054	9.00	0.02*	1.075	0.090		
5.00	0.016	1.90	0.054	11.00	0.03	1.340	0.026		
5.20	0.016	2.30	0.050	13.00	0.03	1.555	0.004		
6.00	0.020*	2.80	0.055	14.40	0.03	1.688	0.010		
						1.801	0.003		

* Not shown on plot



SPECIFICATION TABLE NO. 73 NORMAL SPECTRAL TRANSMITTANCE OF GALLIUM

Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ θ' ω'	Reported Error, %	Composition (weight percent), Specifications and Remarks
1	229	1963	298	0.231-0.574	$\sim 0^\circ$ $\sim 0^\circ$		Vacuum deposited thin film of gallium (19 m μ thick); measured in vacuum; spectral Philips Zn, Cd and Hg Lamp sources; 3.5 \AA min $^{-1}$ evaporation rate.
2	229	1963	298	0.233-0.579	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as above specimen and conditions except 11 m μ thick; 2 \AA min $^{-1}$ evaporation rate.
3	229	1963	298	0.236-0.578	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as above specimen and conditions except 9.5 m μ thick.
4	229	1963	298	0.236-0.581	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as above specimen and conditions except 5 m μ thick; 2 \AA min $^{-1}$ evaporation rate.
5	229	1963	298	0.242-0.577	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as above specimen and conditions except 4.5 m μ thick.
6	229	1963	298	0.242-0.584	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as above specimen and conditions except 2.5 m μ thick; 2 \AA min $^{-1}$ evaporation rate.
7	229	1963	298	0.250-0.579	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as above specimen and conditions except 42 m μ thick; 300 \AA min $^{-1}$ evaporation rate; supercooled liquid suspected in the evaporated film.
8	229	1963	298	0.248-0.578	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as curve 7 specimen and conditions except 28 m μ thick; 11 \AA min $^{-1}$ evaporation rate.
9	229	1963	298	0.258-0.579	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as curve 7 specimen and conditions except 21 m μ thick.
10	229	1963	298	0.247-0.579	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as curve 7 specimen and conditions except 14 m μ thick; 11 \AA min $^{-1}$ evaporation rate.
11	229	1963	298	0.249-0.581	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as curve 7 specimen and conditions except 10 m μ thick.
12	229	1963	298	0.252-0.581	$\sim 0^\circ$ $\sim 0^\circ$		Different sample, same as curve 7 specimen and conditions except 7 m μ thick; 11 \AA min $^{-1}$ evaporation rate.

DATA TABLE NO. 73 NORMAL SPECTRAL TRANSMITTANCE OF GALLIUM

[Temperature, T, K; Transmittance, τ ; Wavelength, λ , μ]

λ	τ	λ	τ	λ	τ	λ	τ
<u>CURVE 1</u> T = 298		<u>CURVE 4</u> T = 298		<u>CURVE 7</u> T = 298		<u>CURVE 10</u> T = 298	
0.231	0.162	0.236	0.396	0.250	0.050	0.247	0.282
0.249	0.160	0.250	0.362	0.265	0.057	0.262	0.318
0.261	0.155	0.267	0.338	0.280	0.068	0.278	0.369
0.278	0.140	0.277	0.317	0.313	0.107	0.314	0.468
0.311	0.122	0.333	0.294	0.333	0.130	0.336	0.525
0.329	0.107	0.365	0.324	0.364	0.161	0.367	0.607
0.359	0.094	0.408	0.404	0.405	0.224	0.405	0.684
0.401	0.081	0.435	0.463	0.432	0.251	0.438	0.741
0.430	0.081	0.495	0.572	0.487	0.314	0.494	0.794
0.484	0.078	0.551	0.658	0.548	0.356	0.550	0.836
0.547	0.091	0.581	0.693	0.579	0.370	0.579	0.851
0.574	0.105						
<u>CURVE 2</u> T = 298		<u>CURVE 5</u> T = 298		<u>CURVE 8</u> T = 298		<u>CURVE 11</u> T = 298	
0.233	0.233	0.242	0.504	0.248	0.122	0.249	0.257
0.250	0.224	0.253	0.481	0.262	0.147	0.268	0.279
0.263	0.207	0.265	0.459	0.279	0.180	0.283	0.318
0.276	0.189	0.279	0.441	0.311	0.280	0.315	0.432
0.314	0.161	0.319	0.414	0.332	0.325	0.334	0.492
0.333	0.148	0.336	0.420	0.364	0.388	0.365	0.544
0.360	0.144	0.366	0.442	0.402	0.473	0.406	0.639
0.402	0.170	0.404	0.483	0.433	0.529	0.436	0.679
0.434	0.193	0.436	0.527	0.490	0.582	0.493	0.755
0.490	0.250	0.490	0.610	0.549	0.651	0.549	0.802
0.549	0.313	0.557	0.697	0.578	0.673	0.581	0.833
0.579	0.352	0.577	0.722				
<u>CURVE 3</u> T = 298		<u>CURVE 6</u> T = 298		<u>CURVE 9</u> T = 298		<u>CURVE 12</u> T = 298	
0.236	0.330	0.242	0.564	0.258	0.158	0.252	0.412
0.248	0.324	0.253	0.515	0.275	0.190	0.268	0.437
0.280	0.282	0.267	0.493	0.311	0.264	0.284	0.482
0.337	0.234	0.284	0.481	0.333	0.303	0.317	0.587
0.364	0.222	0.318	0.505	0.362	0.370	0.336	0.644
0.405	0.225	0.334	0.544	0.406	0.424	0.366	0.708
0.433	0.239	0.364	0.604	0.434	0.462	0.407	0.787
0.489	0.272	0.409	0.701	0.491	0.538	0.436	0.825
0.549	0.316	0.435	0.747	0.550	0.589	0.496	0.860
0.578	0.335	0.497	0.818	0.579	0.614	0.551	0.875
		0.554	0.860			0.581	0.879
		0.584	0.875				

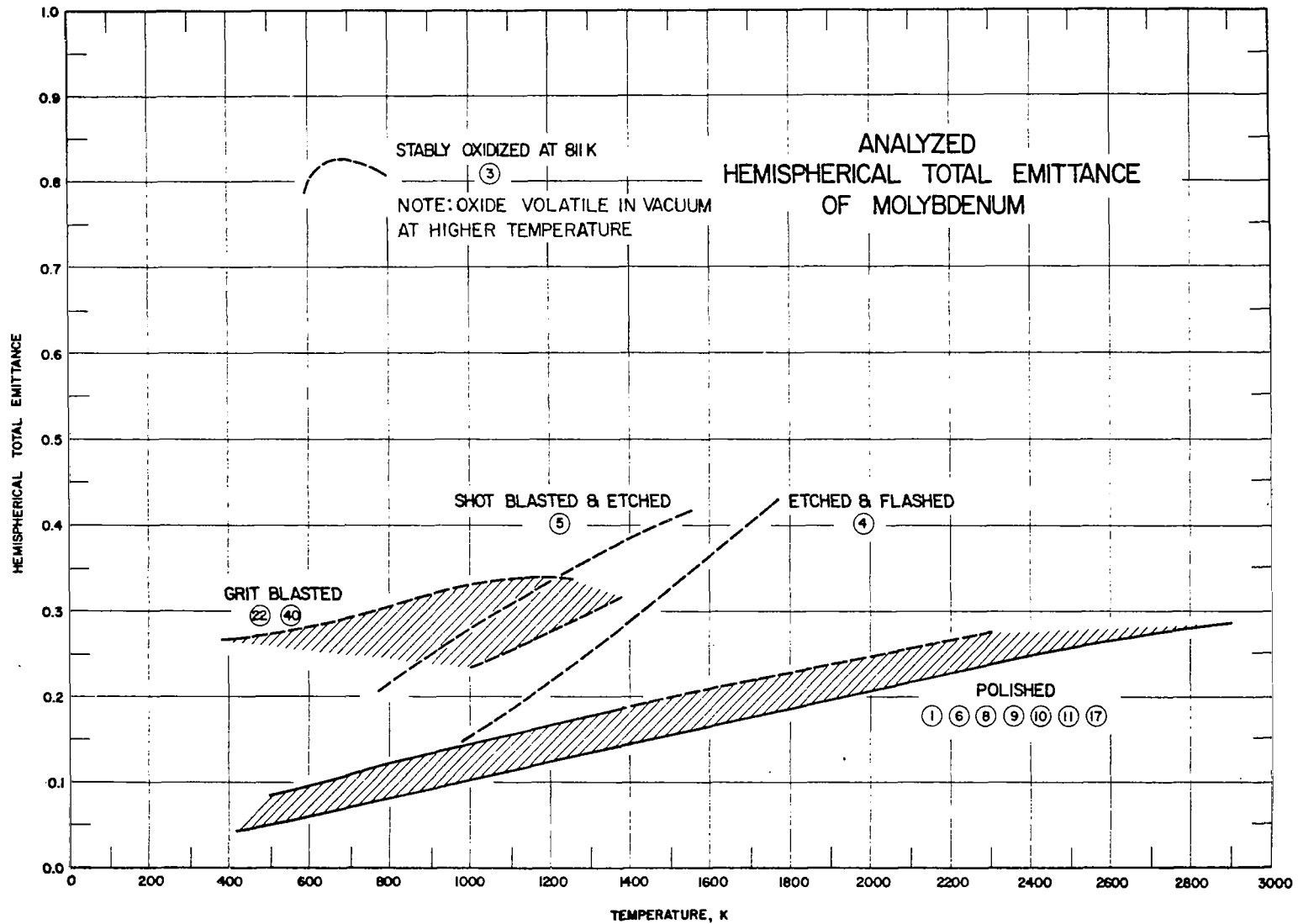
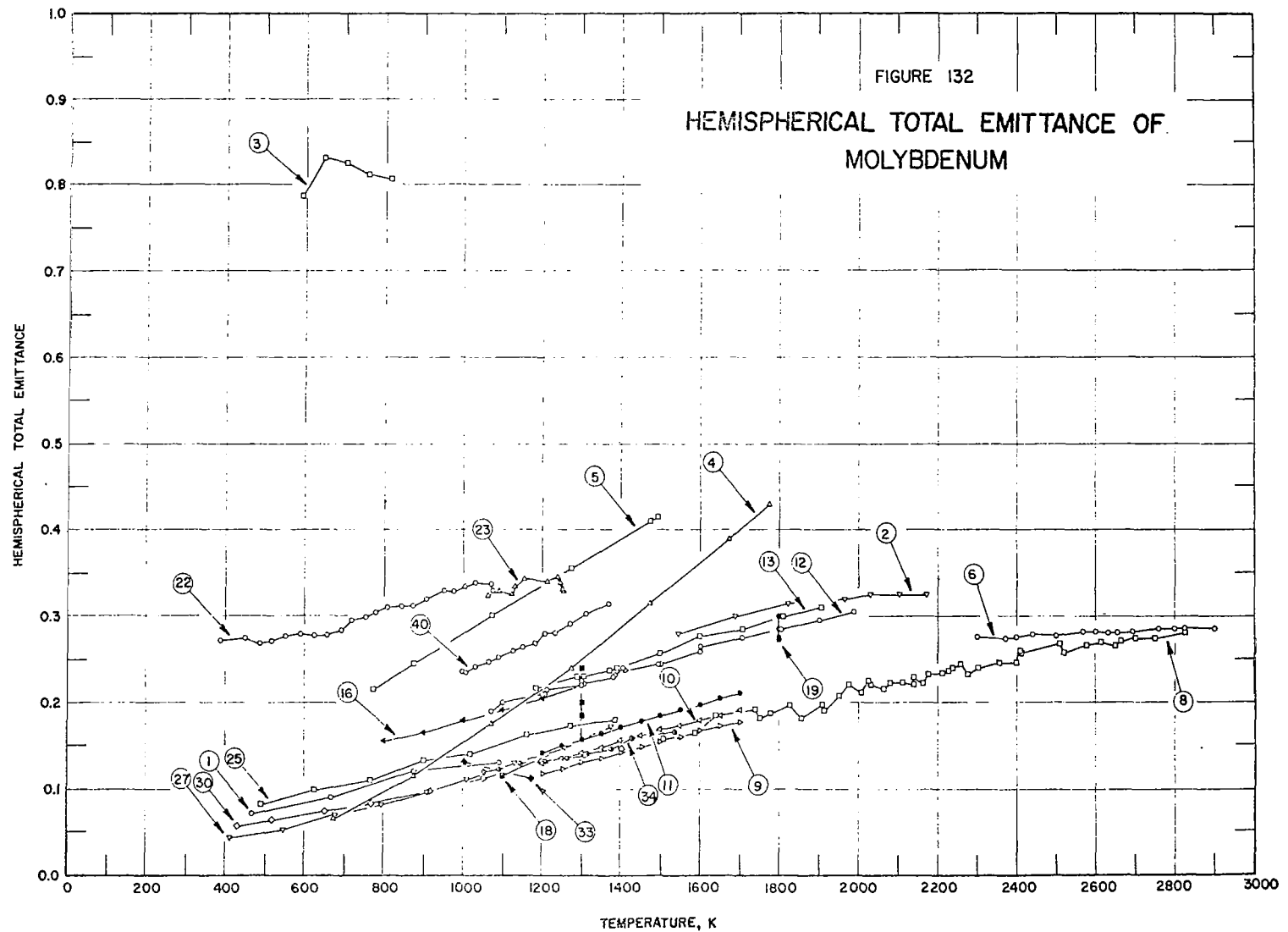


FIGURE 132

HEMISPHERICAL TOTAL EMITTANCE OF MOLYBDENUM



SPECIFICATION TABLE NO. 132 HEMISPHERICAL TOTAL EMITTANCE OF MOLYBDENUM

Curve No.	Ref. No.	Year	Temperature Range, K	Reported Error, %	Composition (weight percent), Specifications and Remarks
1	47	1961	468-1093	< 10	Vacuum arc cast, machined, extruded, recrystallized, rolled; disc (0.04 in. thick); ground with 600 grit carborundum and polished on a wet cloth lap with unlevigated jewelers rouge; measured in vacuum (10^{-5} mm Hg).
2	69	1960	1544-2172	± 10	Measured in vacuum.
3	22	1958	589-811	≤ 2	Stably oxidized at 811 K in quiescent air.
4	58	1961	673-1773	± 2.5	Lightly etched and flashed in vacuum at 2073 K for 10 min; measured in vacuum ($< 5 \times 10^{-6}$ mm Hg); data extracted from smooth curve.
5	58	1961	773-1493	± 2.5	Shot-blasted and pickled in hydrochloric acid to remove iron; measured in vacuum ($< 5 \times 10^{-6}$ mm Hg); data extracted from smooth curve.
6	70	1960	2300-2900		0.18 Fe, 0.073 Si, 0.04 C, 0.036 Mn, 0.005 O ₂ , 0.01 others, Mo balance; cast under inert gas; hot rolled; successively polished with No. 1-, 0-, 00-, 000-, and 0000- abrasive papers; measured in argon.
7	71	1962	1540-2180	± 10	Measured in vacuum ($< 10^{-5}$ mm Hg).
8	72	1963	1506-2825		0.07 - 0.09 Fe, 0.04 - 0.06 Nb, 0.001 - 0.003 Mn, 0.001 - 0.003 Si, 0.0004 - 0.0006 Cu, 0.0001 - 0.0005 Mg, Mo balance; thin walled tube; polished using felt with a GOI paste; annealed; measured in vacuum.
9	73	1964	1200-1700	< 2.3	99.96 Mo, 0.004 SiO ₂ , 0.004 CaO and MgO, 0.026 sesquioxides; prepared by rubbing with abrasive paper; surface roughness 0.063 - 0.050 μ RMS; measured in vacuum (10^{-3} to 10^{-4} mm Hg); [Author's designation: Specimen 1].
10	73	1964	1200-1700	< 2.3	Different sample, same as curve 9 specimen and conditions; [Author's designation: Specimen 2].
11	73	1964	1200-1700	< 2.3	Different sample, same as curve 9 specimen; same conditions except surface roughness 1.25 - 1.00 μ RMS; [Author's designation: Specimen 3].
12	54	1962	1070-1990	± 4	Degreased with acetone, cleaned with a rubber eraser, wiped with acetone; measured in vacuum (10^{-4} to 10^{-6} mm Hg); same data reported for both samples; [Author's designation: Sample No. 1 and Sample No. 2].
13	54	1962	1185-1905	± 4	Degreased with acetone, cleaned with a rubber eraser, wiped with acetone; aged for 1 hr at 1773 K; measured in vacuum (10^{-4} to 10^{-6} mm Hg); [Author's designation: Sample 2].
14	54	1962	1100-1800	± 4	Polished using rouge in wax on a buffing wheel; measured in vacuum (10^{-4} to 10^{-6} mm Hg); [Author's designation: Sample No. 3].
15	54	1962	800-1300	± 4	Polished using fine aluminum oxide powder on a circular rotatable drum with a rotating lap; measured in vacuum (10^{-4} to 10^{-6} mm Hg); cycle 1; [Author's designation: Sample No. 5].
16	54	1962	800-1300	± 4	Above specimen and conditions; cycle 2.
17	54	1962	1400-1800	± 4	Different sample, same as curve 15 specimen and conditions; cycle 1; [Author's designation: Sample No. 6].
18	54	1962	1100-1300	± 4	Above specimen and conditions; cycle 2.

SPECIFICATION TABLE NO. 132 (continued)

Curve No.	Ref. No.	Year	Temperature Range, K	Reported Error, %	Composition (weight percent), Specifications and Remarks
19	54	1962	1800	± 4	Above specimen and conditions; cycle 3.
20	54	1962	1300-1500	± 4	Above specimen and conditions; cycle 4.
21	54	1962	1000-2000	± 4	Above specimen and conditions; cycle 5.
22	12	1962	385.6-1075.1	± 2.7	Grit blasted with aluminum oxide No. 90 (PMC-3043A); measured in vacuum ($<2.9 \times 10^{-6}$ mm Hg); Run No. 1.
23	12	1962	1061.1-1251.5	± 2.7	Above specimen and conditions; Run No. 2A.
24	12	1962	1255.1-1235.9	± 2.7	Above specimen and conditions; Run No. 2B.
25	12	1962	491.2-1385.2	± 2.7	Vapor-blasted with Techline Liquabrasive, PMC-3067, grit No. 325; measured in vacuum ($<5 \times 10^{-6}$ mm Hg); Run No. 1.
26	12	1962	545.2-1375.7	± 2.7	Above specimen and conditions; Run No. 2.
27	12	1962	412-1373	± 2.7	Above specimen and conditions; Run No. 3.
28	12	1962	410-1244	± 2.7	Above specimen and conditions; Run No. 4A.
29	12	1962	1378-1273	± 2.7	Above specimen and conditions; Run No. 4B.
30	12	1962	429.2-1401.2	± 2.7	Chemically cleaned; measured in vacuum ($<5 \times 10^{-6}$ mm Hg); Run No. 1.
31	12	1962	449.7-1405.2	± 2.7	Above specimen and conditions; Run No. 2.
32	12	1962	407.2-1374.2	± 2.7	Above specimen and conditions; Run No. 3A.
33	12	1962	1002.4-1169.2	± 2.7	Above specimen and conditions; Run No. 3B.
34	12	1962	1054-1536	± 2.3	As received; measured in vacuum ($<2 \times 10^{-6}$ mm Hg); Run No. 1A.
35	12	1962	1539-1100	± 2.3	Above specimen and conditions; Run No. 1B.
36	12	1962	1242-1540	± 2.3	Above specimen and conditions; Run No. 2.
37	12	1962	1045-1539	± 2.7	As received; measured in vacuum ($<2 \times 10^{-6}$ mm Hg); Run No. 1A.
38	12	1962	1535-1097	± 2.7	Above specimen and conditions; Run No. 1B.
39	12	1962	1239-1541	± 2.7	Above specimen and conditions; Run No. 2.
40	12	1962	1368-998	± 2.7	Grit blasted with aluminum oxide No. 90 (PMC-3043A); measured in vacuum ($<5.1 \times 10^{-6}$ mm Hg); Run No. 1.

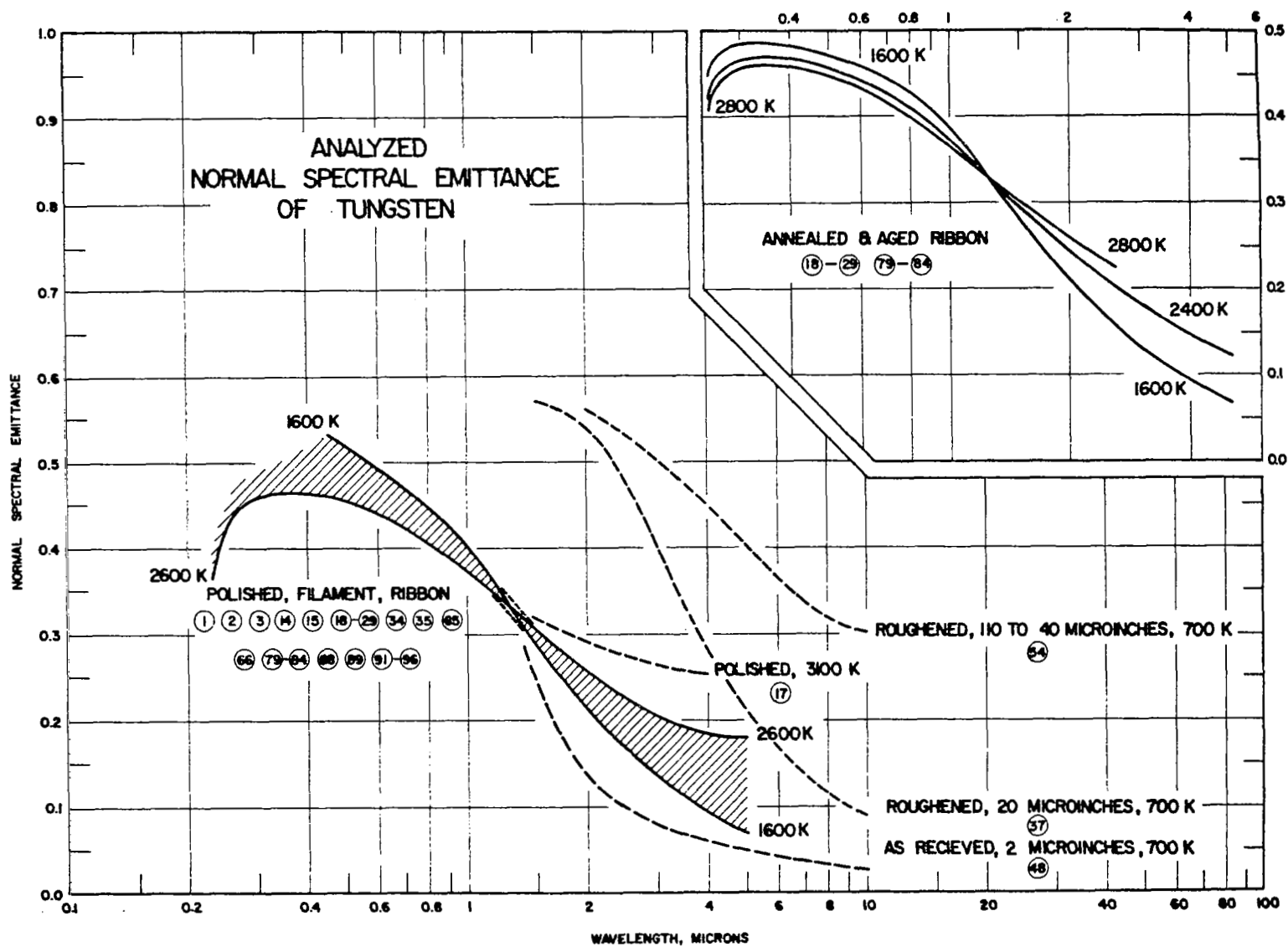
T	€	T	€	T	€	T	€	T	€	T	€	T	€
<u>CURVE 1</u>		<u>CURVE 6 (cont.)</u>		<u>CURVE 8 (cont.)</u>		<u>CURVE 10 (cont.)</u>		<u>CURVE 13 (cont.)</u>		<u>CURVE 18</u>		<u>CURVE 22 (cont.)</u>	
468	0.07	2440	0.279	2175	0.2325	1650	0.186	1290	0.230	1100	0.115	778.2	0.303
668	0.09	2500	0.278	2212	0.2337	1700	0.192	1305	0.230	1200	0.135*	808.5	0.310
873	0.12	2565	0.282	2225	0.2375			1370	0.237	1300	0.157*	845.3	0.312
1093	0.13	2600	0.282	2218	0.2400	<u>CURVE 11</u>		1390	0.240	1300	0.185	870.6	0.312
		2630	0.281	2237	0.2450	1200	0.143	1500	0.257	1300	0.200	904.0	0.320
<u>CURVE 2</u>		2655	0.281	2275	0.2325	1250	0.150	1600	0.277	1300	0.240	952.2	0.330
1544	0.280	2695	0.282	2300	0.2400	1300	0.157	1705	0.285			975.7	0.329
1686	0.300	2760	0.285	2356	0.2475	1350	0.164	1810	0.300	<u>CURVE 19</u>		1001.7	0.334
1822	0.315	2800	0.285	2400	0.2475	1400	0.171	1905	0.310	1800	0.300	1030.2	0.339
1967	0.320	2825	0.285	2406	0.2600	1450	0.178			1800	0.275	1071.5	0.337
2033	0.325	2900	0.285	2412	0.2575	1500	0.185	<u>CURVE 14*</u>		1800	0.273	1075.1	0.330
2103	0.325			2518	0.2575	1550	0.192	1100	0.195			<u>CURVE 23</u>	
2172	0.325	<u>CURVE 7*</u>		2506	0.2687	1600	0.198	1195	0.207			1061.1	0.323
		1540	0.275	2575	0.2675	1650	0.205	1400	0.235	<u>CURVE 20*</u>		1092.2	0.329
<u>CURVE 3</u>		1690	0.300	2612	0.2700	1700	0.211	1600	0.255	1300	0.215	1123.6	0.326
589	0.785	1830	0.310	2662	0.2712	<u>CURVE 12</u>		1800	0.280	1300	0.230	1134.1	0.335
644	0.830	1970	0.320	2650	0.2662	1070	0.190	<u>CURVE 15*</u>		1500	0.262	1156.4	0.344
700	0.825	2040	0.325	2700	0.2750	1100	0.200	800	0.085	<u>CURVE 21*</u>		1210.2	0.341
755	0.810	2110	0.326	2750	0.2750	1200	0.210	900	0.090	1000	0.185	1234.8	0.346
811	0.805	2180	0.326	2825	0.2812	1205	0.210	1000	0.100	1100	0.200	1241.6	0.339
<u>CURVE 4</u>		<u>CURVE 8</u>		<u>CURVE 9</u>		1210	0.215	1100	0.113	1200	0.215	1247.8	0.332
673	0.065	1506	0.1580	1200	0.117	1300	0.220	1200	0.132	1300	0.230	1251.5	0.328
873	0.115	1588	0.1650	1250	0.123	1300	0.225	1300	0.157	1400	0.247	<u>CURVE 24*</u>	
1073	0.175	1638	0.1850	1300	0.130	1305	0.223	1310	0.170	1600	0.280	1255.1	0.324
1273	0.240	1738	0.1925	1350	0.136	1380	0.230	1310	0.177	1700	0.295	1235.9	0.320
1473	0.315	1750	0.1825	1400	0.143	1385	0.233	1310	0.185	1800	0.305	<u>CURVE 25</u>	
1673	0.390	1775	0.1875	1450	0.149	1405	0.240	1300	0.210				

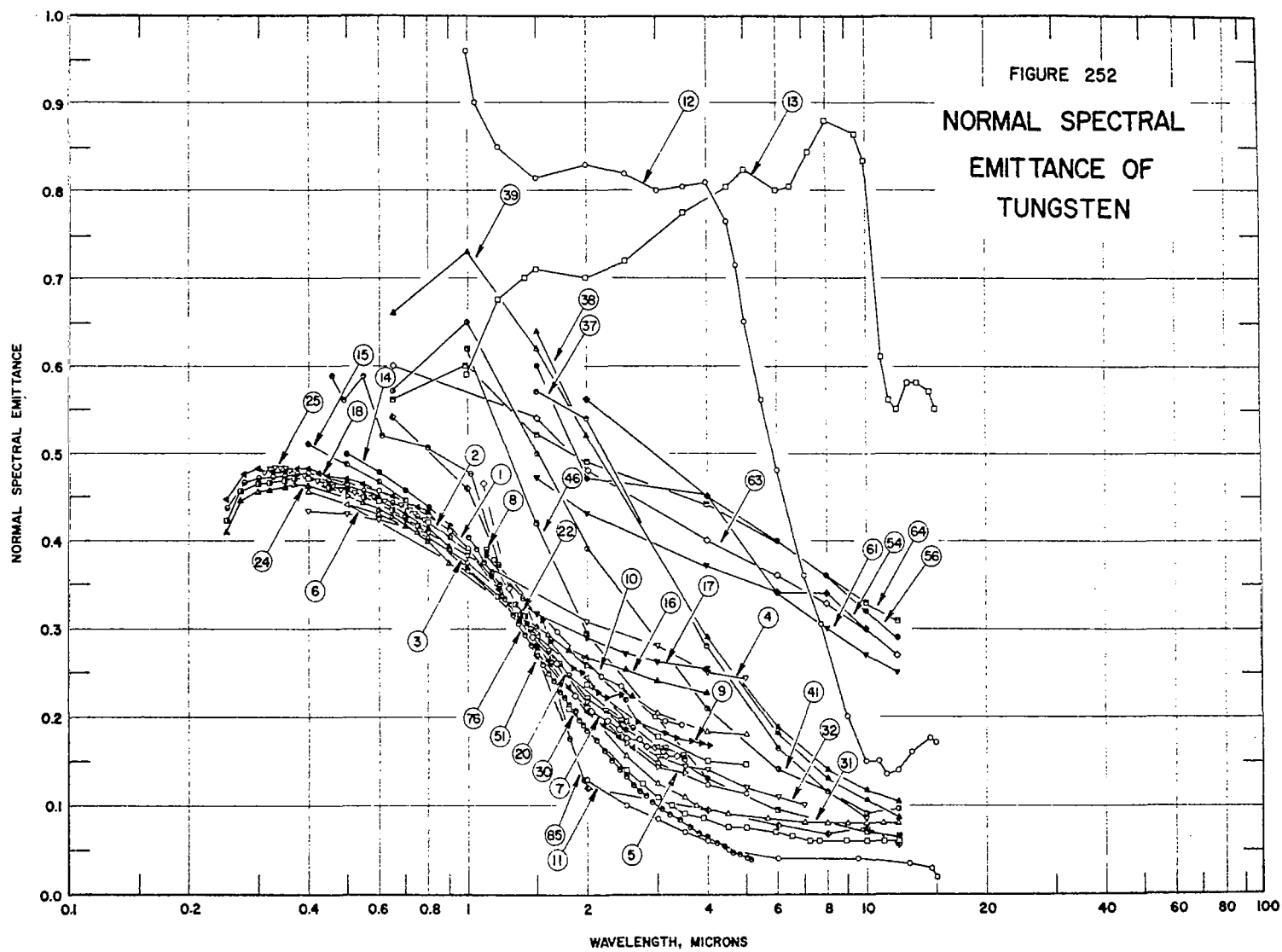
* Not shown on plot

DATA TABLE NO. 132 (continued)

T	€	T	€	T	€	T	€
<u>CURVE 26 (cont.)</u>		<u>CURVE 31*</u>		<u>CURVE 35 (cont.)</u>		<u>CURVE 40</u>	
545.2	0.055	449.7	0.043	1135	0.109	1368	0.314
685.7	0.065	594.2	0.054	1100	0.104	1308	0.302
813.2	0.082	696.2	0.062			1266	0.290
936.2	0.101	784.2	0.069	<u>CURVE 36*</u>		1233	0.281
1094.2	0.126	913.7	0.081	1242	0.122	1204	0.280
1191.7	0.143	1052.9	0.098	1290	0.128	1180	0.269
1308.7	0.163	1177.2	0.112	1337	0.134	1149	0.264
1375.7	0.173	1318.2	0.135	1387	0.143	1125	0.260
		1405.2	0.146	1431	0.149	1093	0.253
<u>CURVE 27</u>		<u>CURVE 32*</u>		1481	0.156	1063	0.248
412	0.043	407.2	0.034	1540	0.165	1034	0.242
545	0.052	502.8	0.041	<u>CURVE 37*</u>		1004	0.235
675	0.068	590.2	0.055	1045	0.122	998	0.236
797	0.081	767.0	0.066	1083	0.125		
909	0.096	889.2	0.077	1140	0.131		
1007	0.111	1159.2	0.113	1201	0.135		
1131	0.129	1283.2	0.130	1258	0.139		
1255	0.148	1374.2	0.142	1311	0.143		
1373	0.172			1366	0.150		
<u>CURVE 28*</u>		<u>CURVE 33</u>		1475	0.158		
410	0.038	1002.4	0.130	1539	0.165		
553	0.050	1169.2	0.112	<u>CURVE 38*</u>			
646	0.061	<u>CURVE 34</u>		1535	0.167		
768	0.073	1054	0.118	1480	0.156		
892	0.088	1089	0.123	1429	0.150		
990	0.100	1145	0.128	1385	0.143		
1114	0.121	1209	0.132	1333	0.136		
1244	0.139	1263	0.136	1264	0.131		
<u>CURVE 29*</u>		1315	0.142	1237	0.124		
1378	0.163	1374	0.147	1184	0.117		
1273	0.144	1426	0.158	1131	0.110		
		1538	0.165	1097	0.105		
<u>CURVE 30</u>		<u>CURVE 35*</u>		<u>CURVE 39*</u>			
429.2	0.056	1539	0.165	1239	0.124		
520.2	0.063	1481	0.156	1286	0.130		
648.2	0.073	1431	0.149	1333	0.136		
763.2	0.082	1387	0.143	1385	0.144		
919.2	0.097	1335	0.135	1429	0.150		
1054.2	0.112	1290	0.128	1480	0.156		
1198.2	0.128	1243	0.122	1541	0.164		
1307.7	0.139	1188	0.115				
1401.2	0.147						

* Not shown on plot





SPECIFICATION TABLE NO. 252 NORMAL SPECTRAL EMITTANCE OF TUNGSTEN

Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ°	Reported Error, %	Composition (weight percent), Specifications and Remarks
1	112	1963	1605	0.4-5.0	$\sim 0^\circ$		Polished single crystal tungsten; measured in argon (1140 mm Hg) along 002 crystal plane; data extracted from smooth curve.
2	112	1963	2140	0.4-5.0	$\sim 0^\circ$		Above specimen and conditions.
3	112	1963	2639	0.4-5.0	$\sim 0^\circ$		Above specimen and conditions.
4	112	1963	2650	0.4-5.0	$\sim 0^\circ$		Above specimen and conditions.
5	113	1961	1830	0.5-3.5	$\sim 0^\circ$		Measured in argon (760 mm Hg); data extracted from smooth curve.
6	113	1961	2040	0.5-3.5	$\sim 0^\circ$		Above specimen and conditions.
7	113	1961	1316	1.10-3.47	$\sim 0^\circ$		Trace of surface oxidation observed; measured in vacuum.
8	113	1961	1340	1.12-3.48	$\sim 0^\circ$		Above specimen and conditions.
9	113	1961	1382	1.13-4.04	$\sim 0^\circ$		Above specimen and conditions except measured in argon (760 mm Hg).
10	113	1961	1429	1.16-3.48	$\sim 0^\circ$		Above specimen and conditions.
11	86	1961	523	2.00-15.00	$\sim 0^\circ$	± 5	As received; data extracted from smooth curve.
12	86	1961	773	1.00-15.00	$\sim 0^\circ$		Different sample, same as curve 11 specimen and conditions.
13	86	1961	1023	1.00-15.00	$\sim 0^\circ$	± 5	Different sample, same as curve 11 specimen and conditions.
14	95	1963	1600	0.50-4.00	$\sim 0^\circ$		99.9 W from Carbide Specialty Co.; polished with carbide paper of 240, 400, and 600 grit, respectively, and then with silk cloth and felt cloth; washed in acetone, then alcohol, and dried with dry nitrogen; data extracted from smooth curve.
15	95	1963	2000	0.40-4.00	$\sim 0^\circ$		Above specimen and conditions.
16	95	1963	2800	0.40-4.00	$\sim 0^\circ$		Above specimen and conditions.
17	95	1963	3100	0.40-4.00	$\sim 0^\circ$		Above specimen and conditions.
18	114	1954	1600	0.25-2.60	$\sim 0^\circ$	0.1	0.014-0.015 Fe, 0.004-0.008 Si, 0.001-0.003 Mn, 0.0003-0.0006 Mg, W balance; heated at 2400 K, treated in hydrogen, annealed at 2400 K for 100 hrs; measured in vacuum (5×10^{-6} mm Hg); data extracted from smooth curve.
19	114	1954	1800	0.25-2.60	$\sim 0^\circ$	0.1	Above specimen and conditions.
20	114	1954	2000	0.25-2.60	$\sim 0^\circ$	0.1	Above specimen and conditions.
21	114	1954	2200	0.25-2.60	$\sim 0^\circ$	0.1	Above specimen and conditions.
22	114	1954	2400	0.25-2.60	$\sim 0^\circ$	0.1	Above specimen and conditions.
23	114	1954	2600	0.25-2.60	$\sim 0^\circ$	0.1	Above specimen and conditions except measured in argon (500 mm Hg).
24	114	1954	2800	0.25-2.60	$\sim 0^\circ$	0.1	Above specimen and conditions.
25	115	1959	1600	0.310-0.800	$\sim 0^\circ$		Better than 99.99 percent pure; heated at 2750 K for 1/2 hr in vacuum, then annealed at 2500 K for 30 hrs, at 2800 K for 1/2 hr, and at 2500 K for 20 hrs; measured in vacuum (3×10^{-8} to 9.5×10^{-8} mm Hg).

SPECIFICATION TABLE NO. 252 (continued)

Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ°	Reported Error, %	Composition (weight percent), Specifications and Remarks
26	115	1959	1800	0.310-0.800	$\sim 0^\circ$		Above specimen and conditions.
27	115	1959	2000	0.310-0.800	$\sim 0^\circ$		Above specimen and conditions.
28	115	1959	2200	0.310-0.800	$\sim 0^\circ$		Above specimen and conditions.
29	115	1959	2400	0.310-0.800	$\sim 0^\circ$		Above specimen and conditions.
30	12	1962	1200	0.50-12.00	$\sim 0^\circ$	± 3	Measured in vacuum ($< 10^{-7}$ mm Hg).
31	12	1962	1428	0.50-12.00	$\sim 0^\circ$	± 3	Above specimen and conditions.
32	12	1962	1972	0.45-7.00	$\sim 0^\circ$	± 3	Above specimen and conditions.
33	241	1963	1660	1.10-1.70	0	< 1	Single crystal; oriented so that surface of interest coincided with closed packed plane; optically polished; heated at several hundred degrees above temperature of interest in 90 Ar + 10 H atm; computed from optical constants.
34	241	1963	1790	0.9-1.7	0°	< 1	Above specimen and conditions.
35	241	1963	1950	0.90-1.70	0°	< 1	Above specimen and conditions.
36	241	1963	2050	0.90-1.70	0°	< 1	Above specimen and conditions.
37	227	1964	693	1.5-10	$\sim 0^\circ$		Impurities < 40 ppm; grit blasted; surface roughness 17 microinches rms, 21 microinches rms after emittance test; preheated in vacuum at 1000 K for 0.5 hr; measured in vacuum (8×10^{-6} mm Hg); [Author's designation: Sample 2].
38	227	1964	860	1.5-12	$\sim 0^\circ$		Above specimen and conditions.
39	227	1964	1033	0.65-12	$\sim 0^\circ$		Above specimen and conditions.
40	227	1964	1200	0.65-12	$\sim 0^\circ$		Above specimen and conditions.
41	227	1964	1373	0.65-12	$\sim 0^\circ$		Above specimen and conditions.
42	227	1964	1603	0.65-4	$\sim 0^\circ$		Above specimen and conditions.
43	227	1964	698	1.5-10	$\sim 0^\circ$		Above specimen and conditions except second temperature cycle.
44	227	1964	860	1.5-10	$\sim 0^\circ$		Above specimen and conditions.
45	227	1964	1029	1-12	$\sim 0^\circ$		Above specimen and conditions.
46	227	1964	1196	1-12	$\sim 0^\circ$		Above specimen and conditions.
47	227	1964	1378	1.5-4	$\sim 0^\circ$		Above specimen and conditions.
48	227	1964	702	1.5-10	$\sim 0^\circ$		Impurities < 40 ppm; as received; surface roughness 1.5 microinches rms before emittance test, 2.4 microinches rms after emittance; preheated in vacuum for 2 hrs; measured in vacuum (7×10^{-6} mm Hg); [Author's designation: Sample 1].
49	227	1964	870	1.5-10	$\sim 0^\circ$		Above specimen and conditions.
50	227	1964	1039	1.5-12	$\sim 0^\circ$		Above specimen and conditions.

SPECIFICATION TABLE NO. 252 (continued)

Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ°	Reported Error, %	Composition (weight percent), Specifications and Remarks
51	227	1964	1208	0.65-12	$\sim 0^\circ$		Above specimen and conditions.
52	227	1964	1378	0.65-4	$\sim 0^\circ$		Above specimen and conditions.
53	227	1964	1603	0.65-4	$\sim 0^\circ$		Above specimen and conditions.
54	227	1962	692	2-10	$\sim 0^\circ$		Impurities <40 ppm; grit blasted; surface roughness 110 microinches rms before emittance test, 38 microinches after emittance test; preheated in vacuum at 1000 K for 0.5 hr; measured in vacuum (5×10^{-5} mm Hg); $\omega = 3.4 \times 10^{-4}$ sr.; [Author's designation: Sample 3].
55	227	1964	863	1.5-12	$\sim 0^\circ$		Above specimen and conditions.
56	227	1964	1040	1.5-12	$\sim 0^\circ$		Above specimen and conditions.
57	227	1964	1214	0.65-12	$\sim 0^\circ$		Above specimen and conditions.
58	227	1964	1383	0.65-12	$\sim 0^\circ$		Above specimen and conditions.
59	227	1964	699	1.5-10	$\sim 0^\circ$		Above specimen and conditions.
60	227	1964	529	1.5-10	$\sim 0^\circ$		Above specimen and conditions.
61	227	1964	700	1.5-12	$\sim 0^\circ$		Above specimen and conditions.
62	227	1964	858	1.5-12	$\sim 0^\circ$		Above specimen and conditions.
63	227	1964	1198	0.65-12	$\sim 0^\circ$		Above specimen and conditions.
64	227	1964	1358	0.65-12	$\sim 0^\circ$		Above specimen and conditions.
65	76	1962	1600	0.50-4.00	$\sim 0^\circ$		Prepared from micronized powder; hot pressed at >2273 K; sintered, polished, etched, then degassed by heating to ~ 973 K; measured in argon; data extracted from smooth curve.
66	76	1962	2000	0.40-4.00	$\sim 0^\circ$		Above specimen and conditions.
67	76	1962	2800	0.40-4.00	$\sim 0^\circ$		Above specimen and conditions.
68	76	1962	3100	0.40-4.00	$\sim 0^\circ$		Above specimen and conditions.
69	239	1959	1429	1.157-3.486	$\sim 0^\circ$	5	Highly polished; measured in argon.
70	239	1959	1382	1.127-4.038	$\sim 0^\circ$	5	Different sample, same as above specimen and conditions.
71	239	1959	1340	1.120-3.485	$\sim 0^\circ$	5	Different sample, same as above specimen and conditions.
72	239	1959	1316	1.096-3.459	$\sim 0^\circ$	5	Different sample, same as above specimen and conditions.
73	237	1965	1800	0.467-0.698	$\sim 0^\circ$		Chemically pure; measured in vacuum; authors assumed $\epsilon = 1 - \rho$ and computed ρ from optical constants.
74	237	1965	2150	0.467-0.698	$\sim 0^\circ$		Above specimen and conditions.
75	237	1965	2520	0.467-0.698	$\sim 0^\circ$		Above specimen and conditions.

SPECIFICATION TABLE NO. 252 (continued)

Curve No.	Ref. No.	Year	Temperature K	Wavelength Range, μ	Geometry θ°	Reported Error, %	Composition (weight percent), Specifications and Remarks
76	338	1965	1244	1.00-5.10	0°		Ribbon; black body (at 1336 K) used as reference standard.
77	338	1965	1339	1.00-5.10	0°		Above specimen and conditions.
78	338	1965	1413	1.00-5.10	0°		Above specimen and conditions.
79	338	1965	1629	1.00-5.10	0°		Above specimen and conditions.
80	338	1965	1833	1.00-5.10	0°		Above specimen and conditions.
81	338	1965	2002	1.00-5.10	0°		Above specimen and conditions.
82	338	1965	2160	1.00-5.10	0°		Above specimen and conditions.
83	338	1965	2327	1.00-5.10	0°		Above specimen and conditions.
84	338	1965	2441	1.00-5.10	0°		Above specimen and conditions.
85	323	1966	300	0.46-2.00	0°		Filament (0.25-0.32 mm in dia); baked for 1 hr at 798 K in vacuum, cooled, heated for 5-10 min in vacuum, and cooled; measured in argon (600 mm Hg); data calculated from optical constants.
86	323	1966	1100	0.45-2.00	0°		Above specimen and conditions.
87	323	1966	1500	0.46-2.00	0°		Above specimen and conditions.
88	323	1966	2000	0.46-2.00	0°		Above specimen and conditions.
89	323	1966	2500	0.46-2.00	0°		Above specimen and conditions.
90	331	1917	2143	0.3478-0.5641	$\sim 0^\circ$	5	Measured in nitrogen; data is mean of three or more measurements.
91	333	1962	1600	0.230-0.269	$\sim 0^\circ$		Tungsten ribbon; data extracted from smooth curve
92	333	1962	1800	0.229-0.270	$\sim 0^\circ$		Above specimen and conditions.
93	333	1962	2000	0.229-0.271	$\sim 0^\circ$		Above specimen and conditions.
94	333	1962	2200	0.231-0.268	$\sim 0^\circ$		Above specimen and conditions.
95	333	1962	2400	0.231-0.268	$\sim 0^\circ$		Above specimen and conditions.
96	333	1962	2600	0.231-0.269	$\sim 0^\circ$		Above specimen and conditions.
97	333	1962	2800	0.229-0.268	$\sim 0^\circ$		Above specimen and conditions.

DATA TABLE NO. 252 NORMAL SPECTRAL EMITTANCE OF TUNGSTEN

[Wavelength, λ , μ ; Emittance, ϵ ; Temperature, T, K]

λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ
<u>CURVE 1</u> T = 1605		<u>CURVE 4 (cont.)</u>		<u>CURVE 7 (cont.)</u>		<u>CURVE 10 (cont.)</u>		<u>CURVE 12 (cont.)</u>		<u>CURVE 14 (cont.)</u>		<u>CURVE 17</u> T = 3100		<u>CURVE 18 (cont.)</u>	
		2. 0	0. 308	3. 12	0. 156	2. 17	0. 246	11. 25	0. 135	1. 50	0. 280			2. 40	0. 176
0. 4	0. 478	3. 0	0. 270	3. 35	0. 155	2. 43	0. 235	12. 00	0. 140	2. 00	0. 208*	0. 40	0. 472*	2. 60	0. 164
0. 6	0. 455	4. 0	0. 250	3. 47	0. 150*	2. 93	0. 200	13. 00	0. 160	2. 50	0. 184	0. 50	0. 458		
0. 8	0. 427	5. 0	0. 243			3. 12	0. 195	14. 50	0. 175	3. 00	0. 163*	0. 60	0. 442*	<u>CURVE 19*</u> T = 1800	
0. 9	0. 410			<u>CURVE 8</u> T = 1340		3. 48	0. 190	15. 00	0. 170	4. 00	0. 130	0. 70	0. 426		
1. 0	0. 390	<u>CURVE 5</u> T = 1830										0. 80	0. 412*		
2. 0	0. 210					<u>CURVE 11</u> T = 523		<u>CURVE 13</u> T = 1023		<u>CURVE 15</u> T = 2000		0. 90	0. 396*	0. 25	0. 442
3. 0	0. 147			1. 12	0. 390							1. 00	0. 380*	0. 27	0. 471
4. 0	0. 123	0. 5	0. 450	1. 45	0. 300							1. 50	0. 318	0. 30	0. 478
5. 0	0. 113	1. 0	0. 380*	1. 70	0. 260	2. 00	0. 130	1. 00	0. 590	0. 40	0. 510	2. 00	0. 290	0. 32	0. 476
		1. 5	0. 295	2. 00	0. 220	2. 50	0. 100	1. 20	0. 675	0. 50	0. 488	2. 50	0. 272	0. 35	0. 476
<u>CURVE 2</u> T = 2140		2. 0	0. 218	2. 22	0. 208	3. 00	0. 085	1. 40	0. 700	0. 60	0. 466	3. 00	0. 262	0. 37	0. 479
		2. 5	0. 170	2. 42	0. 195	3. 50	0. 070	1. 50	0. 710	0. 70	0. 445	4. 00	0. 252	0. 40	0. 478
		3. 0	0. 146	2. 92	0. 165*	4. 00	0. 060	2. 00	0. 700	0. 80	0. 425*			0. 45	0. 470
0. 4	0. 470	3. 5	0. 138	3. 13	0. 164	4. 50	0. 050	2. 50	0. 720	0. 90	0. 404	<u>CURVE 18</u> T = 1600		0. 50	0. 466
0. 6	0. 445			3. 48	0. 156	6. 00	0. 040	3. 50	0. 775	1. 00	0. 381*			0. 55	0. 460
0. 8	0. 422	<u>CURVE 6</u> T = 2040				9. 50	0. 040	4. 50	0. 805	1. 42	0. 306*			0. 60	0. 452
1. 0	0. 387			<u>CURVE 9</u> T = 1382		12. 75	0. 035	5. 00	0. 825	1. 50	0. 282*	0. 25	0. 448	0. 65	0. 446
2. 0	0. 237					14. 50	0. 030	6. 00	0. 800	2. 00	0. 236*	0. 27	0. 476	0. 70	0. 440
3. 0	0. 177	0. 5	0. 440			15. 00	0. 020	6. 50	0. 805	2. 50	0. 220	0. 30	0. 482	0. 75	0. 434
4. 0	0. 150	1. 0	0. 365	1. 13	0. 382			7. 25	0. 845	3. 00	0. 204*	0. 32	0. 478	0. 80	0. 426
5. 0	0. 145	1. 5	0. 293	1. 55	0. 310	<u>CURVE 12</u> T = 773		8. 00	0. 880	4. 00	0. 180*	0. 35	0. 479	0. 90	0. 406
		2. 0	0. 244	1. 62	0. 285			9. 50	0. 865			0. 37	0. 482	1. 00	0. 386
<u>CURVE 3</u> T = 2639		2. 5	0. 195	1. 70	0. 263*			10. 00	0. 835	<u>CURVE 16</u> T = 2800		0. 40	0. 481	1. 20	0. 345
		3. 0	0. 165	1. 85	0. 255	1. 00	0. 960	11. 00	0. 610			0. 42	0. 478	1. 27	0. 328
		3. 5	0. 150	1. 94	0. 250	1. 05	0. 900	11. 50	0. 560			0. 45	0. 474	1. 35	0. 312
0. 4	0. 455			2. 14	0. 228	1. 20	0. 850	12. 00	0. 550	0. 40	0. 478*	0. 50	0. 469	1. 50	0. 284
0. 6	0. 430	<u>CURVE 7</u> T = 1316		2. 25	0. 221	1. 50	0. 815	12. 75	0. 580	0. 50	0. 467	0. 55	0. 464	1. 60	0. 269
0. 8	0. 403			2. 43	0. 225	2. 00	0. 830	13. 50	0. 580	0. 60	0. 450	0. 60	0. 456*	1. 80	0. 242
1. 0	0. 375			2. 68	0. 193	2. 50	0. 820	14. 50	0. 570	0. 70	0. 430	0. 65	0. 450	2. 40	0. 187
2. 0	0. 258	1. 10	0. 465	2. 92	0. 185	3. 00	0. 800	15. 00	0. 550	0. 80	0. 412	0. 70	0. 444*	2. 60	0. 175
3. 0	0. 205	1. 27	0. 345	3. 12	0. 183	3. 50	0. 805			0. 90	0. 393	0. 75	0. 438		
4. 0	0. 183	1. 47	0. 290	3. 33	0. 175	4. 00	0. 810	<u>CURVE 14</u> T = 1600		1. 00	0. 367	0. 80	0. 432	<u>CURVE 20</u> T = 2000	
5. 0	0. 180	1. 67	0. 260	3. 62	0. 172	4. 50	0. 765			1. 42	0. 306*	0. 90	0. 412*		
		1. 86	0. 225	3. 84	0. 170	4. 75	0. 715			1. 50	0. 294*	1. 00	0. 390*		
<u>CURVE 4</u> T = 2650		2. 05	0. 208	4. 04	0. 169	5. 00	0. 650	0. 50	0. 500	2. 00	0. 267	1. 20	0. 344	0. 25	0. 436
		2. 24	0. 195			5. 50	0. 560	0. 60	0. 478	2. 50	0. 252	1. 27	0. 328	0. 27	0. 466
		2. 40	0. 185	<u>CURVE 10</u> T = 1429		6. 00	0. 480	0. 70	0. 457	3. 00	0. 240	1. 35	0. 310	0. 30	0. 470
0. 4	0. 433	2. 50	0. 175			7. 00	0. 360	0. 80	0. 437	4. 00	0. 228	1. 50	0. 280*	0. 32	0. 472
0. 5	0. 430	2. 70	0. 174			7. 75	0. 305	0. 90	0. 415			1. 60	0. 263	0. 35	0. 473
0. 6	0. 425	2. 82	0. 167	1. 16	0. 373	9. 00	0. 200	1. 00	0. 390*			1. 80	0. 234	0. 37	0. 476
0. 8	0. 405	2. 97	0. 165	1. 68	0. 299	10. 00	0. 150	1. 15	0. 360			2. 00	0. 210*	0. 40	0. 474*
1. 0	0. 383	3. 10	0. 160	1. 99	0. 263	10. 75	0. 150	1. 42	0. 306			2. 20	0. 190	0. 45	0. 467

* Not shown on plot

DATA TABLE NO. 252 (continued)

λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ
<u>CURVE 20 (cont.)</u> T = 2000		<u>CURVE 22</u> T = 2400		<u>CURVE 23 (cont.)</u>		<u>CURVE 25 (cont.)</u>		<u>CURVE 26 (cont.)</u>		<u>CURVE 27 (cont.)</u>		<u>CURVE 29 (cont.)</u>		<u>CURVE 30 (cont.)</u>	
0.50	0.462	0.25	0.422	1.00	0.360	0.340	0.481	0.520	0.453	0.740	0.422	0.340	0.472	6.45	0.065
0.55	0.456	0.27	0.456	1.20	0.339	0.360	0.480	0.540	0.451	0.760	0.420	0.350	0.472	7.20	0.060
0.60	0.448*	0.30	0.465	1.32	0.328	0.370	0.479*	0.460	0.449	0.780	0.418	0.360	0.471	7.60	0.060
0.65	0.442	0.32	0.465	1.40	0.317	0.380	0.477*	0.580	0.447	0.800	0.416	0.370	0.470	8.90	0.060
0.75	0.428	0.35	0.467	1.50	0.299	0.390	0.475	0.600	0.444	<u>CURVE 28*</u> T = 2200		0.380	0.469	10.00	0.060
0.80	0.420*	0.37	0.470	1.60	0.288	0.400	0.473*	0.620	0.441			0.390	0.467	11.00	0.060
0.90	0.400*	0.40	0.468*	1.80	0.268	0.420	0.469	0.640	0.438			0.400	0.466	12.00	0.060
1.00	0.382*	0.45	0.460	2.40	0.224	0.440	0.465	0.660	0.436			0.420	0.463		
1.20	0.342*	0.50	0.455*	2.60	0.214	0.460	0.462	0.680	0.435	0.310	0.471	0.440	0.459	<u>CURVE 31</u> T = 1428	
1.27	0.328*	0.55	0.450	<u>CURVE 24</u> T = 2800		0.480	0.459	0.700	0.433	0.320	0.473	0.460	0.456		
1.35	0.313*	0.60	0.440*			0.500	0.457*	0.720	0.429	0.330	0.474	0.480	0.452		
1.50	0.288*	0.65	0.434			0.520	0.455	0.740	0.426	0.340	0.474	0.500	0.449	0.50	0.456*
1.60	0.273	0.70	0.428*	0.25	0.410	0.540	0.453*	0.760	0.423	0.350	0.474	0.520	0.446	0.60	0.445*
1.80	0.247	0.75	0.418	0.27	0.445	0.560	0.452	0.780	0.421	0.360	0.473	0.540	0.443	0.70	0.428*
2.40	0.196*	0.80	0.409*	0.30	0.456	0.580	0.450	0.800	0.419	0.370	0.472	0.560	0.441	1.00	0.380*
2.60	0.187	0.90	0.396*	0.32	0.457	0.600	0.447	<u>CURVE 27*</u> T = 2000		0.380	0.471	0.580	0.437	1.50	0.275
<u>CURVE 21*</u> T = 2200		0.90	0.373*	0.35	0.461	0.620	0.445			0.390	0.469	0.600	0.434	2.00	0.225*
		1.00	0.339	0.37	0.463	0.640	0.442*			0.400	0.468	0.620	0.430	2.50	0.155
		1.20	0.328	0.40	0.461	0.660	0.441			0.420	0.464	0.640	0.426	3.00	0.125
0.25	0.430	1.32	0.313	0.40	0.454*	0.680	0.440*	0.310	0.474	0.440	0.461	0.660	0.424	3.50	0.110
0.27	0.460	1.40	0.288*	0.45	0.448*	0.700	0.437	0.320	0.476	0.460	0.457	0.680	0.421	3.75	0.100
0.30	0.470	1.50	0.273*	0.50	0.443	0.720	0.434	0.330	0.477	0.480	0.454	0.700	0.419	4.50	0.090
0.32	0.468	1.60	0.247*	0.55	0.443	0.740	0.430	0.340	0.477	0.500	0.451	0.720	0.417	5.65	0.085
0.35	0.470	1.80	0.196*	0.60	0.434	0.760	0.427*	0.350	0.476	0.520	0.448	0.740	0.415	7.00	0.080
0.37	0.473	2.40	0.185*	0.65	0.427	0.780	0.424	0.360	0.475	0.540	0.446	0.760	0.413	8.00	0.080
0.40	0.470	2.60		0.70	0.419	0.800	0.422*	0.370	0.474	0.560	0.443	0.780	0.412	9.00	0.080
0.45	0.464	<u>CURVE 23*</u> T = 2600		0.75	0.410	<u>CURVE 26*</u> T = 1800		0.380	0.473	0.580	0.440	0.800	0.411	10.00	0.080
0.50	0.458			0.80	0.400			0.390	0.471	0.600	0.437	<u>CURVE 30</u> T = 1200		11.00	0.080
0.55	0.453			0.90	0.373			0.400	0.469	0.620	0.433			12.00	0.080
0.60	0.444	0.25	0.416	1.00	0.367*	0.310	0.476	0.420	0.466	0.640	0.430				
0.65	0.438	0.27	0.450	1.20	0.337*	0.320	0.479	0.440	0.462	0.660	0.428	<u>CURVE 32</u> T = 1972			
0.70	0.432	0.30	0.460	1.32	0.328*	0.330	0.480	0.460	0.459	0.680	0.426			0.50	0.465*
0.75	0.423	0.33	0.461	1.35	0.318	0.340	0.479	0.480	0.456	0.700	0.424			0.66	0.450*
0.80	0.414	0.35	0.466	1.50	0.302	0.350	0.479	0.500	0.453	0.720	0.421			0.70	0.445*
0.90	0.396	0.40	0.464	1.60	0.292	0.360	0.478	0.520	0.450	0.740	0.419	0.50	0.380*	0.50	0.446*
1.00	0.377	0.45	0.457	1.80	0.271	0.370	0.476	0.540	0.448	0.760	0.416	1.80	0.210	0.55	0.444*
1.20	0.340	0.50	0.451	2.40	0.233*	0.380	0.475	0.560	0.446	0.780	0.415	2.50	0.140	0.60	0.440*
1.32	0.328	0.55	0.446	2.60	0.224	0.390	0.473	0.580	0.443	0.800	0.413	2.75	0.125	0.65	0.433*
1.50	0.284	0.60	0.438	<u>CURVE 25</u> T = 1600		0.400	0.471	0.600	0.440	<u>CURVE 29*</u> T = 2400		3.00	0.110	0.70	0.423*
1.60	0.278	0.65	0.430			0.420	0.467	0.620	0.437			3.25	0.100	0.80	0.410*
1.80	0.255	0.70	0.423			0.440	0.463	0.640	0.434			3.50	0.090	1.50	0.275*
2.40	0.205	0.75	0.414			0.460	0.460	0.660	0.432			3.90	0.085	2.00	0.225*
2.60	0.195	0.80	0.404	0.310	0.479	0.480	0.457	0.680	0.430	0.310	0.468	4.50	0.075	2.50	0.175*
		0.90	0.388	0.320	0.482	0.500	0.455	0.700	0.428	0.320	0.471	5.00	0.075	3.00	0.155
				0.330	0.482			0.720	0.425	0.330	0.472	5.90	0.070	3.50	0.145

* Not shown on plot

DATA TABLE NO. 252 (continued)

λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ
CURVE 65* T = 1600		CURVE 68(cont.)* T = 3100		CURVE 71(cont.)* T = 1340		CURVE 75* T = 2520		CURVE 76(cont.) T = 1244		CURVE 77(cont.)* T = 1339		CURVE 78(cont.)* T = 1413		CURVE 79(cont.)* T = 1629	
0.50	0.495	1.50	0.320	2.420	0.193	0.467	0.454	4.00	0.063	3.80	0.076	3.60	0.089	3.40	0.114
0.70	0.450	2.00	0.291	2.919	0.165	0.499	0.452	4.20	0.058	4.00	0.071	3.80	0.083	3.60	0.107
0.90	0.410	3.00	0.260	3.121	0.164	0.548	0.446	4.40	0.053	4.20	0.066	4.00	0.077	3.80	0.100
1.15	0.360	4.00	0.240	3.485	0.159	0.578	0.441	4.60	0.049	4.40	0.061	4.20	0.072	4.00	0.094
1.50	0.290					0.654	0.429	4.80	0.045	4.60	0.057	4.40	0.067	4.20	0.088
1.72	0.240					0.698	0.423	5.00	0.041	4.80	0.053	4.60	0.062	4.40	0.083
2.00	0.210							5.10	0.039	5.00	0.048	4.80	0.058	4.60	0.078
2.10	0.200									5.10	0.047	5.00	0.054	4.80	0.074
2.50	0.185											5.10	0.052	5.00	0.070
4.00	0.128													5.10	0.068
CURVE 66* T = 2000		CURVE 69* T = 1429		CURVE 72* T = 1316		CURVE 76 T = 1244		CURVE 77* T = 1339		CURVE 78* T = 1413		CURVE 79* T = 1629		CURVE 80* T = 1833	
0.40	0.505	1.157	0.378	1.096	0.464	1.00	0.402	1.00	0.398	1.00	0.396	1.00	0.388	1.00	0.382
0.50	0.480	1.676	0.298	1.264	0.345	1.05	0.389	1.05	0.385	1.05	0.382	1.05	0.377	1.05	0.372
0.70	0.440	1.984	0.261	1.457	0.290	1.10	0.373	1.10	0.369	1.10	0.368	1.10	0.364	1.10	0.361
0.90	0.400	2.164	0.246	1.655	0.259	1.15	0.360	1.15	0.358	1.15	0.357	1.15	0.354	1.15	0.352
1.15	0.350	2.416	0.235	1.860	0.224	1.20	0.345	1.20	0.344	1.20	0.343	1.20	0.342	1.20	0.342
1.50	0.290	2.929	0.199	2.049	0.206	1.24	0.333	1.24	0.333	1.24	0.333	1.24	0.333	1.24	0.333
1.75	0.260	3.122	0.194	2.231	0.194	1.30	0.318	1.30	0.319	1.30	0.319	1.30	0.319	1.30	0.319
2.00	0.240	3.486	0.189	2.389	0.182	1.35	0.305	1.35	0.306	1.35	0.308	1.35	0.308	1.35	0.308
2.50	0.220			2.539	0.175	1.40	0.292	1.40	0.294	1.40	0.296	1.40	0.296	1.40	0.296
4.00	0.180			2.694	0.171	1.45	0.280	1.45	0.283	1.45	0.286	1.45	0.286	1.45	0.286
				2.823	0.167	1.50	0.269*	1.50	0.272	1.50	0.275	1.50	0.275	1.50	0.275
				2.970	0.165	1.55	0.259	1.55	0.262	1.55	0.265	1.55	0.265	1.55	0.265
				3.092	0.158	1.60	0.249	1.60	0.253	1.60	0.256	1.60	0.256	1.60	0.256
				3.220	0.154	1.65	0.240	1.65	0.244	1.65	0.248	1.65	0.248	1.65	0.248
				3.340	0.154	1.70	0.231	1.70	0.236	1.70	0.240	1.70	0.240	1.70	0.240
				3.459	0.150	1.75	0.222	1.75	0.227	1.75	0.232	1.75	0.232	1.75	0.232
						1.80	0.214	1.80	0.217	1.80	0.221	1.80	0.221	1.80	0.221
						1.85	0.206	1.85	0.212	1.85	0.217	1.85	0.217	1.85	0.217
						1.90	0.198	1.90	0.204	1.90	0.209	1.90	0.209	1.90	0.209
						1.95	0.191	1.95	0.200	1.95	0.203	1.95	0.203	1.95	0.203
						2.00	0.184	2.00	0.197	2.00	0.199	2.00	0.199	2.00	0.199
						2.10	0.172	2.10	0.194	2.10	0.197	2.10	0.197	2.10	0.197
						2.20	0.161	2.20	0.180	2.20	0.186	2.20	0.186	2.20	0.186
						2.30	0.150	2.30	0.169	2.30	0.175	2.30	0.175	2.30	0.175
						2.40	0.141	2.40	0.158	2.40	0.165	2.40	0.165	2.40	0.165
						2.50	0.133	2.50	0.149	2.50	0.156	2.50	0.156	2.50	0.156
						2.60	0.124	2.60	0.141	2.60	0.147	2.60	0.147	2.60	0.147
						2.70	0.117	2.70	0.132	2.70	0.139	2.70	0.139	2.70	0.139
						2.80	0.110	2.80	0.125	2.80	0.132	2.80	0.132	2.80	0.132
						2.90	0.104	2.90	0.118	2.90	0.125	2.90	0.125	2.90	0.125
						3.00	0.099	3.00	0.112	3.00	0.119	3.00	0.119	3.00	0.119
						3.20	0.089	3.20	0.106	3.20	0.114	3.20	0.114	3.20	0.114
						3.40	0.082	3.40	0.097	3.40	0.104	3.40	0.104	3.40	0.104
						3.60	0.075	3.60	0.090	3.60	0.096	3.60	0.096	3.60	0.096
						3.80	0.068	3.80	0.082	3.80	0.088	3.80	0.088	3.80	0.088
CURVE 67* T = 2800		CURVE 70* T = 1382		CURVE 73* T = 1800		CURVE 74* T = 2150									
0.40	0.480	1.127	0.382	0.467	0.470	0.467	0.463								
0.50	0.460	1.551	0.302	0.499	0.466	0.499	0.460								
0.70	0.425	1.615	0.286	0.548	0.460	0.548	0.453								
0.90	0.390	1.693	0.272	0.578	0.456	0.578	0.449								
1.21	0.330	1.846	0.255	0.654	0.445	0.654	0.437								
1.50	0.295	1.934	0.252	0.698	0.441	0.698	0.431								
2.00	0.260	2.135	0.229												
3.00	0.231	2.239	0.223												
4.00	0.220	2.422	0.224												
CURVE 68* T = 3100		CURVE 71* T = 1340													
0.40	0.470	1.120	0.386												
0.50	0.451	1.449	0.295												
0.70	0.420	1.696	0.262												
0.90	0.390	2.002	0.223												
		2.204	0.208												

* Not shown on plot

DATA TABLE NO. 252 (continued)

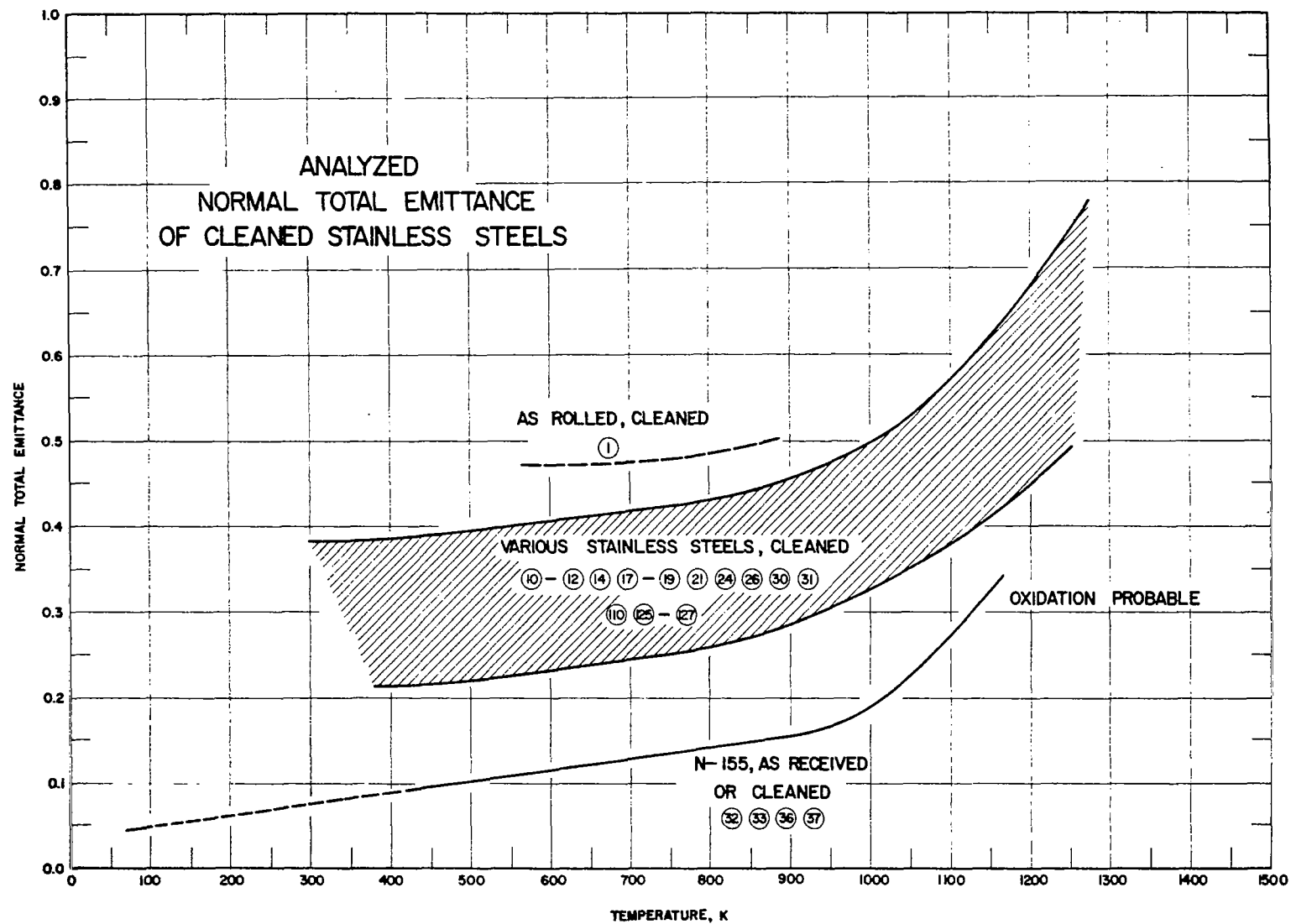
λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ	λ	ϵ
CURVE 65* T = 1600		CURVE 68(cont.)* T = 3100		CURVE 71(cont.)* T = 1340		CURVE 75* T = 2520		CURVE 76(cont.) T = 1244		CURVE 77(cont.)* T = 1339		CURVE 78(cont.)* T = 1413		CURVE 79(cont.)* T = 1629	
0.50	0.495	1.50	0.320	2.420	0.193	0.467	0.454	4.00	0.063	3.80	0.076	3.60	0.089	3.40	0.114
0.70	0.450	2.00	0.291	2.919	0.165	0.499	0.452	4.20	0.058	4.00	0.071	3.80	0.083	3.60	0.107
0.90	0.410	3.00	0.260	3.121	0.164	0.548	0.446	4.40	0.053	4.20	0.066	4.00	0.077	3.80	0.100
1.15	0.360	4.00	0.240	3.485	0.159	0.578	0.441	4.60	0.049	4.40	0.061	4.20	0.072	4.00	0.094
1.50	0.290					0.654	0.429	4.80	0.045	4.60	0.057	4.40	0.067	4.20	0.088
1.72	0.240					0.698	0.423	5.00	0.041	4.80	0.053	4.60	0.062	4.40	0.083
2.00	0.210	CURVE 69* T = 1429		CURVE 72* T = 1316				5.10	0.039	5.00	0.048	4.80	0.058	4.60	0.078
2.10	0.200					CURVE 76 T = 1244		CURVE 77* T = 1339		5.10	0.047	5.00	0.054	4.80	0.074
2.50	0.185	1.157	0.378	1.096	0.464							5.10	0.052	5.00	0.070
4.00	0.128	1.676	0.298	1.264	0.345									5.10	0.068
		1.984	0.261	1.457	0.290	1.00	0.402			CURVE 78* T = 1413		CURVE 79* T = 1629		CURVE 80* T = 1833	
CURVE 66* T = 2000		2.164	0.246	1.655	0.259	1.05	0.389	1.00	0.398	1.00	0.396	1.00	0.388	1.00	0.382
0.40	0.505	2.416	0.235	1.860	0.224	1.10	0.373	1.05	0.385	1.05	0.382	1.05	0.377	1.05	0.372
0.50	0.480	2.929	0.199	2.049	0.206	1.15	0.360	1.10	0.369	1.10	0.368	1.10	0.364	1.10	0.361
0.70	0.440	3.122	0.194	2.231	0.194	1.20	0.345	1.15	0.358	1.15	0.357	1.15	0.354	1.15	0.352
0.90	0.400	3.486	0.189	2.389	0.182	1.24	0.333	1.20	0.344	1.20	0.343	1.20	0.342	1.20	0.342
1.15	0.350	CURVE 70* T = 1382		2.539	0.175	1.30	0.318	1.24	0.333	1.24	0.333	1.24	0.333	1.24	0.333
1.50	0.290			2.694	0.171	1.35	0.305	1.30	0.319	1.30	0.319	1.30	0.311	1.30	0.322
1.75	0.260	1.127	0.382	2.823	0.167	1.40	0.292	1.35	0.306	1.35	0.306	1.35	0.301	1.35	0.313
2.00	0.240	1.551	0.302	2.970	0.165	1.45	0.280	1.40	0.294	1.40	0.294	1.40	0.292	1.40	0.304
2.50	0.220	1.615	0.286	3.092	0.158	1.50	0.269*	1.45	0.283	1.45	0.286	1.45	0.282	1.45	0.296
4.00	0.180	1.693	0.272	3.220	0.154	1.55	0.259	1.50	0.272	1.50	0.272	1.50	0.274	1.50	0.288
		1.846	0.255	3.340	0.154	1.60	0.249	1.55	0.262	1.55	0.275	1.55	0.274	1.55	0.280
CURVE 67* T = 2800		1.693	0.272	3.459	0.150	1.65	0.240	1.60	0.253	1.55	0.265	1.50	0.282	1.45	0.296
0.40	0.480	1.846	0.255	CURVE 73* T = 1800		1.70	0.231	1.65	0.244	1.60	0.256	1.55	0.274	1.50	0.288
0.50	0.460	1.934	0.252			1.75	0.222	1.70	0.236	1.65	0.248	1.60	0.265	1.55	0.280
0.70	0.425	2.135	0.229	0.467	0.470	1.80	0.214	1.75	0.227	1.70	0.240	1.65	0.257	1.60	0.273
0.90	0.390	2.239	0.223	0.499	0.466	1.85	0.206	1.80	0.219	1.75	0.232	1.70	0.250	1.65	0.266
1.21	0.330	2.422	0.224	0.548	0.460	1.90	0.198	1.85	0.212	1.80	0.224	1.75	0.242	1.70	0.259
1.50	0.295	2.685	0.191	0.578	0.456	1.95	0.191	1.90	0.204	1.85	0.217	1.80	0.236	1.75	0.252
2.00	0.260	2.919	0.184	0.654	0.445	2.00	0.184	1.95	0.197	1.90	0.209	1.85	0.229	1.80	0.245
3.00	0.231	3.128	0.181	0.698	0.441	2.10	0.172	2.00	0.194	1.95	0.203	1.90	0.222	1.85	0.239
4.00	0.220	3.335	0.173			2.20	0.161	2.10	0.180	2.00	0.197	1.95	0.217	1.90	0.233
		3.627	0.171	CURVE 74* T = 2150		2.30	0.150	2.20	0.169	2.10	0.186	2.00	0.210	1.95	0.227
CURVE 68* T = 3100		3.843	0.170			2.40	0.141	2.30	0.158	2.20	0.175	2.10	0.199	2.00	0.221
0.40	0.470	4.038	0.169			2.50	0.133	2.40	0.149	2.30	0.165	2.20	0.189	2.10	0.211
0.50	0.451	CURVE 71* T = 1340		0.467	0.463	2.60	0.124	2.50	0.141	2.40	0.156	2.30	0.180	2.20	0.201
0.70	0.420	1.120	0.386	0.499	0.460	2.70	0.117	2.60	0.132	2.50	0.147	2.40	0.171	2.30	0.192
0.90	0.390	1.449	0.295	0.548	0.453	2.80	0.110	2.70	0.125	2.60	0.139	2.50	0.163	2.40	0.184
		1.696	0.262	0.578	0.449	2.90	0.104	2.80	0.118	2.70	0.132	2.60	0.156	2.50	0.176
		2.002	0.223	0.654	0.437	3.00	0.099	2.90	0.112	2.80	0.125	2.70	0.150	2.60	0.169
		2.204	0.208	0.698	0.431	3.20	0.089	3.00	0.106	2.90	0.119	2.80	0.143	2.70	0.163
						3.40	0.082	3.20	0.097	3.00	0.114	2.90	0.137	2.80	0.157
						3.60	0.075	3.40	0.090	3.20	0.104	3.00	0.132	2.90	0.154
						3.80	0.068	3.60	0.082	3.40	0.096	3.20	0.122	3.00	0.146
														3.20	0.136

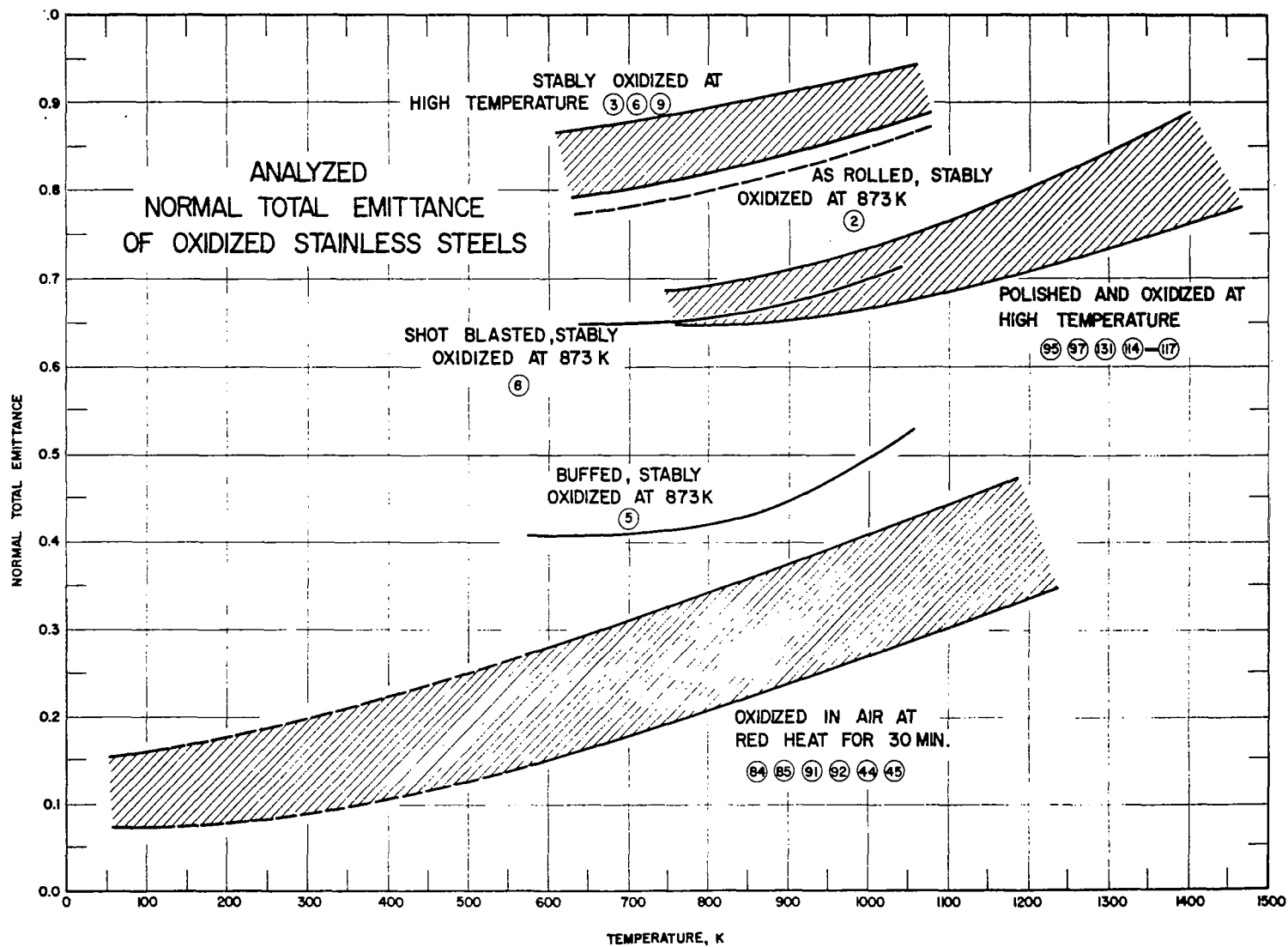
* Not shown on plot

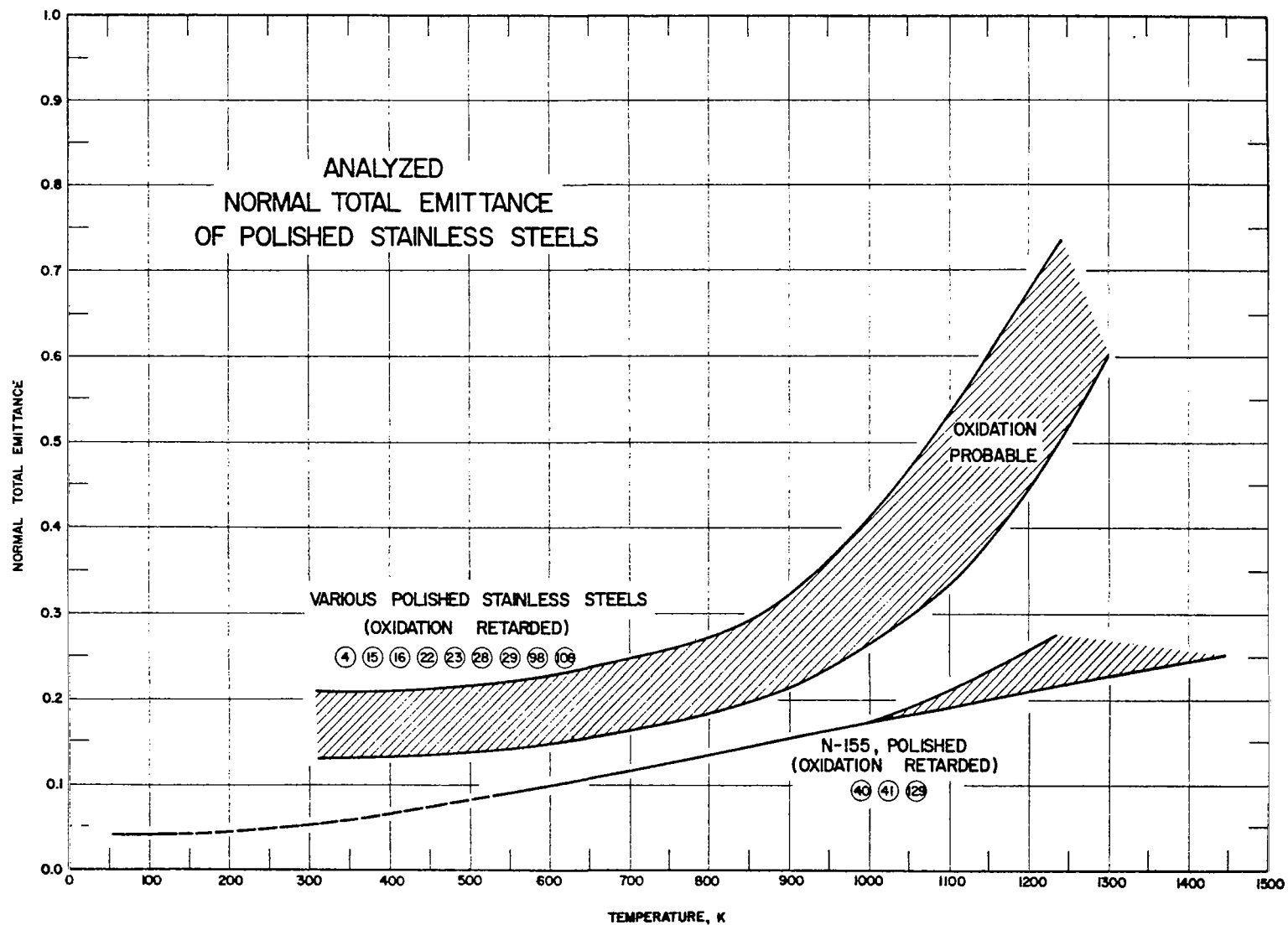
DATA TABLE NO. 252 (continued)

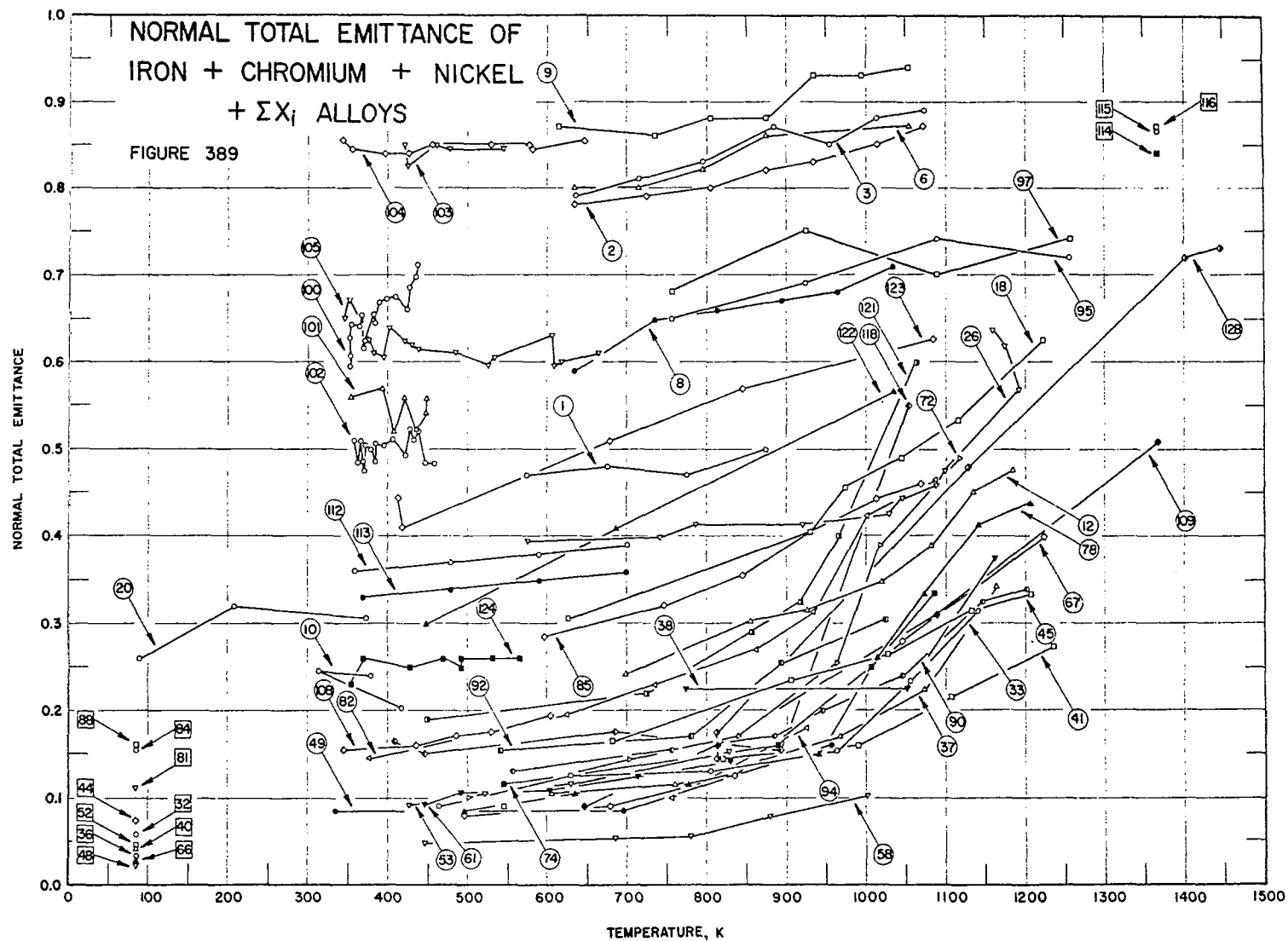
λ	ϵ	λ	ϵ
CURVE 94*		CURVE 96*	
T = 2200		T = 2600	
0.231	0.364	0.231	0.363
0.247	0.427	0.241	0.400
0.264	0.452	0.260	0.437
0.285	0.466	0.283	0.454
0.306	0.469	0.301	0.460
0.331	0.467	0.340	0.462
0.375	0.471	0.377	0.465
0.451	0.463	0.419	0.460
0.553	0.452	0.502	0.450
0.604	0.444	0.567	0.443
0.704	0.432	0.654	0.430
0.807	0.414	0.756	0.413
0.918	0.392	0.883	0.390
1.27	0.329	1.04	0.364
1.60	0.279	1.27	0.328
2.00	0.236	1.60	0.289
2.40	0.206	2.00	0.250
2.68	0.191	2.41	0.224
		2.69	0.210
CURVE 95*		CURVE 97*	
T = 2400		T = 2800	
0.231	0.363		
0.238	0.397	0.229	0.350
0.247	0.419	0.238	0.386
0.258	0.439	0.248	0.412
0.273	0.455	0.268	0.440
0.293	0.463	0.293	0.455
0.314	0.465	0.332	0.457
0.331	0.464	0.364	0.462
0.373	0.469	0.384	0.462
0.400	0.466	0.450	0.453
0.451	0.459	0.554	0.443
0.526	0.452	0.604	0.433
0.562	0.448	0.704	0.420
0.608	0.440	0.822	0.397
0.706	0.428	0.979	0.371
0.820	0.407	1.11	0.350
0.942	0.384	1.27	0.328
1.14	0.350	1.51	0.302
1.27	0.329	1.85	0.270
1.55	0.290	2.21	0.245
1.83	0.259	2.68	0.220
2.21	0.228		
2.68	0.201		

* Not shown on plot









SPECIFICATION TABLE NO. 389 NORMAL TOTAL EMITTANCE OF [IRON + CHROMIUM + NICKEL + ΣX_i] ALLOYS

Curve No.	Ref. No.	Year	Temperature Range, K	Geometry θ	Reported Error, %	Composition (weight percent), Specifications and Remarks
1	68	1952	573-873	$\sim 0^\circ$		Stainless steel Vickers F.D.P.; nominal composition: 18 Cr, 8 Ni, Fe balance; as rolled; cleaned with CCl_4 ; measured in air.
2	68	1952	633-1073	$\sim 0^\circ$		Different sample, same as curve 1 specimen and conditions; oxidized at 873 K until steady state reached.
3	68	1952	633-1073	$\sim 0^\circ$		Different sample, same as curve 1 specimen and conditions except oxidized at 1173 K until steady state reached.
4	68	1952	663-893	$\sim 0^\circ$		Different sample, same as curve 1 specimen and conditions except buffed.
5	68	1952	573-1053	$\sim 0^\circ$		Different sample, same as curve 4 specimen and conditions; oxidized at 873 K until steady state reached.
6	68	1952	633-1053	$\sim 0^\circ$		Different sample, same as curve 4 specimen and conditions except oxidized at 1173 K until steady state reached.
7	68	1952	583-873	$\sim 0^\circ$		Different sample, same as curve 1 specimen and conditions except shot blasted with fused alumina.
8	68	1952	633-1033	$\sim 0^\circ$		Different sample, same as curve 7 specimen and conditions; oxidized at 873 K until steady state reached.
9	68	1952	613-1053	$\sim 0^\circ$		Different sample, same as curve 7 specimen and conditions except oxidized at 1173 K until steady state reached.
10	160	1954	377-414	$\sim 0^\circ$		Stainless steel 301; nominal composition: 16.00-18.00 Cr, 6.00-8.00 Ni, 2.00 max Mn, 1.00 max Si, 0.15 max C, Fe balance; cleaned with methyl alcohol; measured in vacuum (10^{-3} mm Hg); [Author's designation: Sample 6].
11	160	1954	505-1280	$\sim 0^\circ$		Different sample, same as above specimen and conditions except measured in argon (10^{-3} mm Hg); [Author's designation: Sample 11].
12	160	1954	697-1183	$\sim 0^\circ$		Above specimen and conditions.
13	160	1954	88.9-364	$\sim 0^\circ$		Different sample, same as curve 10 specimen and conditions except scrubbed with Bon Ami on a wet cloth, washed and dried, wiped with toluene and alcohol; [Author's designation: Sample 2].
14	160	1954	439-1347	$\sim 0^\circ$		Different sample, same as above specimen and conditions except measured in argon (10^{-3} mm Hg); heating and cooling; [Author's designation: Sample 12].
15	160	1954	305-412	$\sim 0^\circ$		Different sample, same as curve 10 specimen and conditions except polished, then finished with a wool buff and rouge and washed; surface free from scratches; [Author's designation: Sample 4].
16	160	1954	469-1258	$\sim 0^\circ$		Different sample, same as above specimen and conditions except measured in argon (10^{-3} mm Hg); [Author's designation: Sample 13].
17	160	1954	310-423	$\sim 0^\circ$		Stainless steel 316; nominal composition: 16.00-18.00 Cr, 10.00-14.00 Ni, 2.00-3.00 Mo, 2.00 max Mn, 1.00 max Si, 0.08 max C; cleaned with methyl alcohol; measured in vacuum (10^{-3} mm Hg); heating; [Author's designation: Sample 1].

SPECIFICATION TABLE NO. 389 (continued)

Curve No.	Ref. No.	Year	Temperature Range, K	Geometry θ°	Reported Error, %	Composition (weight percent), Specifications and Remarks
18	160	1954	623-1221	$\sim 0^\circ$		Different sample, same as above specimen and conditions except measured in argon (10^{-3} mm Hg); [Author's designation: Sample 10].
19	160	1954	496-1219	$\sim 0^\circ$		Above specimen and conditions.
20	160	1954	88.9-372	$\sim 0^\circ$		Different sample, same as curve 17 specimen and conditions except scrubbed with Bon Ami on a wet cloth, washed and dried, wiped with toluene and alcohol; [Author's designation: Sample 2].
21	160	1954	517-1402	$\sim 0^\circ$		Different sample, same as above specimen and conditions except measured in argon (10^{-3} mm Hg); [Author's designation: Sample 11].
22	160	1954	314-432	$\sim 0^\circ$		Different sample, same as curve 17 specimen and conditions except polished and then finished with a wool buff and rouge and washed; surface free from scratches; [Author's designation: Sample 3].
23	160	1954	281-1417	$\sim 0^\circ$		Different sample, same as above specimen and conditions except measured in argon (10^{-3} mm Hg); [Author's designation: Sample 12].
24	160	1954	318-423	$\sim 0^\circ$		Stainless steel 347; nominal composition: 17.00-19.00 Cr, 9.00-13.00 Ni, 2.00 max Mn, 1.00 max Si, 0.08 max C, 10 x C min Nb-Ta, Fe balance; cleaned with methyl alcohol; measured in vacuum (10^{-3} mm Hg); heating; [Author's designation: Sample 7].
25	160	1954	319-423	$\sim 0^\circ$		Different sample, same as curve 24 specimen and conditions; [Author's designation: Sample 13].
26	160	1954	574-1191	$\sim 0^\circ$		Different sample, same as curve 24 specimen and conditions except measured in argon (10^{-3} mm Hg); [Author's designation: Sample 12].
27	160	1954	474-1192	$\sim 0^\circ$		Above specimen and conditions except heating and cooling.
28	160	1954	303-426	$\sim 0^\circ$		Different sample, same as curve 24 specimen and conditions except polished and then finished with a wool buff and rouge and washed; surface free from scratches; [Author's designation: Sample 6].
29	160	1954	488-1294	$\sim 0^\circ$		Different sample, same as curve 28 specimen and conditions except measured in argon (10^{-3} mm Hg); [Author's designation: Sample 15].
30	160	1954	303-434	$\sim 0^\circ$		Different sample, same as curve 24 specimen and conditions except scrubbed with Bon Ami on a wet cloth, washed and dried, wiped with toluene and alcohol; [Author's designation: Sample 5].
31	160	1954	493-1267	$\sim 0^\circ$		Different sample, same as curve 30 specimen and conditions except measured in argon (10^{-3} mm Hg); [Author's designation: Sample 14].
32	34	1957	83.2	$\sim 0^\circ$	± 10	Cobalt alloy N-155; nominal composition: 21 Cr, 20 Co, 20 Ni, 3 Mo, 3 W, 1.5 Mn, 1 Nb, 0.5 Si, 0.15 C, 0.15 N, Fe balance; as received; measured in vacuum (5×10^{-4} mm Hg).
33	34	1957	461-1139	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions; increasing temp, cycle 1.
34	34	1957	533-850	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.

SPECIFICATION TABLE NO. 389 (continued)

Curve No.	Ref. No.	Year	Temperature Range, K	Geometry g'	Reported Error, %	Composition (weight percent), Specifications and Remarks
35	34	1957	522	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
36	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 32 specimen and conditions except cleaned with liquid detergent.
37	34	1957	603-1161	$\sim 0^\circ$	$\bullet 10$	Different sample, same as above specimen and conditions; increasing temp, cycle 1.
38	34	1957	772-1161	$\sim 0^\circ$	$\bullet 10$	Above specimen and conditions; cycle 2.
39	34	1957	800-1003	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
40	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 32 specimen and conditions except polished.
41	34	1957	544-1233	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions; increasing temp, cycle 1.
42	34	1957	811-989	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
43	34	1957	911-1108	$\sim 0^\circ$	$\bullet 10$	Above specimen and conditions; cycle 3.
44	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 32 specimen and conditions except oxidized in air at red heat for 30 min.
45	34	1957	680-1205	$\sim 0^\circ$	$\bullet 10$	Different sample, same as above specimen and conditions; increasing temp, cycle 1.
46	34	1957	769-911	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
47	34	1957	611-1289	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
48	34	1957	83.2	$\sim 0^\circ$	$\bullet 10$	Stainless steel type PH 15-7 Mo; nominal composition: 15 Cr, 7 Ni, 2.25 Mo, 1.15 Al, 0.70 Mn, 0.40 Si, 0.07 C, Fe balance; surface roughness ~ 2 microinches rms; measured in vacuum (5×10^{-4} mm Hg).
49	34	1957	333-955	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions; increasing temp, cycle 1.
50	34	1957	661-866	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
51	34	1957	511-950	$\sim 0^\circ$	$\bullet 10$	Above specimen and conditions; cycle 3.
52	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 48 specimen and conditions; surface roughness ~ 15 microinches.
53	34	1957	425-828	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions; increasing temp, cycle 1.
54	34	1957	483-878	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
55	34	1957	625-844	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
56	34	1957	83.2	$\sim 0^\circ$	± 10	Stainless steel type 17-7 PH; nominal composition: 17 Cr, 7 Ni, 1.15 Al, 0.70 Mn, 0.40 Si, 0.07 C, Fe balance; surface roughness ~ 2 microinches rms; measured in vacuum (5×10^{-4} mm Hg).
57	34	1957	472-755	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
58	34	1957	444-1000	$\sim 0^\circ$	$\bullet 10$	Above specimen and conditions; cycle 2.
59	34	1957	650-933	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.

SPECIFICATION TABLE NO. 389 (continued)

Curve No.	Ref. No.	Year	Temperature Range, K	Geometry θ'	Reported Error, %	Composition (weight percent), Specifications and Remarks
60	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 56 specimen and conditions; surface roughness ~ 15 microinches rms.
61	34	1957	444-828	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
62	34	1957	661-1053	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
63	34	1957	605	$\sim 0^\circ$	± 10	Above specimen and conditions; decreasing temp, cycle 2.
64	34	1957	478-722	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
65	34	1957	616	$\sim 0^\circ$	± 10	Above specimen and conditions; decreasing temp, cycle 3.
66	34	1957	83.2	$\sim 0^\circ$	± 10	Stainless steel type 316; nominal composition: 16.00-18.00 Cr, 10.00-14.00 Ni, 2.00-3.00 Mo, 2.00 max Mn, 1.00 max Si, 0.08 max C, Fe balance; surface roughness ~ 2 microinches rms; measured in vacuum (5×10^{-4} mm Hg).
67	34	1957	494-1222	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
68	34	1957	505-1039	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
69	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 66 specimen and conditions; surface roughness ~ 15 microinches rms.
70	34	1957	444-628	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
71	34	1957	466-855	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
72	34	1957	500-1116	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
73	34	1957	83.2	$\sim 0^\circ$	± 10	Stainless steel type 321; nominal composition: 17.00-19.00 Cr, 9.00-12.00 Ni, 2.00 max Mn, 1.00 max Si, 0.08 max C, 5 x C min Ti, Fe balance; bright finish; measured in vacuum (5×10^{-4} mm Hg).
74	34	1957	544-1083	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
75	34	1957	622-894	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
76	34	1957	872-1122	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
77	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 73 specimen and conditions; surface roughness ~ 2 microinches rms.
78	34	1957	494-1205	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
79	34	1957	633-994	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
80	34	1957	728-1061	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
81	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 73 specimen and conditions except dull finish; surface roughness ~ 6 microinches rms.
82	34	1957	375-1089	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions.
83	34	1957	561-761	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.

SPECIFICATION TABLE NO. 389 (continued)

Curve No.	Ref. No.	Year	Temperature Range, K	Geometry θ'	Reported Error, %	Composition (weight percent), Specifications and Remarks
84	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 73 specimen and conditions except oxidized in air at red heat for 30 min; surface roughness ~ 6 microinches rms.
85	34	1957	594-1069	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
86	34	1957	544-789	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
87	34	1957	950	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 3.
88	34	1957	83.2	$\sim 0^\circ$	± 10	Stainless steel type AM 350; nominal composition: 16.50 Cr, 4.25 Ni, 2.75 Mo, 0.75 Mn, 0.35 Si, 0.10 C, 0.10 N, Fe balance; surface roughness ~ 2 microinches rms; measured in vacuum (5×10^{-4} mm Hg).
89	34	1957	422-605	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
90	34	1957	553-1200	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
91	34	1957	83.2	$\sim 0^\circ$	± 10	Different sample, same as curve 88 specimen and conditions except oxidized in air at red heat for 30 min.
92	34	1957	539-1022	$\sim 0^\circ$	± 10	Different sample, same as above specimen and conditions except increasing temp, cycle 1.
93	34	1957	900-1161	$\sim 0^\circ$	± 10	Above specimen and conditions; cycle 2.
94	170	1959	755-922	$\sim 0^\circ$		Stainless steel 321; titanium stabilized 18 Cr, 8 Ni austenitic stainless steel; electropolished; computed from spectral data.
95	170	1959	755-1255	$\sim 0^\circ$		Different sample, same as curve 94 specimen and conditions except also oxidized in air at 1255 K for 1/2 hr.
96	170	1959	755-922	$\sim 0^\circ$		Different sample, same as curve 94 specimen and conditions except sandblasted by 40-mesh glass sand and air at 40 psi.
97	170	1959	755-1255	$\sim 0^\circ$		Different sample, same as curve 94 specimen and conditions except also oxidized in air at 1255 K for 1/2 hr.
98	15	1947	373	$\sim 0^\circ$		Alleghany metal; nominal composition: 17-20 Cr, 7-10 Ni, 0.50 max Mn, 0.20 C, Fe balance; No. 4 polish.
99	157	1944	356-441	$\sim 0^\circ$		Stainless steel 18-8; nominal composition: 18.45 Cr, 8.79 Ni, 0.50 Mn, 0.10 C, Fe balance; oxidized at 811 K; measured in air.
100	157	1944	350-435	$\sim 0^\circ$		Different sample, same as curve 99 specimen and conditions except oxidized at 1089 K.
101	157	1944	351-446	$\sim 0^\circ$		Different sample, same as curve 99 specimen and conditions except chromic and sulfuric blackened.
102	157	1944	355-456	$\sim 0^\circ$		Different sample, same as curve 99 specimen and conditions except sand blasted.
103	155	1948	419-594	$\sim 0^\circ$		Stainless steel 18-8; nominal composition: 18.45 Cr, 8.79 Ni, 0.50 Mn, 0.10 C, Fe balance; sand blasted and weathered.
104	155	1948	342-646	$\sim 0^\circ$		Different sample, same as curve 103 specimen and conditions except oxidized at 1089 K and weathered.

SPECIFICATION TABLE NO. 389 (continued)

Curve No.	Ref. No.	Year	Temperature Range, K	Geometry θ'	Reported Error, %	Composition (weight percent), Specifications and Remarks
105	155	1948	344-661	$\sim 0^\circ$		Different sample, same as curve 103 specimen and conditions except chromic and sulphuric acid treated and weathered.
106	155	1948	455-650	$\sim 0^\circ$		Different sample, same as curve 103 specimen and conditions except unpolished.
107	155	1948	353-655	$\sim 0^\circ$		Different sample, same as curve 103 specimen and conditions.
108	155	1948	344-603	$\sim 0^\circ$		Above specimen and conditions except polished with Aerobright and Bon Ami.
109	107	1960	644-1644	$\sim 0^\circ$	± 20	Stainless steel 301; nominal composition: 16.00-18.00 Cr, 6.00-8.00 Ni, 2.00 Mn, 1.00 max Si, 0.15 max C, Fe balance; measured in demineralized helium gas.
110	99	1958	366-699	$\sim 0^\circ$	< 9	Type 321 corrosion-resistant steel; MIL-S-6721; nominal composition: 17.00-19.00 Cr, 9.00-12.00 Ni, 2.00 max Mn, 1.00 max Si, 0.08 max C, 5 x C min Ti, Fe balance; measured in air.
111	99	1958	366-699	$\sim 0^\circ$	< 9	Different sample, same as curve 110 specimen and conditions except heated at 647 K for 1000 hrs.
112	99	1958	366-699	$\sim 0^\circ$		Different sample, same as curve 110 specimen and conditions except calculated from spectral β ($2\pi, 0^\circ$).
113	99	1958	366-699	$\sim 0^\circ$		Different sample, same as curve 112 specimen and conditions except heated at 647 K for 1000 hrs.
114	164	1961	1366	$\sim 0^\circ$	± 2.5	Stainless steel type 303; nominal composition: 17.00-19.00 Cr, 8.00-10.00 Ni, 2.00 max Mn, 1.00 max Si, 0.15 min S, 0.15 max C, Fe balance; mechanically polished and cleaned; oxidized in quiescent air at 1366 K for 10 min; measured in air.
115	164	1961	1366	$\sim 0^\circ$	± 2.5	Above specimen and conditions except oxidized in quiescent air at 1366 K for 25 min.
116	164	1961	1366	$\sim 0^\circ$	± 2.5	Above specimen and conditions except oxidized in quiescent air at 1366 K for 40 min.
117	164	1961	1366	$\sim 0^\circ$	± 2.5	Above specimen and conditions except oxidized in quiescent air at 1366 K for 70 min.
118	40	1962	408-1061	$\sim 0^\circ$		Cobalt alloy N-155 (surface N-1); nominal composition: 21 Cr, 20 Co, 20 Ni, 3 Mo, 3 W, 1.5 Mn, 1 Nb, 0.5 Si, 0.15 C, 0.15 N, Fe balance; as received; increasing temp.
119	40	1962	408-655	$\sim 0^\circ$		Above specimen and conditions; decreasing temp.
120	40	1962	367	$\sim 0^\circ$		Different sample, same as curve 118 specimen and conditions except highly polished, mirror finish; oxide formation at 873 K for 3 hrs.
121	40	1962	447-1061	$\sim 0^\circ$		Different sample, same as curve 118 specimen and conditions except surface N-2; increasing temp.
122	40	1962	1033-447	$\sim 0^\circ$		Above specimen and conditions; decreasing temp.
123	40	1962	411-1084	$\sim 0^\circ$		Different sample, same as curve 121 specimen and conditions except heat treated; same results for increasing and decreasing temp.
124	40	1962	352-564	$\sim 0^\circ$		Poroloy (18-8 stainless steel); nominal composition: 18.45 Cr, 8.79 Ni, 0.50 Mn, 0.10 C, Fe balance; porosity between 28 and 31%.

SPECIFICATION TABLE NO. 389 (continued)

Curve No.	Ref. No.	Year	Temperature Range, K	Geometry θ°	Reported Error, %	Composition (weight percent), Specifications and Remarks
125	40	1962	367	$\sim 0^\circ$		Different sample, same as above specimen and conditions except porosity 28%.
126	40	1962	367	$\sim 0^\circ$		Different sample, same as above specimen and conditions except porosity 31%.
127	40	1962	367	$\sim 0^\circ$		Different sample, same as above specimen and conditions except porosity 43%.
128	75	1962	911-1444	$\sim 0^\circ$		Stainless steel 304; nominal composition: 18.00-20.00 Cr, 8.00-12.00 Ni, 2.00 max Mn, 1.00 max Si, 0.08 max C, Fe balance; machine finished; helium purge.
129	92	1963	1328-1466	$\sim 0^\circ$		Haynes Alloy N-155 (Multimet); 23.98-36.15 Fe, 19-21 Ni, 18.5-21 Co, 20-22.5 Cr, 2-3 W, 0.75-1.25 Nb and Ta, 2.5-3.5 Mo, 1.0-2.0 Mn, 0.5 max Cu, 1.0 max Si, 0.03 max S, 0.04 max P, 0.1-0.2 N ₂ , 0.08-0.16 C; polished; surface roughness 1 to 2 μ (RMS) measured with profilometer; measured in vacuum (3 to 4 x 10 ⁻⁴ mm Hg); 1st cycle.
130	92	1963	1289-1600	$\sim 0^\circ$		Above specimen and conditions; 2nd cycle.
131	92	1963	1239-1452	$\sim 0^\circ$		Curve 129 specimen and conditions except oxidized.
132	273	1962	805-1442	$\sim 0^\circ$		Stainless steel 304; nominal composition: 18.00-20.00 Cr, 8.00-12.00 Ni, 2.00 max Mn, 1.00 max Si, 0.08 max C, Fe balance; mechanical finish; measured in He gas.

DATA TABLE NO. 389 NORMAL TOTAL EMITTANCE OF [IRON + CHROMIUM + NICKEL + ΣX_i] ALLOYS[Temperature, T, K; Emittance, ϵ]

T	ϵ	T	ϵ	T	ϵ	T	ϵ	T	ϵ	T	ϵ	T	ϵ	T	ϵ
<u>CURVE 1</u>		<u>CURVE 6</u>		<u>CURVE 11^u(cont.)</u>		<u>CURVE 16^u</u>		<u>CURVE 21^u</u>		<u>CURVE 25^u</u>		<u>CURVE 29^u(cont.)</u>		<u>CURVE 36</u>	
573	0.47	633	0.85	1250	0.577	526	0.233	538	0.314	423	0.424	1183	0.636	83.2	0.033
673	0.48	713	0.85	507	0.557	707	0.212	792	0.312	319	0.463	547	0.497		
773	0.47	793	0.87	1218	0.547	908	0.261	929	0.330	359	0.424			<u>CURVE 37</u>	
873	0.50	873	0.86	505	0.573	1067	0.466	1051	0.396			<u>CURVE 30^u</u>		603	0.105
		1053	0.87			1129	0.536	1160	0.559	<u>CURVE 26</u>		434	0.324	758	0.115
<u>CURVE 2</u>		<u>CURVE 7^u</u>		<u>CURVE 12</u>		1186	0.640	1217	0.663	574	0.394	303	0.381	966	0.170
633	0.77			697	0.241	1239	0.729	1292	0.777	739	0.398	368	0.336	1072	0.225
723	0.79	583	0.40	852	0.301	1258	0.675	1321	0.657	785	0.414			1161	0.345
803	0.80	683	0.43	924	0.315	469	0.427	1397	0.431	919	0.413	<u>CURVE 31^u</u>			
873	0.82	733	0.44	1018	0.344	1178	0.708	1402	0.362	1029	0.426	493	0.367	<u>CURVE 38</u>	
933	0.83	783	0.45	1080	0.389	485	0.549	523	0.331	1045	0.444	661	0.385	772	0.225
1013	0.85	833	0.47	1133	0.451	1185	0.699	1164	0.410	1088	0.458	923	0.437	1050	0.225
1073	0.87	873	0.49	1183	0.477	483	0.502	1329	0.552	1098	0.474	1017	0.475	1161	0.375
						<u>CURVE 17^u</u>		533	0.529	1191	0.568	1111	0.629		
						423	0.247	1329	0.504	1173	0.620	1194	0.696	<u>CURVE 39^u</u>	
						310	0.280	517	0.517	1159	0.639	1247	0.717	800	0.270
						364	0.285					1267	0.705	1003	0.325
						<u>CURVE 18</u>		<u>CURVE 22^u</u>		<u>CURVE 27^u</u>		525	0.394	<u>CURVE 40</u>	
						623	0.306	432	0.169	525	0.394	548	0.485	83.2	0.041
						929	0.405	314	0.196	1192	0.495	1151	0.654	<u>CURVE 41</u>	
						971	0.457	371	0.176	1159	0.637	550	0.521	544	0.090
						1043	0.490			482	0.493	<u>CURVE 32</u>		819	0.145
						1114	0.532			1177	0.650	83.2	0.058	989	0.160
						1221	0.626			474	0.516			1105	0.215
						<u>CURVE 19^u</u>				<u>CURVE 28^u</u>		<u>CURVE 33</u>		1233	0.275
						496	0.329	679	0.174	426	0.150	461	0.090		
						786	0.319	893	0.203	303	0.169	628	0.125		
						1164	0.585	1029	0.283	370	0.173	803	0.130		
						1219	0.674	1127	0.362			961	0.155		
						510	0.572	1169	0.431	<u>CURVE 29^u</u>		1055	0.235	<u>CURVE 42^u</u>	
						1132	0.663	1214	0.594	532	0.211	1139	0.315	811	0.190
						504	0.581	1280	0.629	719	0.217	<u>CURVE 34^u</u>		989	0.240
						<u>CURVE 20</u>		1300	0.597	919	0.232	533	0.150	<u>CURVE 43^u</u>	
						88.9	0.260	1417	0.320	1046	0.281	850	0.125	911	0.230
						206	0.320	502	0.226	1170	0.412	<u>CURVE 35^u</u>		1108	0.280
						372	0.305	1147	0.282	1232	0.553	522	0.140	<u>CURVE 44</u>	
								1327	0.316	1290	0.624			83.2	0.072
								281	0.262	1294	0.656				
										488	0.388				
										1175	0.607				
										537	0.488				

* Not shown on plot

* Not shown on plot

DATA TABLE NO. 389 (continued)

T	€	T	€	T	€	T	€	T	€
<u>CURVE 101</u>		<u>CURVE 104 (cont.)</u>		<u>CURVE 107* (cont.)</u>		<u>CURVE 113</u>		<u>CURVE 122</u>	
351	0.560	528	0.850	563	0.203	366	0.33	1033	0.566
391	0.570	577	0.850	592	0.200	477	0.34	683	0.410
405	0.522	579	0.845	611	0.210	588	0.35	447	0.300
418	0.560	646	0.855	655	0.205	699	0.36		
433	0.524							<u>CURVE 123</u>	
445	0.542	<u>CURVE 105</u>		<u>CURVE 108</u>		<u>CURVE 114</u>		<u>CURVE 130*</u>	
446	0.560							411	0.445
<u>CURVE 102</u>		344	0.650	344	0.155	1366	0.841	416	0.410
355	0.510	350	0.670	433	0.160			677	0.510
360	0.484	375	0.625	483	0.170	<u>CURVE 115</u>		844	0.570
364	0.510	380	0.610	528	0.175			1084	0.628
366	0.486	392	0.605	603	0.195	1366	0.867		
369	0.476	397	0.610*			<u>CURVE 116</u>		<u>CURVE 124</u>	
370	0.504	400	0.640	<u>CURVE 109</u>				352	0.23
377	0.500	419	0.625	644	0.09	1366	0.870	358	0.26
383	0.486	428	0.620	811	0.16			425	0.25
383	0.508	433	0.620*	1089	0.31	<u>CURVE 117*</u>		425	0.26*
393	0.504	436	0.615	1367	0.51			469	0.26
395	0.506*	442	0.615*	1644	0.72*	1366	0.874	489	0.25
404	0.511	483	0.610			<u>CURVE 118</u>		490	0.26
419	0.492	525	0.595	<u>CURVE 110*</u>				491	0.26*
425	0.524	533	0.605	366	0.31	408	0.165	530	0.26
429	0.510	605	0.630	477	0.31	444	0.150	564	0.26
435	0.522	608	0.595	588	0.31	683	0.175	<u>CURVE 125*</u>	
438	0.518*	617	0.600	699	0.33	891	0.155		
444	0.484	661	0.610	<u>CURVE 111*</u>		1061	0.550	367	0.23
456	0.484	<u>CURVE 106*</u>				<u>CURVE 119*</u>		<u>CURVE 126*</u>	
<u>CURVE 103</u>		455	0.205	366	0.31				
419	0.850	472	0.170	477	0.33	408	0.225	367	0.23
422	0.825	542	0.225	588	0.31	655	0.338	<u>CURVE 127*</u>	
458	0.850	550	0.204	699	0.33	<u>CURVE 120*</u>			
475	0.845	583	0.210	<u>CURVE 112</u>		367	0.16	367	0.32
594	0.845	594	0.230			<u>CURVE 121</u>		<u>CURVE 128</u>	
<u>CURVE 104</u>		632	0.220	366	0.36			811	0.145
		650	0.225	477	0.37	447	0.190	811	0.175
342	0.855	<u>CURVE 107*</u>		588	0.38	722	0.220	1128	0.480
353	0.845	353	0.180	699	0.39	855	0.290	1400	0.720
394	0.840	380	0.170			916	0.325	1444	0.730
425	0.840	414	0.180			964	0.400		
453	0.850	503	0.190			1061	0.600		
		539	0.195						

* Not shown on plot

DATA ANALYSIS

The objective of the Analyzed Data Graphs is to give the user an evaluative review of available experimental data. It is quite apparent from a study of data sheets of the previous section that the analysis effort is first a filtering process; it identifies the data which are felt to be reliably or typically identified with the materials and gives the user a good deal of "relief" from the "spaghetti" type of presentation shown on the original or archival graphs. However, even these original graphs are the result of some filtering where grossly uncharacterized and second hand data sources have not been included. For certain circumstances of surface preparation and/or environmental conditions, data can be used with some degree of confidence but a great deal of data can only be considered as typical within certain limits.

The procedure for generating the Analyzed Data Graph varies according to the experimental evidence available for all the related sub-properties of the material being studied. Where there is some assurance that the data are well characterized and can be used with some confidence in engineering applications, the analyzed curve is shown as a solid line. A dotted line curve is used for reasonable extrapolations of well characterized conditions and for conditions felt to be typical and as such should be used with some caution. Shaded areas between solid lines are indicative of the limits in which so-called "typical" data will be found.

Each of the curves or areas is identified by key words which are felt descriptive of the surface conditions, etc.; frequently, these words are seen to be incomplete descriptions and the user will find it necessary, if not desirable in all cases, to consult the specification tables for a more complete description of the test conditions being represented. For this reason, the analyzed data graphs show the original curve number identifying the particular set(s) of data used to derive the analyzed curve.

In the SERIES Volume 7 nearly 25 percent of the data will be represented in analyzed or evaluated form. The sample figures presented in the previous section are some extreme examples selected to demonstrate the variety of approaches used to increase the value of the data to the user. The subsequent remarks on the data sheets of the previous section should give the reader of the

report some appreciation for the procedure/technique used in the data evaluation work.

ALUMINUM - For nearly all the materials, one of the first sets of data to be identified is the ideal surface conditions typified by the terms evaporated films for this material (and also electropolished for other materials to be discussed); the curves 4, 5, 15 and 29 when combined, and also interpolated in the near UV, give the reflectivity of well prepared "evaporated films". Bulk material polished by various techniques is identified by the broad band "polished" according to curves 9, 20 and 26; curve 17, the only visible data, also labeled "polished", is represented by a dashed line since it is uncharacterized data. The other curves on the same figure are all represented by solid lines because the surface conditions are characterized, could possibly be reproduced, and would be useful for many engineering applications.

COPPER - The figure for the normal spectral absorptance (room temperature) demonstrates the use of simply related sub-property data; in this case the results of analysis on the reflectance has been used to generate the absorptance in the visible region. The surface conditions - "electropolished & films" - for original measurements of both reflectance and absorptance have been evaluated to give a consistent recommended curve. The other curve on the figure labeled "mechanically polished, 6" demonstrates the influence of polishing techniques on the sub-property. Other data is shown on the original data sheet, but is seen to lack any particular value in demonstrating an effect or trend.

GALLIUM - The spectral transmittance data for this material has been presented in a hybrid manner; it is indicative of our philosophy to present the data based upon technical clarity. The measurements have been grouped into two figures distinguishing the high and low evaporation rate data. The curves are represented by the original data points and identified only by curve number. This data has not really been analyzed in depth but rather it has been filtered based upon some evaporation rate considerations. The reader has the obligation of labeling the curves as he sees fit and also of determining the reliability of the data which is not apparent without some other supporting data.

MOLYBDENUM - The analyzed data graph for the hemispherical total emittance demonstrates the effective use of bands indicative of limits of an effect, in this

case, for "polished" and "grit blasted conditions". It is assumed for all these curves an attempt is made to retard oxidation; this is the general case unless otherwise specified. The "stably oxidized at 811 K, 3" curve has a note warning the user about an extrapolation to higher temperatures in a vacuum; this note is not based upon data at hand, but rather reflects the experience of the analysis workers.

TUNGSTEN - This graph for the normal spectral emittance shows separately the very special case of "annealed aged ribbon" as commonly found in strip lamps or otherwise can be readily obtained and prepared; this data has its special uses and deserves to be distinguished from other types of specimens. The remaining data is analyzed in a manner consistent with the previous materials. It is especially interesting to note that the most significant information in 97 original data curves has been very concisely shown on one uncluttered graph.

STAINLESS STEEL - The original data graph for the normal total emittance contains 132 curves for measurements on the various alloys in extreme variations of surface conditions; it is understandable that the curves range in emittance from 0 to 1 and fill the whole graph. The first step in the analysis effort was to separate the data into three major groups: polished, cleaned and oxidized. It was quickly apparent that the influence of composition or alloy identification was lost by comparison to environmental effects; the exception to this in N-155. The remaining steps in the analysis procedure were to identify typical conditions.

SUMMARY

This report has described in some detail the problems and progress of a comprehensive program for the compilation and analysis of thermal radiative properties data. The procedures for compilation of the literature - an operation in scientific documentation and data extraction - draws upon the experience of TPRC as an information center*. The analysis work is an evaluative review of the literature attempting to filter out data thought to be of little value and to "recommend" data which is of engineering application use.

*Appendix E, TPRC-Information Center for Thermophysics Research, discusses in some detail the objective and activities of TPRC.

The results of this program will be available as the TPRC SERIES, a collection of three volumes distributed as a commercial publication. With an audience of diversified interests and backgrounds, the SERIES has been designed to be authoritative and yet simple to use; the need for this reference work has necessarily caused a compromise between availability and completeness of coverage. A special feature of this publication, quite distinctive from other data sources or handbooks, is the continuing program to upgrade or maintain its coverage current and the user can always get the "last word" from the SERIES generator, TPRC.

A program of this magnitude with such wide objectives can benefit from the reaction of the technical community. Reaction to the SERIES is very much desired whether it be technical in nature or in regard to the organization or structure. A technical problem of current concern is the Classification Scheme for Coatings. The program can also benefit from personal assistance of specialists in this field - as new papers or reports are generated they should be forwarded to TPRC for their immediate addition to the System; in this direct and simple manner, the coverage can be made current.

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APPENDIX A

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Thermal Radiative Properties
of
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APPENDIX B

Sub-property Nomenclature, Definitions, and Classification

TPRC Series, Volume 7
Thermal Radiative Properties
of
Metallic Elements and Alloys

1.0 Primary Property Definitions

The primary radiation properties -- emittance, reflectance, absorptance, and transmittance -- are all dimensionless quantities descriptive of the radiant energy transport process. They are defined as follows:

Emittance	ratio of the emitted flux per unit area to that of a blackbody radiator at the same temperature, and the same wavelength and geometric viewing conditions.
Reflectance	ratio of some specified portion of the reflected radiant flux to the incident radiant flux.
Absorptance	ratio of the absorbed radiant flux to the incident radiant flux.
Transmittance	ratio of some specified portion of the transmitted radiant flux to the incident radiant flux.

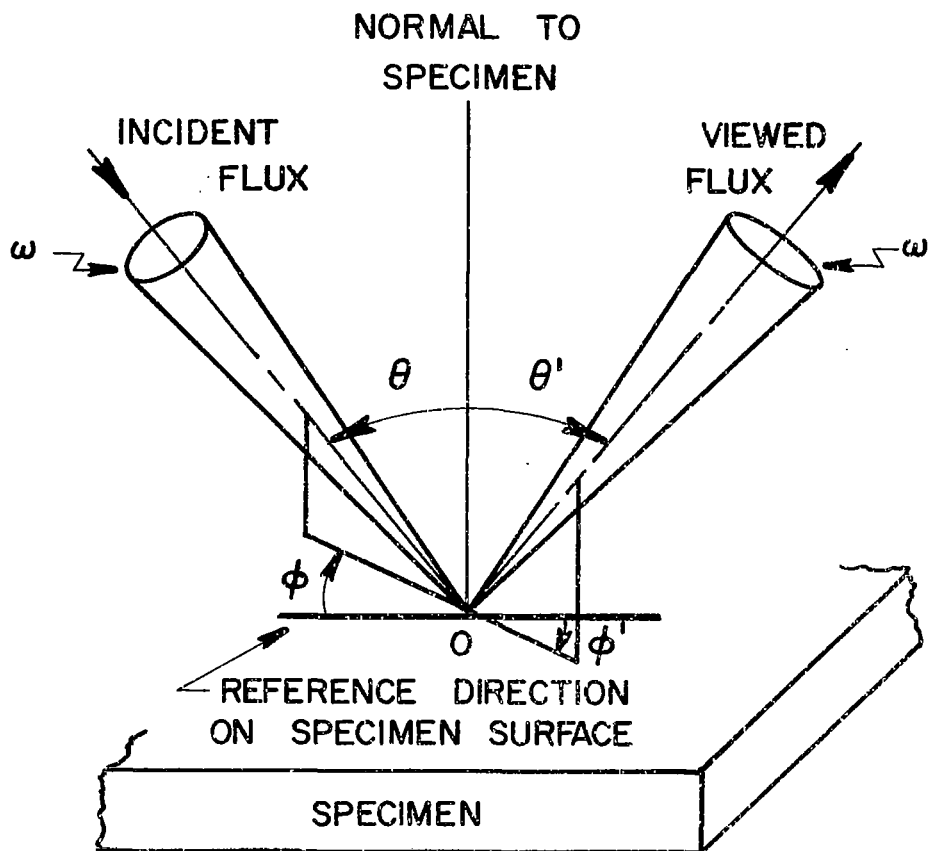
To present the data in a more concise form for efficient retrieval, the primary properties are further categorized into sub-properties. Since there is no universally accepted nomenclature for this division, it has been necessary to develop a consistent set of terms to unambiguously represent the various sub-properties. For the most part, the nomenclature, fully described in the next section, approximates common usage and lends itself well to the compact and systematic organization required of such comprehensive work.

2.0 Sub-Property Designation

Each of the primary properties needs to be further specified by descriptors indicating the geometry of the incident and/or viewing conditions, and the wavelength condition. It is the convention here to list these descriptors in the following order: geometry of incidence (or viewing) and wavelength condition. The terminology of these descriptors is now further discussed.

2.1 Geometry descriptors

These descriptors designate the geometric conditions of incidence and/or viewing (in that order) under which the sample is being observed. Figure B-1 defines the six (6) parameters required to completely specify the incidence and viewing conditions.



- θ Zenith Angle or Co-Latitude, $^{\circ}$
- ϕ Azimuthal Angle or Longitude, $^{\circ}$
- ω Beam Solid Angle, Steradians
- ' (Prime) Refers to Viewing Conditions

FIGURE B-1. NOTATION FOR DESCRIBING GEOMETRIC CONDITIONS

For the purpose of categorizing the various sub-properties, the following three terms are used as the geometric descriptors for all of the primary properties.

Angular	conditions for incidence and/or viewing through a solid angle (steradians) ω and/or ω' at some direction specified by θ or $\theta' \geq 15^\circ$.
Normal	conditions for incidence and/or viewing through a solid angle ω and/or ω' , nearly normal to the specimen (to be interpreted as θ or $\theta' < 15^\circ$).
Hemispherical	conditions for incidence and/or viewing the flux over a hemispherical region; beam geometry, ω or ω' indicated as 2π .

The selection of these descriptors is at best a compromise with standard practice and convenience. For emittance and absorptance, only one beam need be specified - viewing and incidence respectively. For reflectance and transmittance, the geometry of two beams must be specified. However, for convenience it is desirable to group the reflectance and transmittance data as follows: those sub-properties with common incidence geometric descriptors are grouped together.

The grouping of reflectance and transmittance sub-properties by common incidence conditions deserves some explanation. A grouping scheme is desirable to reduce the physical size of the book, and equally important, to bring together simply related sub-properties allowing the user to locate fragments of data that can be put together to generate information not found directly. For example, if one desires information on normal emittance or normal absorptance, it can easily be calculated from the following equation under certain conditions.

$$\epsilon(0^\circ) = \alpha(0^\circ) = 1 - \rho(0^\circ, 2\pi) - \tau(0^\circ, 2\pi)$$

If the sample is opaque, the equation simplifies to

$$\epsilon(0^\circ) = \alpha(0^\circ) = 1 - \rho(0^\circ, 2\pi)$$

and the calculation can be made by simply using normal reflectance.

"Reflectance factor" data is classified according to the sub-property of reflectance which it most nearly approximates. For instance, consider a reflectometer which hemispherically (diffusely), illuminates a specimen and views normally through a small solid angle. If the reflectometer then determines the ratio of the reflected flux from the specimen to that reflected from a perfect diffusing standard, the sub-property is the reflectance factor, $\beta(2\pi, 0^\circ)$. This is numerically equal to the reflectance sub-property $\rho(0^\circ, 2\pi)$. Such data is included in the "normal" category and a note added that it is in reality a reflectance factor.

2.2 Wavelength descriptors

These descriptors indicate spectrum conditions for which the observations are reported. They are:

Spectral	nearly monochromatic or a very narrow band
Total	relative to blackbody wavelength distribution; applicable only to emittance
Integrated	relative to some specified wavelength distribution of the irradiating source or a broad band
Solar	relative to the wavelength distribution of the sun, natural or simulated

The terms "spectral", "total" and "solar" are in common use and need little justification, the last term being a special case of "integrated", separately categorized because of the great current interest in solar property data.

The term "integrated" is a compromise as it has not been used extensively in the literature. The intent here is to group under this term data for broad wavelength bands, over spectral regions of a source, etc. A synonym for this term could be heterochromatic.

3.0 Sub-Property Groupings

The following Table B-1 lists the grouping of the various sub-properties that are presented in the book. This shows that thirty-three (33) sub-properties are classified for retrieval and organizational purposes. The amount of existing data for some of these sub-properties is quite small, but there are good

reasons to present the data using this generalized scheme. First, the clarity of presentation is better by not grouping together data which logically are unrelated. Also, this scheme lends itself especially well to up-dating and expansion in the future.

TABLE B-1. SUB-PROPERTY DESIGNATION

EMITTANCE

Hemispherical Total Emittance
Normal Total Emittance
Angular Total Emittance

Hemispherical Spectral Emittance
Normal Spectral Emittance
Angular Spectral Emittance

REFLECTANCE *

Hemispherical Integrated Reflectance
Normal Integrated Reflectance
Angular Integrated Reflectance

Hemispherical Spectral Reflectance
Normal Spectral Reflectance
Angular Spectral Reflectance

Hemispherical Solar Reflectance
Normal Solar Reflectance
Angular Solar Reflectance

ABSORPTANCE

Hemispherical Integrated Absorptance
Normal Integrated Absorptance
Angular Integrated Absorptance

Hemispherical Spectral Absorptance
Normal Spectral Absorptance
Angular Spectral Absorptance

Hemispherical Solar Absorptance
Normal Solar Absorptance
Angular Solar Absorptance

TRANSMITTANCE*

Hemispherical Integrated Transmittance
Normal Integrated Transmittance
Angular Integrated Transmittance

Hemispherical Spectral Transmittance
Normal Spectral Transmittance
Angular Spectral Transmittance

Hemispherical Solar Transmittance
Normal Solar Transmittance
Angular Solar Transmittance

*The geometry descriptors refer to the conditions of the incident radiant flux.

APPENDIX C

Material Index and Grouping of Materials and List of Figures and Tables

TPRC Series, Volume 7
Thermal Radiative Properties
of
Metallic Elements and Alloys

MATERIAL INDEX

	EMITTANCE						REFLECTANCE									ABSORPTANCE									TRANSMITTANCE									
	Total			Spectral			Integrated			Spectral			Solar			Integrated			Spectral			Solar			Integrated			Spectral			Solar			
Sub-property code	01	02	03	04	05	06	07	08	09	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34
Material Name	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Alpha/Epsilon
Alcoa No. 2 reflector plate	2 3	-	-	-	-	-	-	-	-	-	-	-	-	-	-	37 39	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Alleghany Alloy No. 66	-	1120	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Alleghany Metal	-	1160	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Aluminum	1	6	-	-	10	13	16	-	18	20	23	30 34	-	36	-	38	41	-	-	43	46	-	48	51	-	-	-	-	53	56	-	-	-	59
Aluminum 1075	-	-	-	-	-	-	-	-	-	21	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Aluminum 1100	-	8	-	-	-	14	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Aluminum foil	31 4	7	-	-	-	-	-	-	-	-	24 25	-	-	-	-	39	-	-	-	-	46	-	-	-	-	-	-	-	-	56	-	-	-	-
Aluminum alloys:																																		
24-ST	1004	1007 1008	-	-	-	-	-	1014	-	-	1018	-	-	-	-	-	-	-	-	-	-	-	1027	-	-	-	-	-	-	-	-	-	-	-
53-SO	-	1042	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
75-ST	-	1061	-	-	-	-	-	1070	-	-	-	-	-	-	-	-	-	-	-	-	-	-	1075	-	-	-	-	-	-	-	-	-	-	-
2024	1004	-	-	-	1011	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	1027	-	-	-	-	-	-	-	-	-	-	-	-
2024-T (see 24-ST)																																		

Index

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Index

Sub-property code	EMITTANCE						REFLECTANCE						ABSORPTANCE						TRANSMITTANCE						Alpha/Epsilon									
	Total			Spectral			Integrated			Spectral			Solar			Integrated			Spectral			Solar				Integrated			Spectral			Solar		
	01	02	03	04	05	06	07	08	09	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24		25	26	27	28	29	30	31	32	33
Material Name	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	
Aluminum bronze	-	1097	-	-	1100	-	-	-	-	-	1104	-	-	-	-	-	-	-	-	-	-	-	1107	-	-	-	-	-	-	-	-	-	-	
Aluminum + Cobalt	-	-	-	-	-	-	-	-	-	-	837	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Aluminum + Copper + ΣX _i	1003	1006	-	-	1010	-	-	1013	-	-	1017	-	-	1023	-	-	-	-	-	-	-	-	1026	-	-	-	-	-	-	-	-	-	-	
Aluminum + Iron + ΣX _i	-	-	-	-	-	-	-	-	-	-	1029	-	-	1032	-	-	-	-	-	-	-	-	1036	-	-	-	-	-	-	-	-	-	-	
Aluminum + Magnesium	-	-	-	-	-	-	-	-	-	-	840	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Aluminum + Magnesium + ΣX _i	1039	1041	-	-	-	-	-	1044	-	-	1047	-	-	-	-	-	-	-	-	-	-	-	1052	-	-	-	-	-	-	-	-	-	-	
Aluminum + Manganese + ΣX _i	1054	1056	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Aluminum + Silicon	-	-	-	-	-	-	-	-	-	-	843	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Aluminum + Silver	-	-	-	-	-	-	-	-	-	-	846	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Aluminum + Zinc + ΣX _i	1058	1060	-	-	1063	1066	-	1069	-	-	-	-	-	1072	-	-	-	-	-	-	-	-	1074	-	-	-	-	-	-	-	-	-	-	
Antimony	-	-	-	-	-	-	-	-	-	-	-	61	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	64	-	-	-	-	-	
Armco Iron	291	294	-	-	-	-	304	-	-	306	311	-	-	-	-	-	-	-	-	-	-	-	322	-	-	-	-	-	-	-	-	-	-	
Astrolloy	-	-	-	-	1303	1348	-	-	-	-	1318	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
AZ-31	-	1268	-	-	-	-	-	1271	-	-	1278	-	-	-	-	-	-	-	-	-	-	-	1280	-	-	-	-	-	-	-	-	-	-	
AZ-31B	-	-	-	-	-	-	-	-	-	1274	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Barium	-	-	-	-	-	-	-	-	-	-	-	66	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Beryllium	-	69	-	-	72	74	-	76	-	-	78	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	82	-	-	-	-	-	-	

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Sub-property code	EMITTANCE						REFLECTANCE						ABSORPTANCE						TRANSMITTANCE															
	Total		Spectral				Integrated		Spectral				Solar		Integrated		Spectral				Solar		Integrated		Spectral				Solar					
	01	02	03	04	05	06	07	08	09	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34
Material Name	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Alpha/Epsilon
Cobalt + Iron	-	-	-	-	854	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Cobalt + Nickel	-	-	-	-	856	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Cockron home foil	2	-	-	-	-	-	-	-	-	-	-	-	-	-	-	38	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
	3	-	-	-	-	-	-	-	-	-	-	-	-	-	-	39	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Copper	136	140	-	-	146	-	-	-	-	154	156	162	-	169	-	174	176	-	-	178	183	186	188	-	-	-	-	-	-	192	-	-	-	195
					149							164								180														
Copper, B.S. 1433	-	-	-	-	-	-	-	-	-	-	-	-	-	170	-	-	-	-	-	-	-	189	-	-	-	-	-	-	-	-	-	-	-	-
													171																					
Copper, OFHC	137	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	184	-	-	-	-	-	-	-	-	-	-	-	-	-
Copper + Aluminum + ΣX_1	-	1096	-	-	1099	-	-	-	-	-	1103	-	-	-	-	-	-	-	-	-	-	1106	-	-	-	-	-	-	-	-	-	-	-	-
Copper + Nickel	-	858	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Copper + Nickel + ΣX_1	-	-	-	-	-	-	-	-	-	-	1109	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Copper + Tin	-	-	-	-	-	-	-	-	-	-	860	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Copper + Tin + ΣX_1	-	-	-	-	-	-	-	-	-	-	1111	1113	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Copper + Zinc	862	864	-	-	-	-	-	-	-	-	867	870	-	-	-	873	875	-	-	-	877	-	-	-	-	-	-	-	-	-	-	-	-	-
Copper + Zinc + ΣX_1	-	-	-	-	-	-	1115	-	-	-	1117	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
D-43	1402	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Erbium	-	-	-	-	197	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Gadolinium	-	-	-	-	-	-	-	-	-	-	199	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	202	-	-	-	-	-
Gallium	-	-	-	-	-	-	-	-	-	-	205	-	-	-	-	-	-	-	-	208	-	-	-	-	-	-	-	-	211	-	-	-	-	-

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Sub-property code	EMITTANCE						REFLECTANCE						ABSORPTANCE						TRANSMITTANCE						Alpha/Epsilon	34									
	Total			Spectral			Integrated			Spectral			Solar			Integrated			Spectral			Solar					Integrated			Spectral			Solar		
	01	02	03	04	05	06	07	08	09	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24			25	26	27	28	29	30	31	32	33
Material Name	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular		
M252	-	-	-	-	1306	-	-	-	-	-	1323	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Magnesium	342	-	-	-	-	-	-	-	-	-	345	347	-	349	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Magnesium, L120	-	-	-	-	-	-	-	-	-	-	-	-	-	350 351	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Magnesium alloys:	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
AZ-31	-	1268	-	-	-	-	-	1271	-	-	1278	-	-	-	-	-	-	-	-	-	-	-	1280	-	-	-	-	-	-	-	-	-	-		
AZ-31B	-	-	-	-	-	-	-	-	-	-	1274	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
HK-31	-	1282	-	-	-	-	-	-	-	-	1284	-	-	-	-	-	-	-	-	-	-	-	1286	-	-	-	-	-	-	-	-	-	-		
Magnesium + Aluminum	-	-	-	-	-	-	-	-	-	-	898	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Magnesium + Aluminum + EX ₁	-	1267	-	-	-	-	-	1270	-	-	1273-1277	-	-	-	-	-	-	-	-	-	-	1280	-	-	-	-	-	-	-	-	-	-	-		
Magnesium + Thorium + EX ₁	-	1282	-	-	-	-	-	-	-	-	1284	-	-	-	-	-	-	-	-	-	-	-	1286	-	-	-	-	-	-	-	-	-	-		
Manganese	-	-	-	-	358	-	-	-	-	-	-	360	-	-	-	-	-	-	-	-	363	-	-	-	-	-	-	-	-	-	-	-	-		
Mild steel	1244	1246	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	1253	-	-	-	-	-	-	-	-	-	-	-		
Mo + 0.5 Ti alloy	-	902	905	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Molybdenum	365	370	-	-	374 379	-	-	-	-	-	384	388	-	-	-	-	-	-	-	390	393	-	396	-	-	-	-	-	-	-	-	-	-		
Molybdenum + Titanium	-	901	904	-	907	-	-	-	-	-	910	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Molybdenum + Tungsten	914	-	-	-	916	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Monel	1357	1360	-	-	-	-	-	-	-	-	1370	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		

[illegible]

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Sub-property code	EMITTANCE						REFLECTANCE						ABSORPTANCE						TRANSMITTANCE						Alpha/Epsilon										
	Total			Spectral			Integrated			Spectral			Solar			Integrated			Spectral			Solar													
	01	02	03	04	05	06	07	08	09	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24		25	26	27	28	29	30	31	32	33	
Material Name	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular		
Nickel - Iron - ΣX_i	-	-	-	-	1375	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Nickel + Molybdenum + ΣX_i	1378	1381	-	-	1384	-	-	-	-	-	1387	-	-	-	-	-	-	-	-	-	-	-	1390	-	-	-	-	-	-	-	-	-	-		
Nickel silver	-	-	-	-	-	-	1115	-	-	1117	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Nickel + Silver	-	925	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Nimonic 75	1290	1293	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Niobium	457	462	-	-	464 467	-	-	-	-	-	471	-	-	-	-	-	-	-	-	-	475	-	-	-	-	-	-	-	-	-	-	-	-		
Niobium + Molybdenum + ΣX_i	-	1393	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Niobium + Tantalum + ΣX_i	1395	-	-	-	1398	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Niobium + Tungsten	927	-	-	-	930	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Niobium + Tungsten + ΣX_i	1401	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Niobium + Zirconium	933	936	-	-	938	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
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Osmium	-	-	-	-	478	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
Palladium	480	482	-	-	485 488	-	-	-	-	-	490	-	-	-	-	-	-	-	-	-	493	-	496	-	-	-	-	-	498	-	-	-	-	-	
Phosphor bronze	-	-	-	-	-	-	-	-	-	1111	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Platinum	501	505	-	-	508 511	-	-	-	-	-	518 521 523	-	-	-	-	-	-	-	-	-	525 528	-	531	-	-	-	-	-	-	-	-	-	-	-	-

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	EMITTANCE						REFLECTANCE							ABSORPTANCE							TRANSMITTANCE														
	Total			Spectral			Integrated			Spectral			Solar			Integrated			Spectral			Solar			Integrated			Spectral			Solar				
Sub-property code	01	02	03	04	05	06	07	08	09	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	
Material Name	Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Hemispherical Normal Angular			Alpha/Epsilon	
Platinum, NBS	-	-	-	-	512	-	-	-	-	-	-	-	-	-	-	-	-	-	-	526	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Platinum wire, grade MPTU 4292-53	502	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Platinum + Rhodium	941	943	-	-	946	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
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Pt-10 Rn alloy	941	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Pt-13 Rh alloy	-	-	-	-	947	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Rene' 41	-	1296	-	-	1305 1306	-	-	-	-	-	1321 1323	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
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Rhodium	546	547	-	-	550 553	-	-	-	-	-	555 557	-	-	-	-	-	-	-	-	-	560	-	562	-	-	-	-	-	-	-	-	-	-	-	-
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Sub-property code	EMITTANCE						REFLECTANCE									ABSORPTANCE									TRANSMITTANCE									
	Total			Spectral			Integrated			Spectral			Solar			Integrated			Spectral			Solar			Integrated			Spectral			Solar			
	01	02	03	04	05	06	07	08	09	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34
Material Name	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Alpha/Epsilon
Stainless Steels (continued)																																		
430	-	-	-	-	-	-	1132	-	-	1135	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
431	-	-	-	-	-	-	1192	-	-	1200	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
446	-	1120	-	-	1125	-	1132	-	-	1135	1138	-	-	-	-	-	-	-	-	-	-	-	1147	-	-	-	-	-	-	-	-	-	-	-
A286	-	-	-	-	1262	-	-	-	-	-	1265	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
AM 350	-	1160	-	-	1171	-	-	-	-	-	1203	-	-	-	-	-	-	-	-	-	-	-	1235	-	-	-	-	-	-	-	-	-	-	-
					1177	-					1204	-										1236	-											
											1205	-																						
PH 15-7 Mo	-	1158	-	-	1170	-	-	-	-	-	1203	-	-	-	-	-	-	-	-	-	-	-	1234	-	-	-	-	-	-	-	-	-	-	-
					1177	-					1204	-																						
SF 11	-	-	-	-	-	-	-	-	-	-	886	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
SF 20	-	-	-	-	-	-	-	-	-	-	1203	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Stellite	-	-	-	-	-	-	-	-	-	-	1093	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
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Tellurium	-	-	-	-	-	-	-	-	-	-	-	652	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	655	-	-	-	-	-
Tellurium + Selenium	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	968	-	-	-	-	-	-

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	EMITTANCE						REFLECTANCE						ABSORPTANCE						TRANSMITTANCE															
	Total			Spectral			Integrated			Spectral			Solar			Integrated			Spectral			Solar			Integrated			Spectral			Solar			
Sub-property code	01	02	03	04	05	06	07	08	09	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34
	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Hemispherical	Normal	Angular	Alpha/Epsilon
Material Name																																		
Thallium	-	-	-	-	-	-	-	-	-	-	660	663	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	666	-	-	-	-	
Thorium	-	-	-	-	669	671	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Tin	673	675	-	-	-	-	-	-	-	-	677	-	-	-	-	680	682	-	684	687	-	-	-	-	-	-	-	-	690	-	-	-	-	
Tin + Indium	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	970	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
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469	Zirconium + Hafnium + ΣX_1	1445
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472	Zirconium + Uranium + ΣX_1	1453

APPENDIX D

References to Data Sources

TPRC Series, Volume 7
Thermal Radiative Properties
of
Metallic Elements and Alloys

REFERENCES TO DATA SOURCES

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APPENDIX E

TPRC - Information Center for
Thermophysics Research

1.0 Introduction

The overall activities of the Thermophysical Properties Research Center (TPRC) are divided into four areas: namely, (1) Scientific Documentation, (2) Generation of Data Tables, (3) Experimental Research, and (4) Theoretical Research. Since 1957, TPRC has contributed much to the knowledge of the thermophysical properties of pure and engineering materials [4]. Its results are disseminated at large in the form of two major publications: the Retrieval Guide and the Data Book.

Previously, the Retrieval Guide was published by McGraw-Hill Book Co., Volume 1 in 1960 and Volume 2 in 1963. Early in 1967, as Volume 3 of the Retrieval Guide was nearing completion, it was decided to publish a single comprehensive volume of this work thus merging the earlier Volumes 1 and 2 with the completed Volume 3. This merged, revised, and enlarged edition of the Retrieval Guide (THERMOPHYSICAL PROPERTIES RESEARCH LITERATURE RETRIEVAL GUIDE) was published in October 1967 by the Plenum Publishing Corporation. This definitive work contains a complete coverage of the world literature published from 1920 (in many cases earlier) to June 1964 on thirteen thermophysical properties. Its 2,800 pages (in 3 books) report 45,116 materials, citing 33,700 references representing 26,562 authors and 3,600 separate scientific and technical journals and books in addition to Government reports. Thus, the new Retrieval Guide brings to the scientific and technical community a single reference work heretofore thought impractical to generate. Effective with the completion of this work, the scope of coverage of the Retrieval Guide has been increased to sixteen properties. Furthermore, each property is coded separately instead of by groups.

The TPRC Data Book brought together the available data on the thermophysical properties of materials to provide for engineers and scientists the most comprehensive and authoritative reference data sources. Whenever possible, the recommended "most probable values" of particular properties for particular materials are also included. The original Data Book consisted of loose-leaf data sheets (11" x 17" in size) organized into three volumes. The data sheets constituted the final formal outlet of all TPRC data tables activities on all of its programs. As of December 1966, the Data Book contained 3,322 sheets, reporting 11,425 test specimens and citing 3,424 references.

In view of the continuing rapid growth of this work since 1960, and the extensive physical proportions it has assumed, early in 1967 it was decided to discontinue the procedure of publication in loose-leaf format, and its semi-annual dissemination by TPRC.

Instead, this Data Book is now restructured and extensively revised and will soon be available through a commercial publisher in the form of formal hard-bound volumes grouped by properties. This forthcoming publication is entitled TPRC SERIES ON THERMOPHYSICAL PROPERTIES OF MATTER.

The information processing and data processing for preparing the Retrieval Guide and the Data Book and the status of the various active projects are outlined briefly in the following sections. The activities of Data Tables Division pertaining to the generation of data tables on thermal radiative properties have been described in the text of this report.

2.0 Scientific Documentation

2.1 Literature Search and Information Processing

The Scientific Documentation Division of TPRC provides comprehensive and authoritative source information on the thermophysical properties of all matter through continuing and systematic search, collection, organization, and codification of the existing information in the world literature. From 1957 to 1964 TPRC has searched the world literature primarily through the abstracting journals. Since 1965, TPRC has subscribed to some 80 scientific and technical journals in addition to the abstracting journals, and accordingly, literature search has since been made on both of these two types of journals.

The abstracting journals searched from 1957 to 1964 are listed in Table E-1. This search, covering the period from 1920 to June 1964, involved the scanning of approximately 33,400,000 abstracts out of approximately 81,000,000 abstracts reported by these journals. Out of the 33,400,000 abstracts scanned, only 52,500 (0.16%) were considered pertinent. Subsequent examination and checking of the 52,500 abstracts and the original papers revealed a large number of duplications between the various abstracting journals and nearly 9,900 irrelevant ones, leaving a net total of 28,800 documents obtained from these sources. In addition to abstracts, 4,900 documents came to TPRC's attention from other sources making a total of 33,700

TABLE E-1. WORLD COVERAGE OF RESEARCH LITERATURE ON THERMOPHYSICAL
PROPERTIES THROUGH THE EYES OF 16 ABSTRACTING JOURNALS

(As of December 31, 1964)

ABSTRACTING JOURNALS	COVERAGE
Applied Mechanics Review (AM)	Jan. 1948 to June 1964
ASM Review of Current Metal Literature (in Metals Review) (MR)*	Jan. 1957 to Aug. 1959
ASM Review of Metal Literature (RM)	Jan. 1944 to Dec. 1955 and Jan. 1959 to June 1964
Battelle Technical Review (BR)	Feb. 1957 to June 1964
Ceramics Abstracts-Amer. Ceramic Soc. (JA)	Jan. 1957 to June 1964
Ceramics Abstracts-Brit. Ceramic Soc. (BA)	Jan. 1958 to June 1964
Chemical Abstracts (CA)	Jan. 1920 to June 1964
Masters Theses in the Pure and Applied Sciences - TPRC(MT)	1957 to 1964
Metallurgical Abstracts, Series II (MA)	Jan. 1934 to Aug. 1956 and Sept. 1958 to June 1964
Nuclear Science Abstracts (NA)	Jan. 1963 to June 1964
Physics Abstracts (SA)	Jan. 1957 to June 1964
Refrigeration Abstracts (RA)	Jan. 1946 to Oct. 1957
Scientific and Technical Aerospace Reports-NASA (PA)	Jan. 1957 to June 1964
Technical Abstract Bulletin-DDC (TA)*	Jan. 1957 to June 1964
Technical Translations-CFSTI (TT)*	Jan. 1957 to June 1964
U. S. Government Research Reports-CFSTI (RR)	Jan. 1957 to June 1964

*Ceased publication

documents up to June 1964. These 33,700 references are covered in the revised and expanded edition of the Retrieval Guide.

Subsequently, in preparation for the future volume of the Retrieval Guide, an additional 17,300 reference entries have been made. Thus, as of 31 December 1968 there were 51,000 references in TPRC's Information storage and Retrieval System. The above figure gives an insight as to the magnitude of the effort involved in a thorough search of world knowledge even in a relatively specialized field.

When the retrospective search of the world literature, primarily through the medium of abstracting journals, was completed early in 1956, TPRC reviewed its procedure of using abstracting journals for the identification of current literature on thermophysical properties research. It was recognized that continued use of abstracting journals for research awareness would represent, at best, one to two years of delay in identifying and procuring such literature, with the result that bibliographic searches provided by TPRC could never be on a reasonably current basis. A statistical study was made of the data accumulated at TPRC concerning the yield of 3,600 technical and scientific journals cited to date, and it was found that some 80 journals yielded approximately 50 percent of the total articles. Hence, effective January 1965, TPRC subscribed to these journals and began its search directly from these publications. Simultaneously, with the adoption of this procedure, the effort in searching abstracting journals was reduced, by 1967, to seven abstracting journals: Chemical Abstracts, Dissertation Abstracts, International Aerospace Abstracts, Nuclear Science Abstracts, Scientific and Technical Aerospace Reports-NASA, Technical Abstract Bulletin, and U.S. Government Research and Development Reports. As a result of this policy, TPRC is now able to keep abreast of published research results with an average time lag not to exceed six months.

The problem of procuring research documents from the open literature is beginning to assume major proportions especially in the case of foreign literature and special publications of limited distribution. Therefore, TPRC's specialized holdings, which number 35,800 to date, are assuming increasing importance for rapid access to the world literature on thermophysics and thermophysical properties. It is TPRC's experience that literature retrieval programs which yield bibliographies as their end product are becoming increasingly less useful because of the time lapse involved in

procuring the cited documents. In an attempt to remedy this situation, TPRC has supplemented its long-standing practice of submitting bibliographic responses to literature search requests with standard microfiche copies of documents. The conversion of hard copy document holdings into microfiche was completed in 1967.

2.2 The Retrieval Guide

The comprehensive edition of the THERMOPHYSICAL PROPERTIES RESEARCH LITERATURE RETRIEVAL GUIDE was published in October 1967 by the Plenum Publishing Corporation [1].

This three-book volume represents the printout of a special computer program and provides quick access to world literature on thirteen thermophysical properties of all matter. Its substance and property coverage are listed in Table E-2.

This volume completes the coverage of the world literature published from 1920 (in some cases earlier) to June 1964 on thirteen thermophysical properties. It is a merger of the material contained in the earlier Volumes 1 and 2 together with the material of Volume 3 which was not published separately. The contents of the three books of the Retrieval Guide are as follows:

Book 1 - Primarily constitutes TPRC's classified Directory of Substances in which information on the thirteen thermophysical properties are reported. Book 1 also contains three other major chapters which greatly enhance its usefulness. These consist of: (1) Guide to TPRC Substance Classification Procedure and Numerical Codes; (2) Dictionary of Synonyms and Trade Names with a Listing of Cross References; and (3) Index to Mixtures.

Book 2 - Contains the classified code entries and publication year of each reference for each property of each material. The classified code entries cover the following:

<u>Phys. State:</u>	1-Solid; 2-Liquid; 3-Gas; 4-Semi-solid; 5-Powder; 6-Suspensoid; 7-Sintered; 8-Solid-Gas system; 9-Solid-Liquid system.
<u>Subject:</u>	1-Theoretical; 2-Experimental; 3-Theo. and Exp.; 4-Property values; 5-Theo. and Prop. val.; 6-Exp. and Prop. val.; 7-Theo., Exp., and Prop. val.; 8-Survey, Review, Compendia, or Bibliography.
<u>Language:</u>	1-Eng.; 2-Fr.; 3-Ger.; 4-Dutch; 5-It.; 6-Jap.; 7-Rus.; 8-Span.; 9-Other.
<u>Temperature:</u>	1-Low, 0 to 75 K; 2-Normal, 75 to 1275 K; 3-High, 1275 K and up; 4-(Low+Normal); 5-(Normal+High); 6-(Low+Normal+High); 7-Not specific

TABLE E-2. SUBSTANCE AND PROPERTY COVERAGES OF RETRIEVAL GUIDE*

SUBSTANCE COVERAGE - All Matter

Elements and chemical compounds	9,030
Ferrous and nonferrous alloys	9,970
Mixtures	13,396
Systems, composites, etc.	1,643
Polymers, rubbers, etc.	2,600
Refractories	961
Glasses	1,109
Natural products	1,100
Minerals	662
Paints, surface finishes, coatings	2,632
Slags, scales, aggregates, cermets, fuels, lubricants, fibers, fabrics, pharmaceuticals, insulations, building materials, residues, etc.	1,967
General	46
Total number of substances	45,116

PROPERTY COVERAGE - Transport and
Thermodynamic Properties Encountered
in Heat and Mass Transfer Calculations

Thermal conductivity (including commodation coefficient and thermal contact resistance)	31,050
Specific heat	28,020
Viscosity (Newtonian and non- Newtonian; including fluidity)	46,870
Thermal radiative properties (Emittance, reflectance, absorption, transmittance, and optical constants)	9,400
Diffusion coefficient	21,720
Thermal diffusivity	1,705
Prandtl number	504
Total number of reference entries	139,305

* This storehouse of information has come from 33,700 references representing 26,562 authors and 3,600 separate scientific and technical journals and books in addition to sources of governmental and industrial reports (e.g., Defense Documentation Center, Clearinghouse for Federal Scientific and Technical Information, Atomic Energy Commission, National Aeronautics and Space Administration, research centers, and the like).

Book 3 - Part A provides bibliographic citations for the 33,700 references covering scientific and technical journals in addition to university dissertations and technical reports of governmental agencies, industrial organizations, and research centers and laboratories. Part B contains an index to names of contributing authors.

In January 1967, the scope of property coverage was increased to include the coefficients of linear and volumetric thermal expansion and surface tension. Furthermore, each thermophysical property is coded separately instead of by groups. For instance, the thermal conductivity, accommodation coefficient, and thermal contact resistance, formerly all coded under the property "thermal conductivity", are now coded separately. Similarly, the five entries under thermal radiative properties are now listed separately. Thus, since January 1967 TPRC maintains coverage of over sixteen thermophysical properties for all materials. They are:

1. Thermal conductivity
2. Accommodation coefficient
3. Thermal contact resistance
4. Thermal diffusivity
5. Specific heat at constant pressure
6. Viscosity
7. Emittance
8. Reflectance
9. Absorptance
10. Transmittance
11. Absorptance to emittance ratio
12. Prandtl number
13. Diffusion coefficient
14. Thermal linear expansion
15. Thermal volumetric expansion coefficient
16. Surface tension

2.3 Automation and Computerized Information Storage and Retrieval System

With the installation of the CDC 6500 computer at Purdue University in Summer 1967, TPRC's long-standing need for a remote-accessed time-shared computer capability has been fulfilled. As a result, TPRC now has a fully automated bibliographic search capability to respond to specific inquiries or to process standing requests for a continuing bibliographic service tailored to meet demands for specific technical profiles of individual engineers, scientists, corporations, or laboratories.

Since 1968, TPRC has generated, quarterly, a miniature "Retrieval Guide" for each property in order to serve its own inhouse programs. This service, called the UPDATE PLAN, is also available to subscribers at a very nominal cost.

3.0 Data Tables Projects [2]

3.1 Data Book

Synthesis of existing fragments of knowledge is as important as so-called original observation. The availability of adequate standard reference tables of numerical data is essential to national progress, economy, and defense.

The three-volume loose-leaf 11" x 17" size TPRC Data Book is well known nationally and, indeed, internationally as the most comprehensive and authoritative reference data source of its kind. In view of this work's continuing rapid growth since 1960, and the extensive physical proportions it has assumed, early in 1967 it was decided to discontinue the present procedure of publication in loose-leaf format and its semi-annual dissemination by TPRC. Instead, this Data Book was restructured and extensively revised and will soon be available through a commercial publisher in the form of formal hard-bound volumes. The forthcoming publication is entitled TPRC SERIES ON THERMOPHYSICAL PROPERTIES OF MATTER and will be described in more detail later.

The last supplement of the data sheets in the loose-leaf format was that of December 1966. All the available sets of the loose-leaf Data Book were exhausted by mid-year 1967. The data sheets produced during 1967 were held in abeyance and will be released for the first time in the forthcoming Series. It is anticipated that the new volumes will become available according to the tentative schedule presented subsequently.

The changeover from the loose-leaf TPRC publication to formal commercial publication has entailed a number of considerations. Significant among these were the following:

1. The Data Book was to be reduced from its unconventional 11" x 17" dimensions to 9-1/4" x 11-1/4" with printing back to back.
2. The organization of the contents were completely restructured in order to improve user's convenience. The new series is organized into volumes by properties.

3. In order to obviate the cumbersome merging of supplements and the associated high cost of dissemination, it was decided that each edition of a volume will be updated, revised and enlarged approximately every five years.
4. For those who have need for the most up-to-date information, TPRC will provide specific inquiry service or one may subscribe to the automatic UPDATE PLAN tailored to meet a specific technical profile of an engineer, scientist, corporation, or laboratory.

The above-outlined procedure closely parallels the concept which TPRC has followed during the past ten years for the dissemination of bibliographic information. That is, the major accomplishments are published in formal volumes through commercial channels while TPRC disseminates materials to maintain its publications and audience on a current basis.

3.2 TPRC Series on Thermophysical Properties of Matter

Table E-3 gives an indication of the structure, scope, and publication schedule of the forthcoming TPRC SERIES ON THERMOPHYSICAL PROPERTIES OF MATTER. As shown in Table E-3, to be published in 1969 are Volumes 1 through 8, on which a further summary of statistical data is given in Table E-4.

Each volume of the series comprises three sections: the first section is a text on the theory, estimation, and measurement of the property, the second section presents the available numerical data for the property of the materials, and the third section is a comprehensive material index.

The following brief summaries will serve to characterize each of the active data tables projects.

3.3 Projects

- a. Thermal Radiative Properties (Emittance, Reflectance, Absorptance, and Transmittance)

This group of properties constitutes Volumes 7, 8, and 9 of the new TPRC SERIES. Only Volume 7 (Metallic Elements and Alloys) has now been finished and is ready for publication.

The present tables are organized in a way that is much different from that in the original TPRC Data Book. This is due to the establishment of a new scheme for the designation and categorization of the sub-properties. According to the new scheme, by applying the proper geometric and wavelength descriptors to the prime properties, there are altogether thirty-three sub-properties for any material.

TABLE E-3. PUBLICATION SCHEDULE FOR
TPRC SERIES ON THERMOPHYSICAL PROPERTIES OF MATTER*

	1969	1970	1971	1972	1973	1974	1975	1976	1977
Volume 1. Thermal Conductivity of Metallic Elements and Alloys	1565				Second Edition				
Volume 2. Thermal Conductivity of Nonmetallic Solids	1255				Second Edition				
Volume 3. Thermal Conductivity of Nonmetallic Liquids and Gases	640				Second Edition				
Volume 4. Specific Heat of Metallic Metallic Elements and Alloys	825					Second Edition			
Volume 5. Specific Heat of Nonmetallic Solids	1720					Second Edition			
Volume 6. Specific Heat of Nonmetallic Liquids and Gases	390					Second Edition			
Volume 7. Thermal Radiative Properties of Metallic Elements and Alloys	1650						Second Edition		
Volume 8. Thermal Radiative Properties of Nonmetallic Solids	880						Second Edition		
Volume 9. Thermal Radiative Properties of Coatings		1690					Second Edition		
Volume 10. Thermal Diffusivity		500						Second Edition	
Volume 11. Viscosity		400						Second Edition	
Volume 12. Thermal Expansion of Metallic Elements and Alloys			500						Second Edition
Volume 13. Thermal Expansion of Nonmetallic Solids			500						Second Edition

*The publication schedule shown in the table gives the estimated number of pages for the first edition and the years in which the volumes are to be published. After the second edition, subsequent editions of each volume will be released at intervals of five years.

TABLE E-4. SUMMARY OF STATISTICAL DATA ON VOLUMES 1 TO 7 OF
"TPRC SERIES ON THERMOPHYSICAL PROPERTIES OF MATTER"

		No. of Pages*	No. of Curves	No. of References**
Volume 1.	Thermal Conductivity of Metallic Elements and Alloys	1565	5539	1013
Volume 2.	Thermal Conductivity of Nonmetallic Solids	1255	4627	598
Volume 3.	Thermal Conductivity of Nonmetallic Liquids and Gases	640	1505	725
Volume 4.	Specific Heat of Metallic Elements and Alloys	825	1186	428
Volume 5.	Specific Heat of Non- metallic Solids	1720	1009	449
Volume 6.	Specific Heat of Non- metallic Liquids and Gases	390	863	595
Volume 7.	Thermal Radiative Properties of Metallic Elements and Alloys	1650	5309	352

* Estimated

** These are the references to data sources only, not including those references to the text on the theory, estimation, and measurement of the respective thermophysical properties.

Since Volume 7 is now essentially finished, the major efforts are concentrated on the processing of data on coatings and nonmetallic solids so that Volumes 8 and 9 can be published in 1969 and 1970 respectively. The new classification scheme for coatings is now "finalized" after considerable study by the TPRC staff in consultation with several national experts.

Although in Volume 3 of the Retrieval Guide there are only 2,829 references on thermal radiative properties (8.4 percent), in recent years the number of new research documents on the thermal radiative properties has increased steadily and rapidly. Presently, TPRC's acquisition rate is about 6,000 papers per year, 20 percent related to radiative properties. Also, 50 percent of the papers pertaining to radiative properties contain information on coatings. Therefore, it is a very arduous task just to remain current by processing the incoming documents, 1,200 per year, and it has become a standing practice to try to finish the data processing for the most current research documents first, and then to work backwards on the earlier documents. A similar situation is present in the thermal conductivity and specific heat projects.

b. Thermal Conductivity

Thermal conductivity constitutes Volumes 1, 2, and 3 of the new TPRC SERIES. Data compilation for the thermal conductivity of the elements is totally completed and is being maintained on a current basis.

In Volume 3 of the Retrieval Guide, which contains 33,700 references, there are 7,329 references on thermal conductivity, i.e., 21.7 percent (neglecting the relatively small number of references on accommodation coefficient and thermal contact resistance). The present rate of document input into TPRC's Information Storage and Retrieval System is about 6,000 per year. If the past ratio remains the same, there will be 1,300 new documents per year on thermal conductivity entering the System.

In the forthcoming Volumes 1, 2, and 3 of the TPRC SERIES, the thermal conductivity of all metals and alloys, which are organized into seven groups, are included in Volume 1. Volume 2 presents data for thirty groups of nonmetallic solids. Volume 3 contains the critically evaluated and recommended values for 58 fluids which are organized into four groups; for the elements recommended

values are given for solid, saturated liquid, saturated vapor, and gaseous states while for the other three groups of fluids recommended values are given for saturated liquid and gaseous states only.

The three volumes on thermal conductivity contain 2,287 references to data sources. Past experience indicates that only one out of three to four earlier research documents and only one out of two to three more recent documents contains original experimental data. Therefore, to have 2,287 references to data sources, over 7,000 research documents must have been processed.

c. Specific Heat

Specific heat constitutes Volumes 4, 5, and 6 of the new TPRC SERIES. Tables on the specific heat of the elements and of all the important alloys, compounds, and mixtures have been prepared. Data compilation on the specific heat of metallic elements and alloys and nonmetallic solids has been done here at TPRC while the work on nonmetallic liquids and gases has been done at TPRC's Overseas Branch in Japan.

The specific heat of all metals and alloys, which are organized into four groups, are included in Volume 4. Volume 5 presents data for twenty-four groups of nonmetallic solids. Volume 6 contains the critically evaluated and recommended values for 56 fluids organized into four groups.

In Volume 3 of the Retrieval Guide there are 6,978 references on specific heat, i.e., 20.7 percent of the 33,700 references. If the past ratio remains the same, there will be 1,240 new documents per year on specific heat entering the System.

d. Thermal Diffusivity

Thermal diffusivity will constitute Volume 10 of the new TPRC SERIES. The work on this property has been greatly accelerated and all the previous tables have been extensively updated and revised and new research documents are being processed. There are 597 documents on thermal diffusivity in the Retrieval Guide, i.e., 1.7 percent. About 100 new documents enter the System each year.

e. Thermal Expansion

Thermal expansion will constitute Volumes 12 and 13 of the new TPRC SERIES. The work on this property had been active in 1964-66, and suspended

until mid 1968. This work has been reactivated and Volumes 12 and 13 will be published in early 1971.

f. Viscosity

Viscosity will constitute Volume 11 of the new TPRC SERIES. The work on this property had been suspended from 1964 to 1966. Starting early 1967, this work was reactivated in TPRC's European Branch at the Belgian Institute for High Pressure, Brussels, Belgium, with Dr. P. Hestermans as Senior Investigator. This Volume 11 is planned to be published in early 1970.